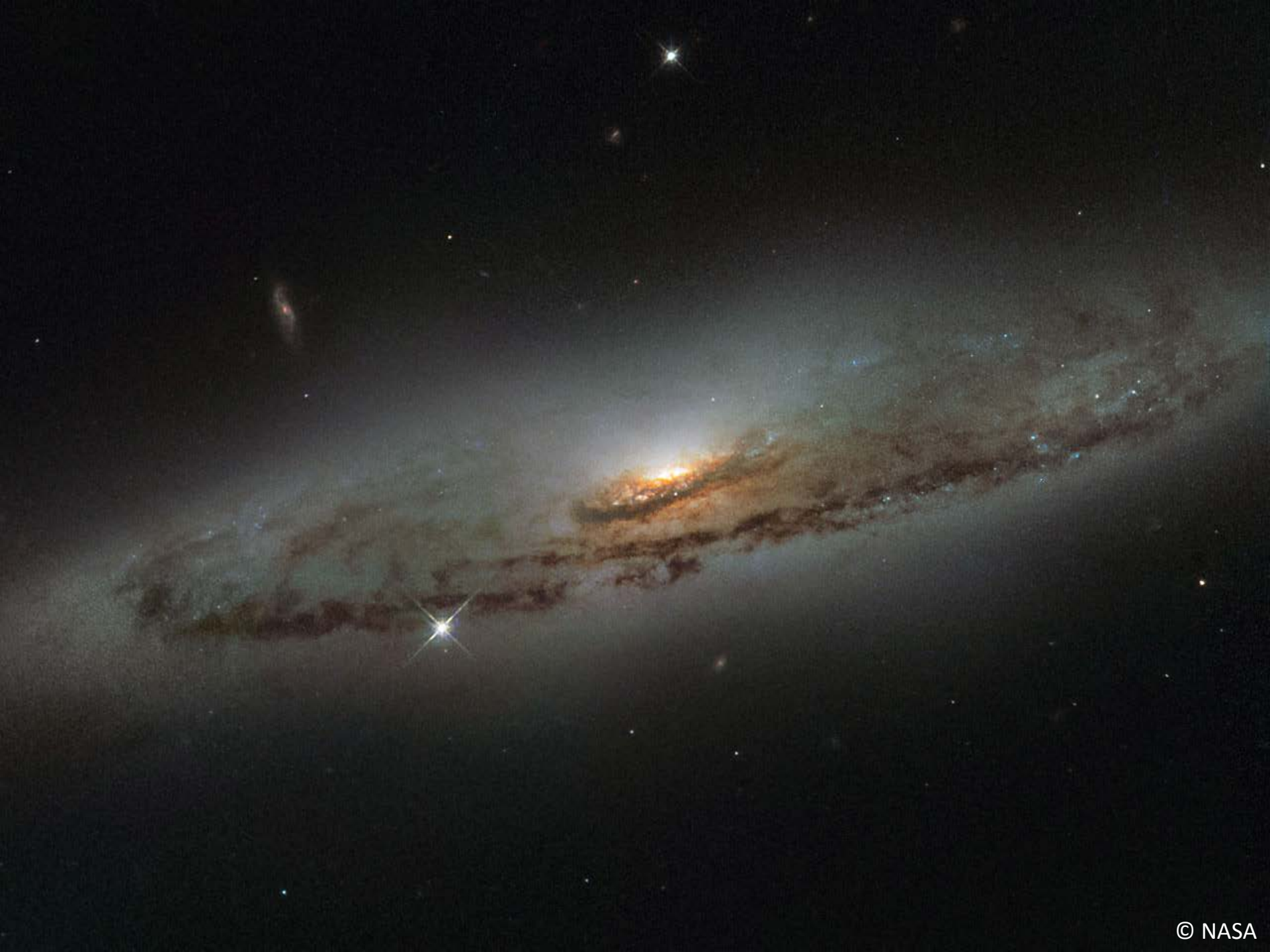


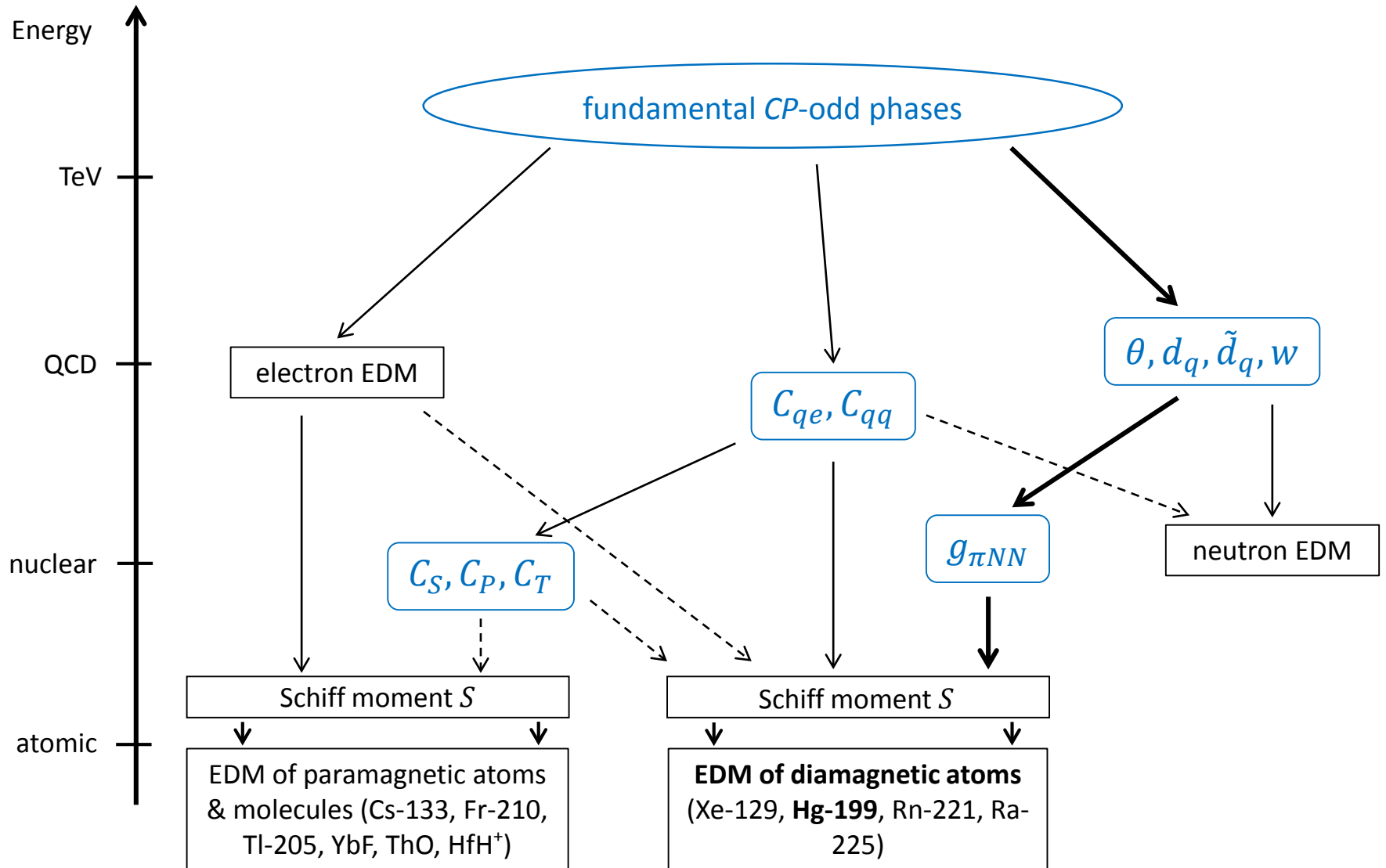
Towards a measurement of the mercury EDM using ultracold atoms

