



Sergey V. Churakov :: Laboratory for Waste Management :: Paul Scherrer Institut

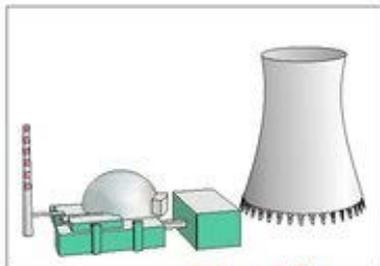
# Labor für Endlagersicherheit

# Outline

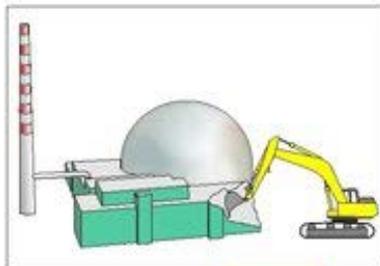
- Status Swiss waste disposal program
- Laboratory for Waste Management (LES)
  - Mission
  - Organizational chart
  - Important infrastructure and capabilities
  - External funding
- Selected research projects
  - Transport and retardation of dose determining nuclides
  - Reactivity of waste and material interfaces
  - Competitive sorption and transport in compacted systems
  - Thermodynamics at elevated temperatures
- Summary

# Origin of Nuclear Waste and Disposal Concepts

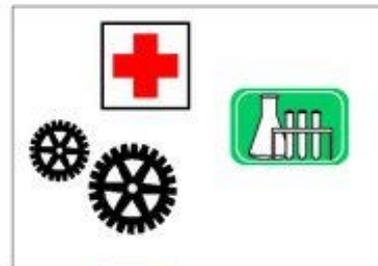
<http://www.ensi.ch>



Betrieb



Abbruch

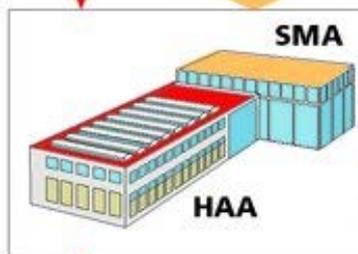


Medizin  
Industrie  
Forschung

HAA

SMA

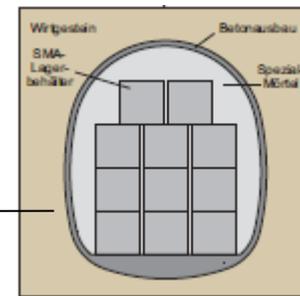
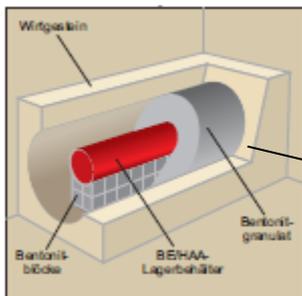
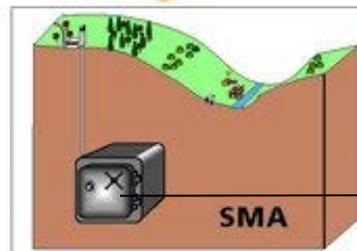
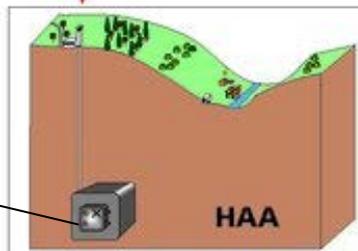
Zwischenlager



SMA

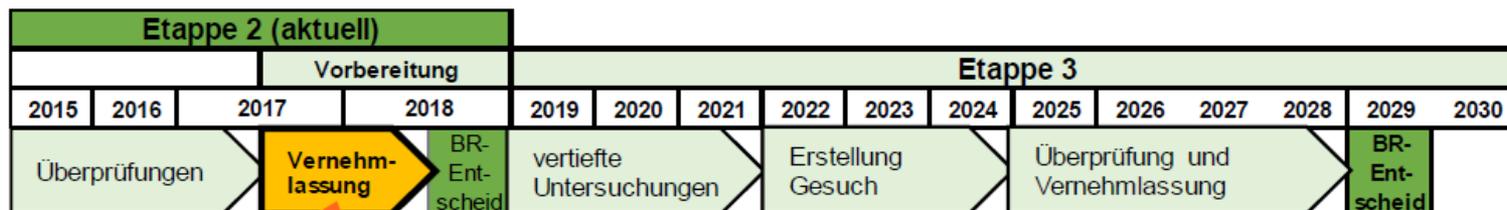
HAA

Geologische Tiefenlager





# ABSCHLUSS ETAPPE 2, AUSBLICK ETAPPE 3



## Vernehmlassung Etappe 2

- ▶ Ab Ende 2017, während 3 Monaten
- ▶ Offen für Kantone, Nachbarstaaten, Gemeinden, Organisationen, Bevölkerung
- ▶ Aufgelegte Dokumente: Ergebnisbericht (inkl. Objektblätter), Erläuterungsbericht, Berichte der Nagra, alle Prüfungsberichte und Stellungnahmen



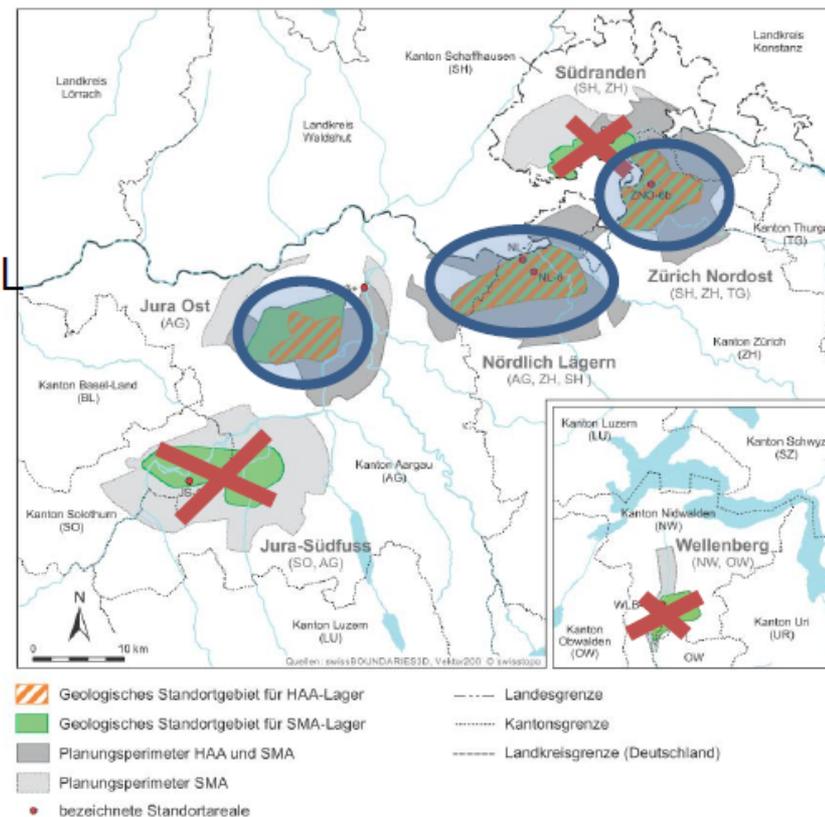
# Gesamtbeurteilung ENSI

## SMA

- NL hat keinen eindeutigen Nachteil
- Zurückstellung der Standortgebiete WLB, JS, SR wegen eindeutiger Nachteile im Vergleich zu ZNO, JO & NL

## HAA

- kein HAA Standortgebiet hat einen eindeutigen Nachteil
- Bestehende Ungewissheiten sind in Etappe 3 SGT reduzierbar
- Kein SG wird zurückgestellt



# LES Mission

LES (Labor für Endlagersicherheit) serves the **national needs, present and future**, in providing important parts of the **scientific basis for the safe disposal of radioactive waste**.

LES supports Nagra by providing **state-of-the-art synthesis reports and data repository safety assessment** in the context of the national waste management programme.

LES carries out a research programme in the areas of:

- **repository in situ conditions**, their evolution and **repository induced effects** including both **modelling and experimental aspects**.
- Interfacial chemistry and transport of radionuclides in repository systems
- fundamental understanding of system behavior for the **long term predictive modelling** and knowhow transfer.

LES maintains:

- **proper balance between applied and basic research**
- **the state-of-the-art expertise and knowledge** in strategic areas
- tight connections to the University of Bern to contribute to the education of young scientists in the field of geochemistry of geological waste disposal.

# LES contribution to the Sectoral Plan stage 3

Maintain **state-of-the-art** functionality of key **models** and **datasets for safety analysis**, including sorption, diffusion and thermodynamics.

Fill **missing gaps in databases**:

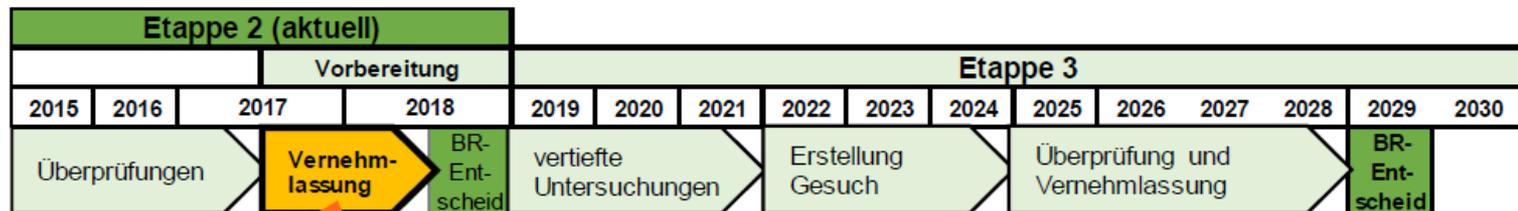
- **Redox sensitive elements** and justification of “chemical analog” arguments
- **Sorption competition / transferability** of data for compacted/disperse systems
- Chemistry of **dose determining radionuclides**

Geochemical evolution of in-situ repository conditions:

- **Production and transport of volatile species**
- **Reactivity of technical barriers and long term evolution of their safety function**

**Sample characterization** from site specific the field explorations

Scientific **documentation for the General License Application**



# LES-organization

## Laboratory for Waste Management/Mineralogy UniBe

Churakov S.V.

Secretariat

Gschwend B.

4402  
Clay Sorption  
Mechanisms

Baeyens B.

4403  
Transport  
Mechanisms

Prasianakis N.

4404  
Diffusion Processes

Van Loon L.

4406  
Cement  
Systems

Wieland E.

Uni Bern  
IfG  
Mineralogy

Churakov S.V.

Dähn R.  
Marques M.  
Thoenen T.

Kéri A.-M.

Wick S.

Geng G.

Marafatto F.

Eltayeb E.

Schaible A.

Lösch H.

Curti E.  
Gimmi T.  
Kosakowski G.

Kulik D.

Miron D.

Nakarai K.

Patel R.

Yang G.

Yang Y.

Hax Damiani L.

Krejci P.

Luraschi P.

N.N.

Gatschet M.

Glaus M.  
Pfungsten W.

Frick S.

Bunic P.

Tits J.  
Hummel W.

Cvetkovic B.

Geng G.

Nedyalkova L.

Mancini A.

Kunz D.

Laube A.

Adams D.  
Armbruster T.  
Cametti G.

Doebelin N.

Eggenberger U.

Fisch M.

Kurganskaya I.

Weibel G.

Schliemann R.

Lemp C.