Simon Stellmer
University of Bonn





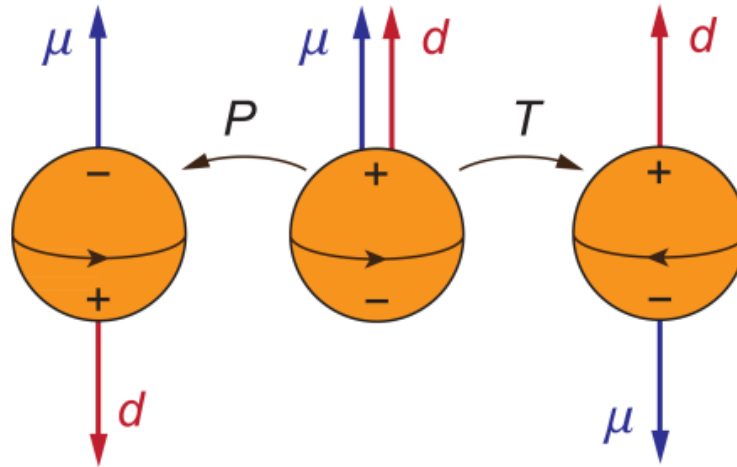
Mechanisms to generate EDMs



The universal approach to measure EDMs

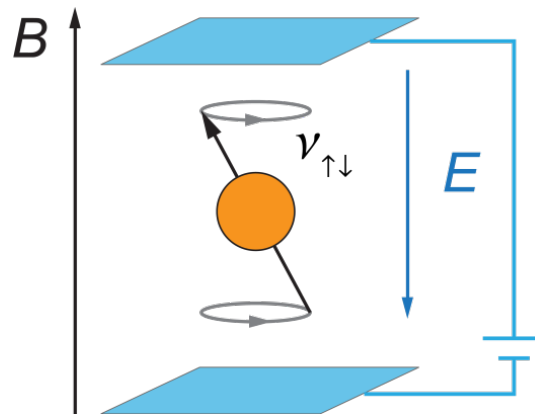


General idea:



T-violation is equivalent to CP-violation (because of CPT!)

General concept:



Larmor frequency:

$$h\nu = |2\mu B \pm 2dE|$$

Sensitivity:

$$\delta d = \frac{\hbar}{2E \sqrt{2N\tau T}}$$

The universal approach to measure EDMs: *colder is better*



Sensitivity:

$$\delta d = \frac{\hbar}{2E \sqrt{2N\tau T}}$$

next-generation experiments:

neutron EDM	spallation sources:	PSI, ESS Lund, ASS, various other institutes	
	reactor UCNs:	ILL, FRM II, various other institutes	
		Sussex/ILL, 2006	2.9 x 10⁻²⁶ e cm

higher neutron flux,
improved cooling in He
or D₂, ...

(n2EDM @ PSI, CryoEDM
@ ILL, FRM II, ...)

electron EDM	atoms:	Cs-133 TI-205	Penn State, Austin Texas, LBNL Berkeley, 2002	1.6 x 10 ⁻²⁷ e cm
		Fr-210	CYRIC, Japan	
molecules:	YbF	Ed Hinds, Imperial College, 2011	1.0 x 10⁻²⁷ e cm	
	ThO	ACME collaboration, Harvard/Yale, 2018	1.1 x 10⁻²⁹ e cm	
	HfF ⁺	Cornell/Ye, JILA, 2017	1.3 x 10⁻²⁸ e cm	