Guest Scientist, Post Doc

Technician

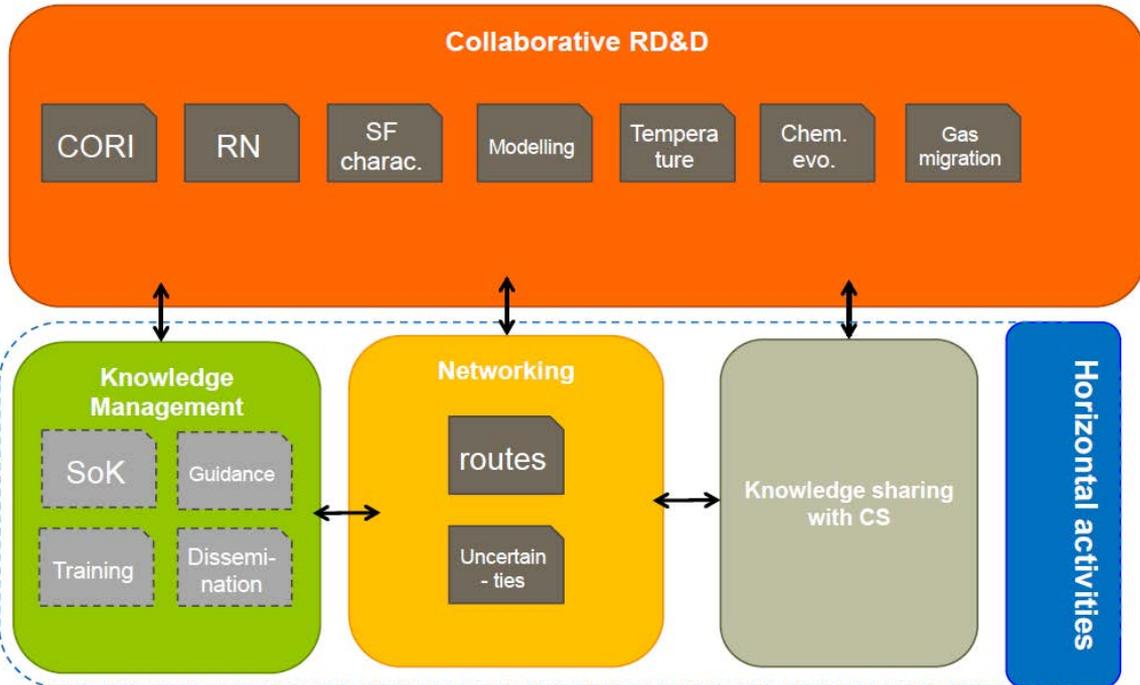
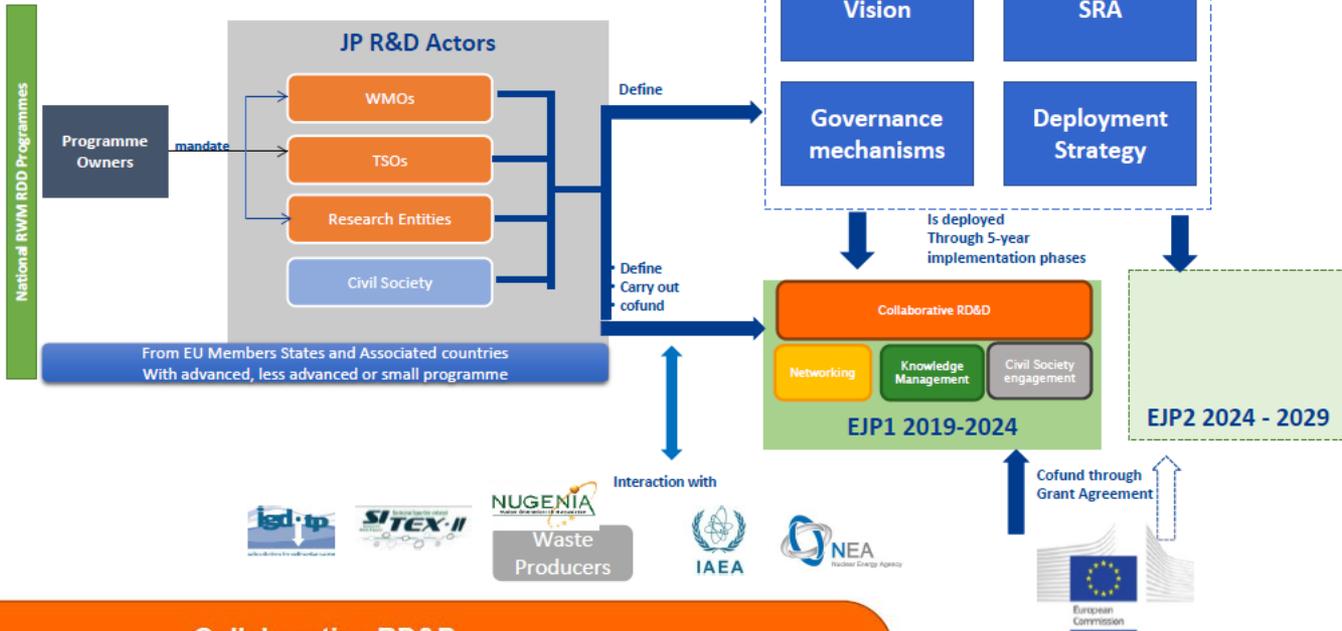
PhD student

MSc student

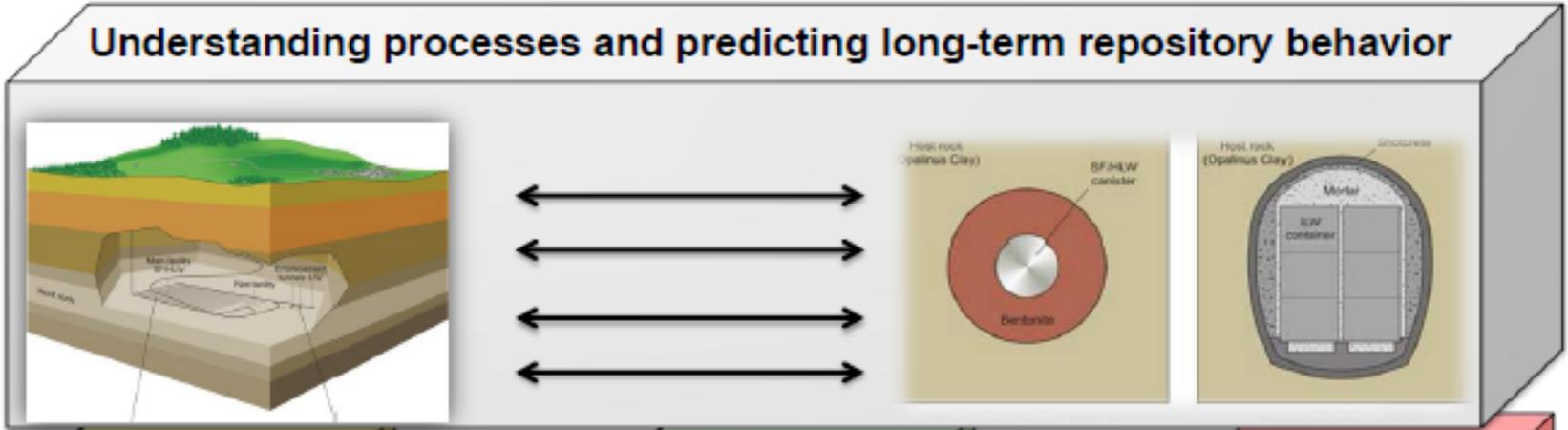
# Recent third party funded projects

- **EU-Horizon 2020:** *“DISCO: Spent fuel dissolution & chemistry in container Thermodynamic equilibria calculations in SF.”* E. Curti
- **EU-COFUND 2Y+1Y Postdoc :** *“Alkali-silica reaction in concrete.”*
- **EU-COFUND 2x2Y Postdoc :** *“Dissolution precipitation in porous media.”*
- **4Y SNF PhD:** *“Resolving dissolution precipitation processes in porous media: Pore scale lattice Boltzmann modelling combined with synchrotron based X - ray characterization. “*
- **3Y SNF PhD project :** *“Transport of sorbed species in clays”*
- **CROSS-PSI postdoc** *“Cryo-microspectroscopy at the microXAS beamline for the investigation of redox- and radiation-sensitive samples*
- **HPC projects at CSCS:** 400`000 Node/Hours (equiv. ~300KCHF )

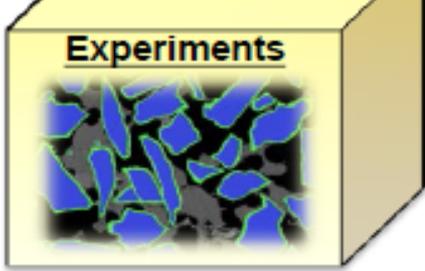
# EURATOM call: EJP1



# Consolidation of modelling and experimental activities



- Diffusion Experiments
- XRF/XRD
- Scanning electron microscopy
- Computer-tomography
- Optical - Microscopy
- Mont Terri



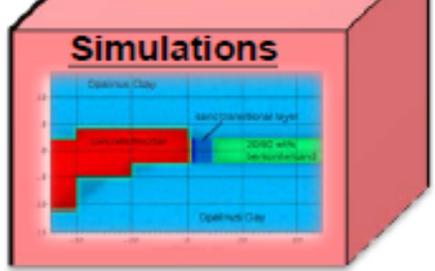
- Diffusion/advection equation
- Sorption models
- Boltzmann equation
- Thermodynamics
- Dissolution/precipitation
- Mineral growth
- Gibbs energy

**Theory / Concepts**

$$\frac{\partial C_i}{\partial t} = \nabla \cdot (SvC_i) + \nabla \cdot (SD_i \nabla C_i) + Q_i$$

$$\frac{\partial S_i}{\partial t} = F_i(C_1, \dots, C_n)$$

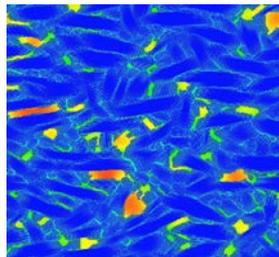
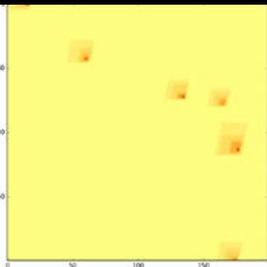
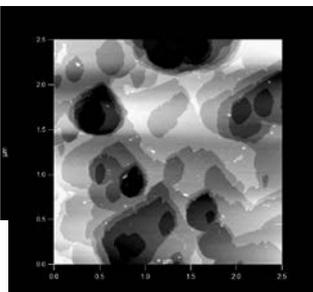
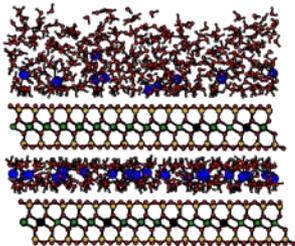
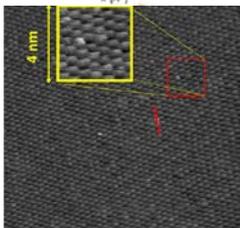
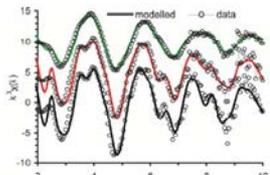
- Reactive transport simulation
- OpenGeoSys-GEMS
- MCOTAC
- Lattice Boltzmann
- Random walk
- Gibbs energy minimization



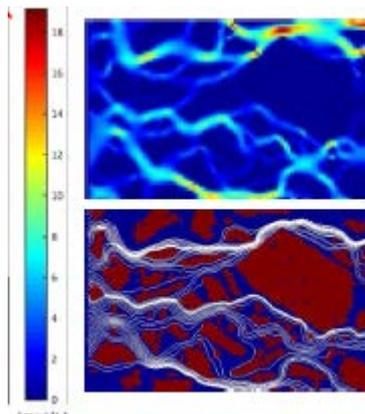
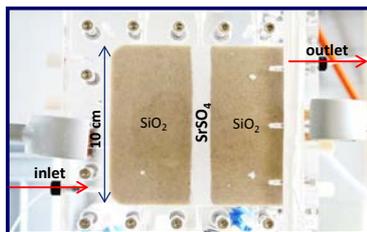
# Mass transport in heterogeneous geochemical systems

Down scaling

Atomic scale

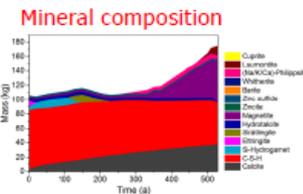
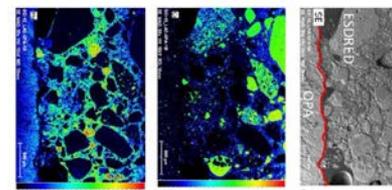
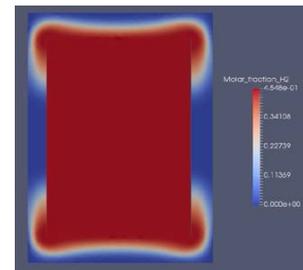
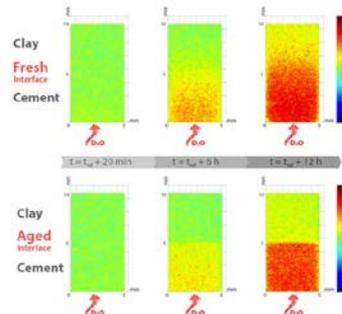


- Ions sorption and transport mechanisms at atomic scale



- [www.nature.com/articles/s41598-017-14142-0](http://www.nature.com/articles/s41598-017-14142-0)
- Pore scale mass reactive mass transport
- Mechanistic description of mineral fluid interaction

- Characterization of material interfaces



- In situ characterization of materials interfaces
- reactive transport simulations and experiments

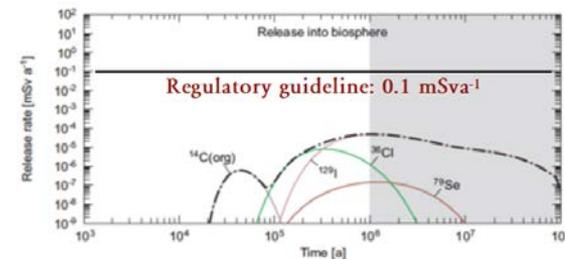
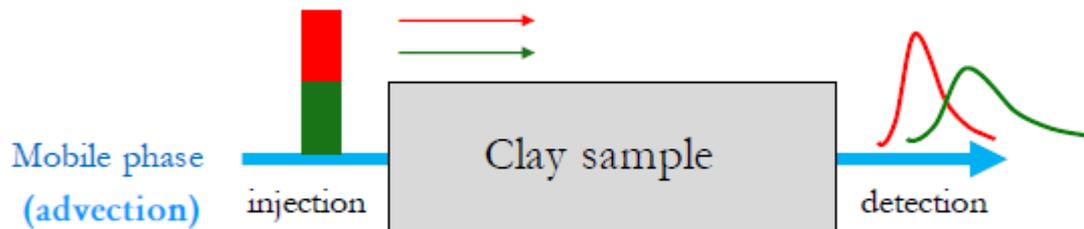


Field Scale

Up scaling



Setup: reference tracer+LMWOC



Can we determine the weak retardation?

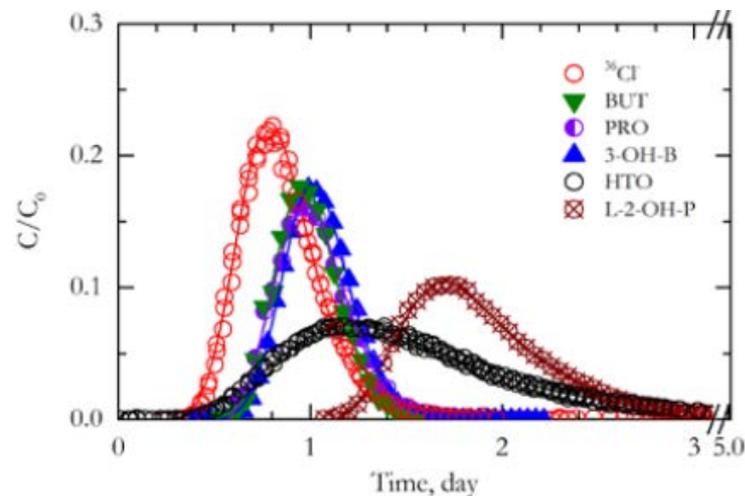
- Yes! (reproducibility, mass balance, sensitivity)

Do we understand the retardation?

- To some extent! (anion exclusion, anion sorption)

Molecular level of structures?

- structure specific (functional group, aliphatic/hydroxylated, etc)
- clay specific (surface properties, etc)
- possible interactions



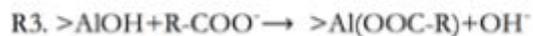
Reaction



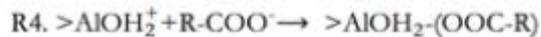
Ion pair



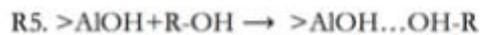
H-bond



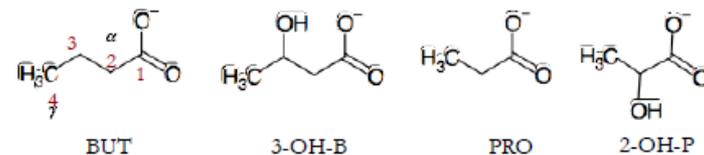
Anion exchange



Ion pair



H-bond

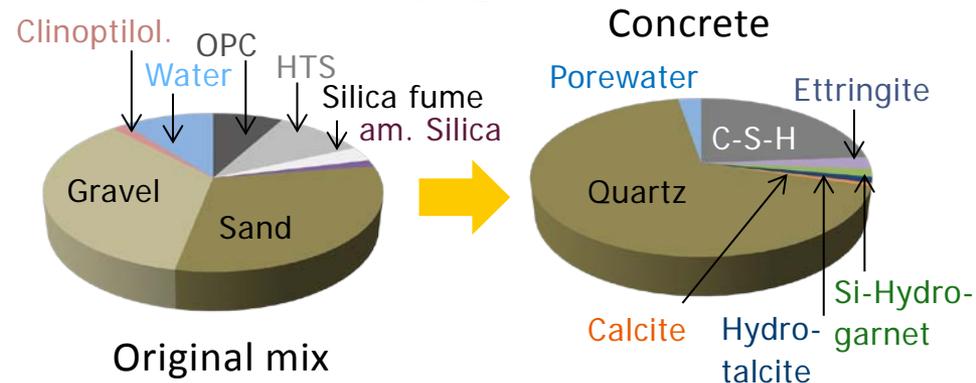
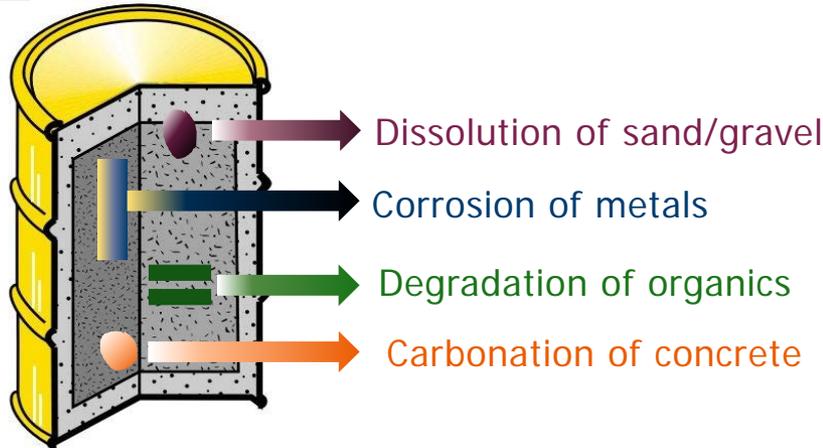


# Interaction of waste with engineered barrier

**Aim:** Geochemical modelling of the temporal evolution of cement-stabilized waste sorts:  
Changes in the mineralogy of the conditioned waste and volume of waste package

Modelled waste sorts: three operational and two decommissioning waste sorts

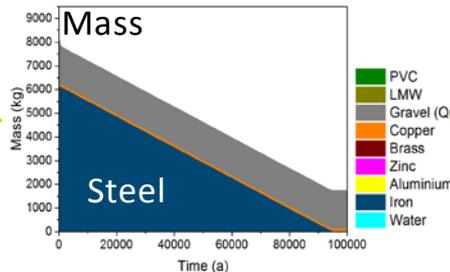
Modelling of the initial composition of the solidifying concrete



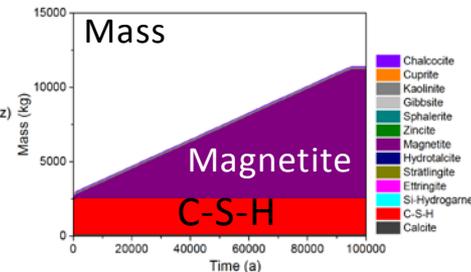
Modelling of temporal evolution (waste materials, mineral composition, volume)



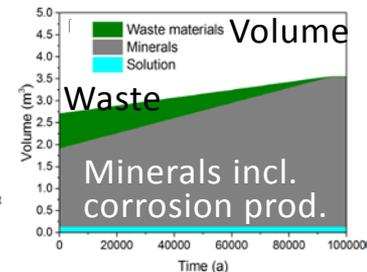
Decommissioning waste



Waste materials



Mineral composition

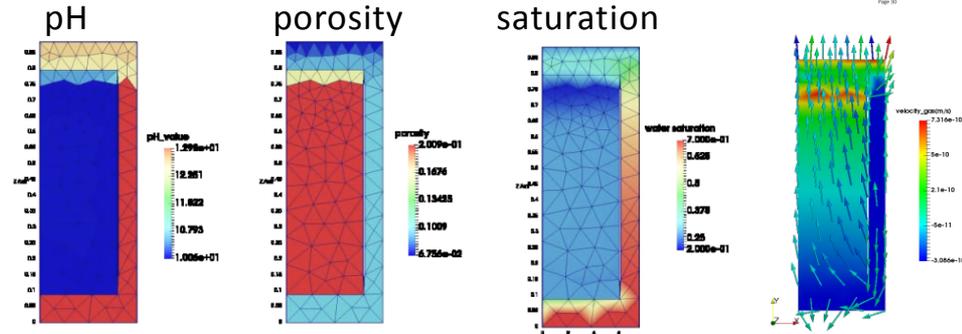
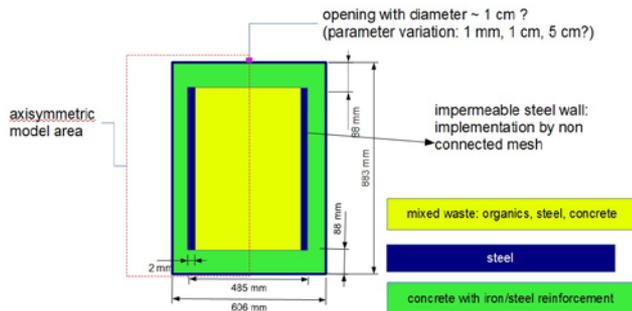
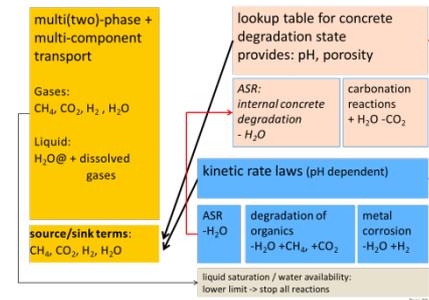


Volume

# Influence of concrete degradation on gas- and water fluxes in a waste package

**Aim:** Modeling of interactions between chemical processes and multi-phase (gas/water) multi-component (several gases) in repository near-field

- Represent concrete degradation (carbonation and alkali-silica-reactions) by a look-up table that provides pH and source-sink terms for coupling with multi-phase transport codes
- pH dependent kinetic laws for metal corrosion



- **Application example:** Gas generation and humidity evolution in a waste package during 40 years of intermediate storage. (in cooperation with Helmholtz Institut für Umweltforschung UFZ)

# Competitive metal sorption on clay minerals

**Motivaton:** In a deep geological repository stable aqueous metals are present from different sources: backfill materials, the corrosion of steel canister, dissolution of the waste forms.

These metals can potentially compete with the released radionuclides for the available sorption sites and reduce their uptake on them.

## Experimental:

### Na-montmorillonite

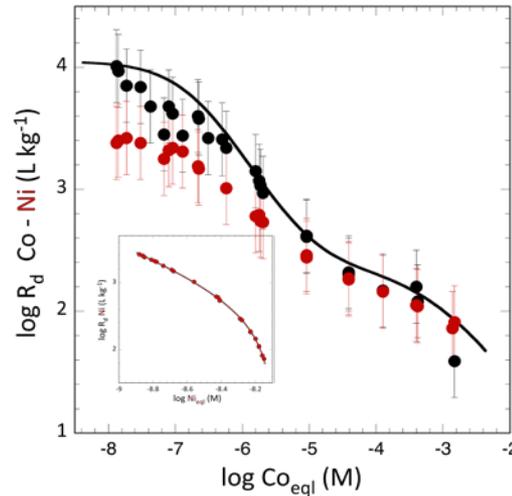
Trace metal: Ni, Eu  $< 10^{-8}$  M

Blocking metal: Co, Ni:  $10^{-8} - 10^{-3}$  M

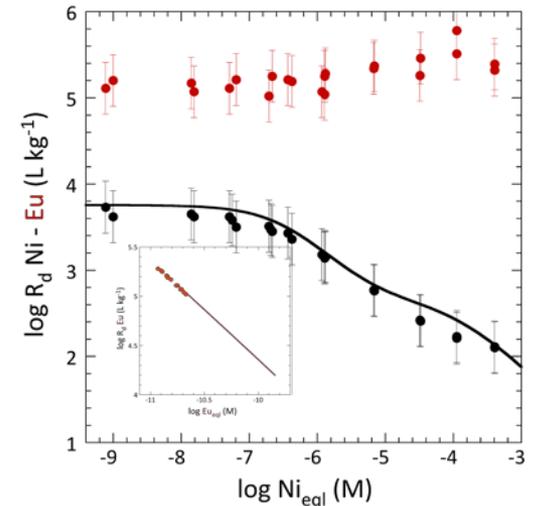
pH = 7

Reaction time: 7 days

## Results and Modelling



Ni<sup>II</sup>-Co<sup>II</sup> are competitive



Eu<sup>III</sup>-Ni<sup>II</sup> are non competitive

## Conclusions:

- Metals with the same valence and hydrolysis behaviour compete: e.g. Ni<sup>II</sup>-Co<sup>II</sup> and Eu<sup>III</sup>-Am<sup>III</sup>
- Metals with different valence are non competitive: e.g. Eu<sup>III</sup>-Ni<sup>II</sup>; Th<sup>IV</sup>-Ni<sup>II</sup>; U<sup>VI</sup>-Ni<sup>II</sup>
- Competitive sorption can be quantified by sorption models (2SPNE SC/CE) and can be taken into consideration in the safety analysis of radioactive waste repositories

# NTB's Sorption Data Base

**NAGRA NTB**

**October 2017**

**The Development of a  
Thermodynamic Sorption Data Base  
(TSDB) for **Illite** and the Application  
to **Argillaceous Rocks**  
(in press)**

**M.H. Bradbury and B. Baeyens**

**NAGRA NTB**

**October 2017**

**The Development of a  
Thermodynamic Sorption Data Base  
(TSDB) for **Montmorillonite** and the  
Application to **Bentonite**  
(in press)**

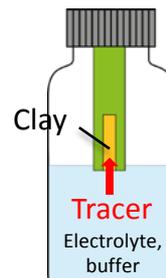
**B. Baeyens and M.H. Bradbury**

# Effect of clay compaction on the transport properties of Eu(III) in illite

## Aim

- To compare sorption data for Eu(III) in illite derived from diffusion experiments with those obtained in dispersed clay suspensions.
- To derive a speciation-based transport model for Eu(III) in compacted illite.

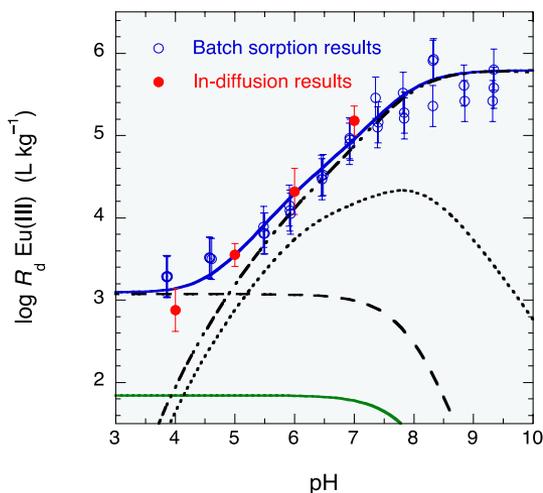
## The in-diffusion technique



1. Equilibrate clay with electrolyte solution.
2. Add tracer (e.g. Eu(III)) to solution phase.
3. Monitor solution concentration of tracer as a function of time.
4. Measure tracer profile in clay at the end of the experiment.

=> Unambiguous information on sorption and diffusion properties

## Results and simulations



- The results of sorption measurements of Eu(III) in compacted and dispersed illite are fully consistent.
- Sorption modelling (blue) of diffusion-derived data can be done using existing sorption models<sup>a</sup> (black lines, representing contributions from individual surface species).
- A minor mobile surface species (green) is additionally introduced to properly reflect the observed diffusion lengths in the in-diffusion measurements (not shown).

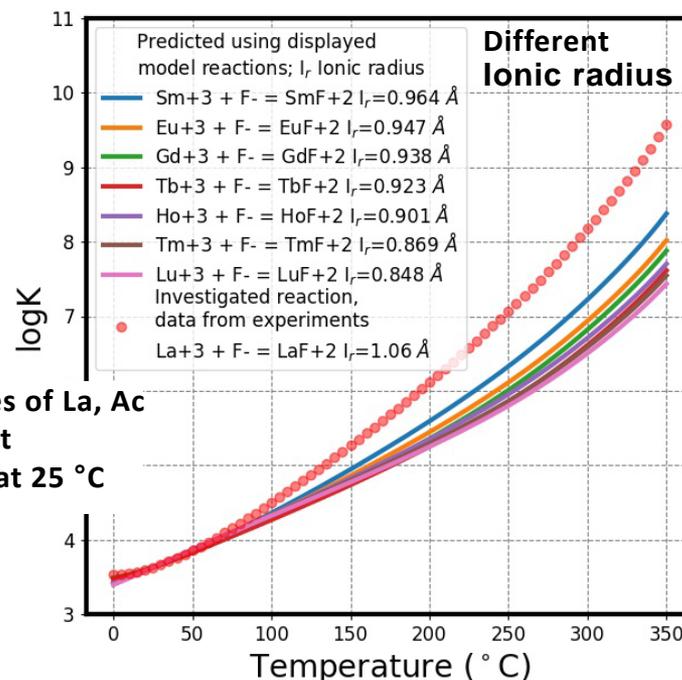
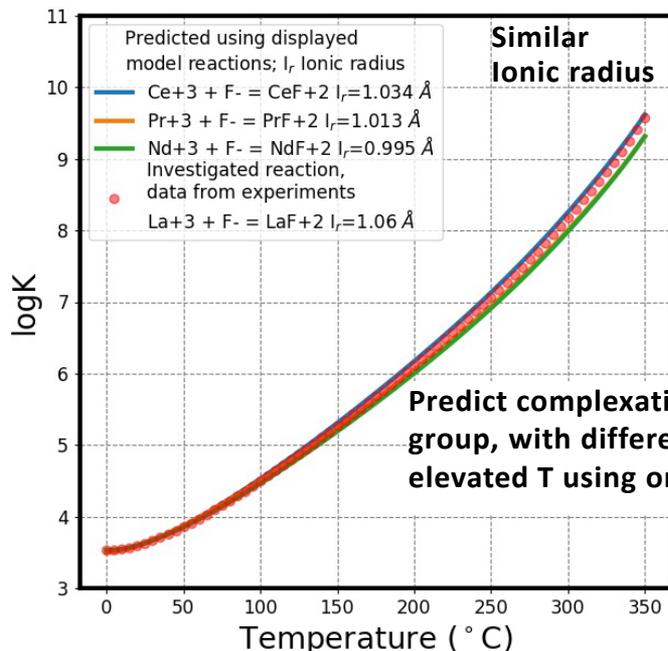
## Conclusion

A speciation-based transport model for Eu(III) has been successfully tested for its ability to robustly predict the diffusion length as a function of broad parameter variations, such as pH, ionic strength, Eu concentration and competing species.

# Predict thermodynamic properties at elevated temperature using *isocoulombic* reactions



- (1) **Generate** chemical reactions from a given list of substances
- (2) **Combine** investigated (unknown temperature effect) and model reactions (known temperature effect) into *isocoulombic* reactions (the same charge types on both sides of the reaction)
- (3) **Investigate** which reactions better predict the properties of investigated reactions, knowing only the logK at 25 °C



## Laboratory for Waste Management (LES)

About LES

Team

Groups

Research Projects

Research Partners and  
Cooperations

Teaching and Education

LES Events

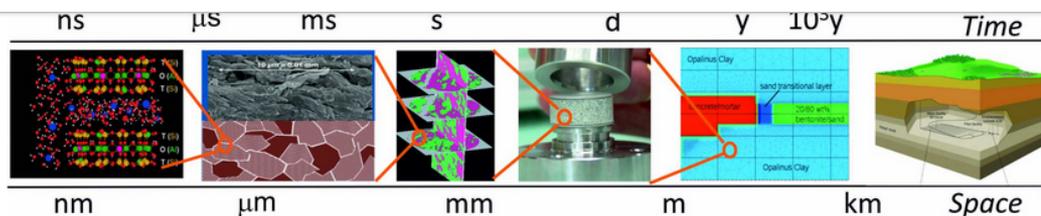
Software and Database

Scientific Highlights

Hot New Papers

Publications

Annual Reports



## Laboratory for Waste Management (LES)

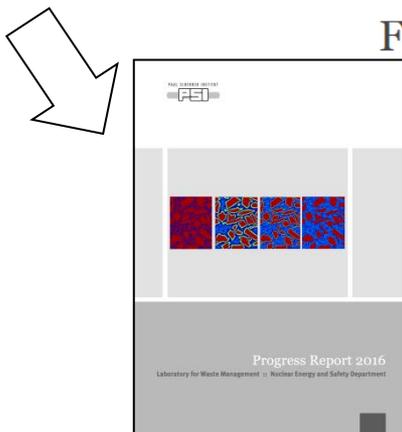
LES is the Swiss competence center for geochemistry and multiscale radionuclide and mass transport in argillaceous rocks and cement and their application to deep geological systems and Swiss radioactive waste repositories. LES offers attractive research projects at the bachelor, master, PhD, and postdoc levels in environmental sciences and nuclear engineering

### Core Competences

- Geochemistry of repository systems
- Retention and migration of radionuclides
- Multiscale reactive transport in natural and engineered barrier systems
- Thermodynamic databases
- Geochemical education

### Facilities, Tools and Infrastructure

- Synchrotron-based material characterization and neutron imaging
- Numerical simulations at atomistic, pore Lattice-Boltzmann, and continuum scales
- Geochemical and thermodynamic modeling tools
- State-of-the-art radiochemical laboratories


**UNIVERSITÄT  
BERN**

#### Contact

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Laboratory for Waste Management  
Prof. Dr. Sergey Churakov  
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3012 Bern

#### Homepage NES

Nuclear Energy and Safety Research  
Division at PSI

Thank you for your attention