laser cooling &
magneto-optical trapping
of molecules

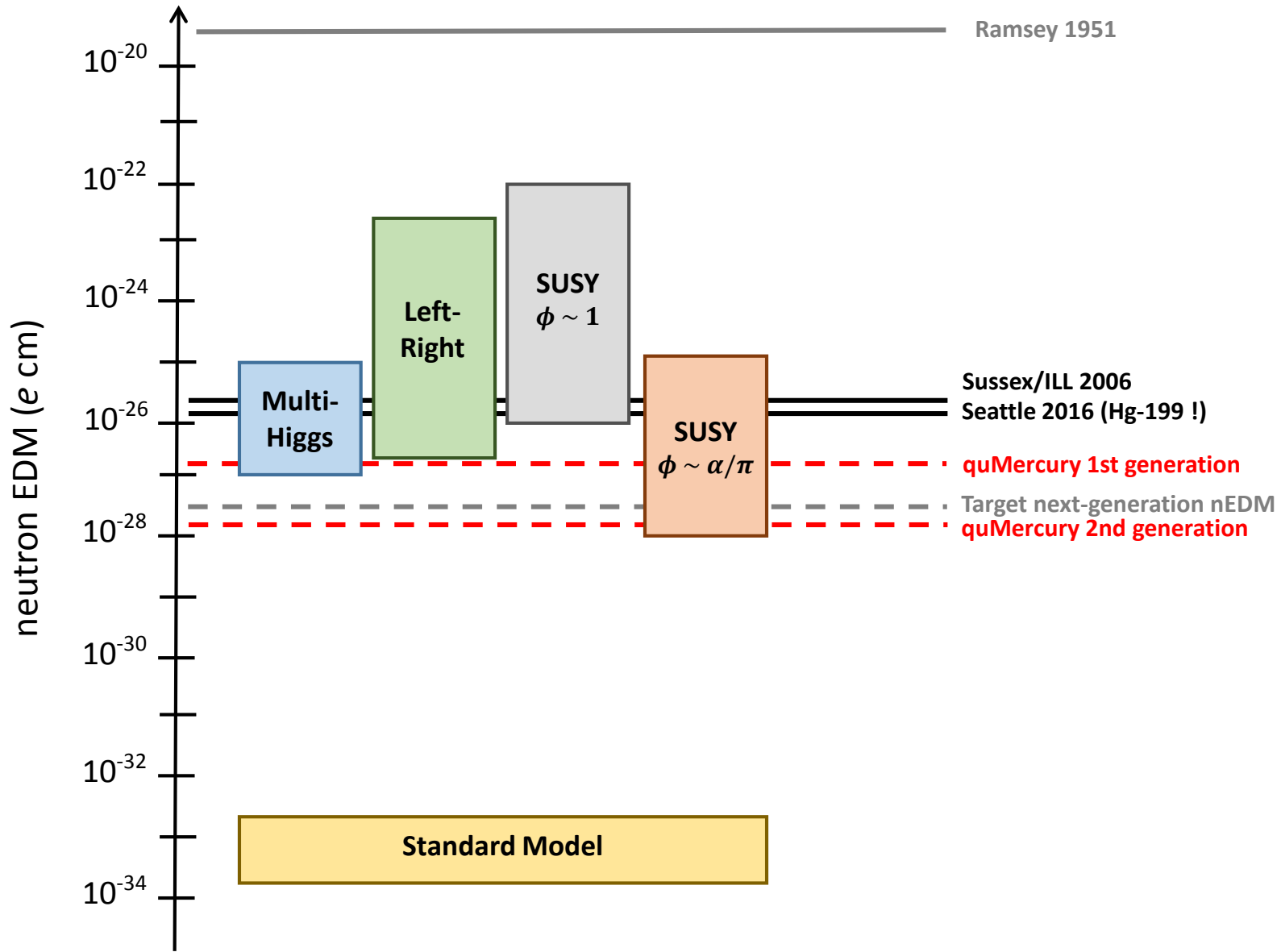
(Imperial, Yale, ACME,
JILA, ...)

atomic EDM	thermal:	Hg-199 Rn-221	Fortson/Heckel, Seattle, 2016 Chupp, TRIUMF	7.4 x 10⁻³⁰ e cm
	MOT:	Ra-225	Argonne, 2016	1.4 x 10 ⁻²³ e cm
	liquid:	Xe-129	Chupp, U of Michigan, 2001	7.0 x 10 ⁻²⁸ e cm

ultracold atoms:

Radium-225 (Argonne)

Excluding beyond-SM theories (example: neutron EDM)



The Washington Hg-199 experiment



VOLUME 59, NUMBER 20

PHYSICAL REVIEW LETTERS

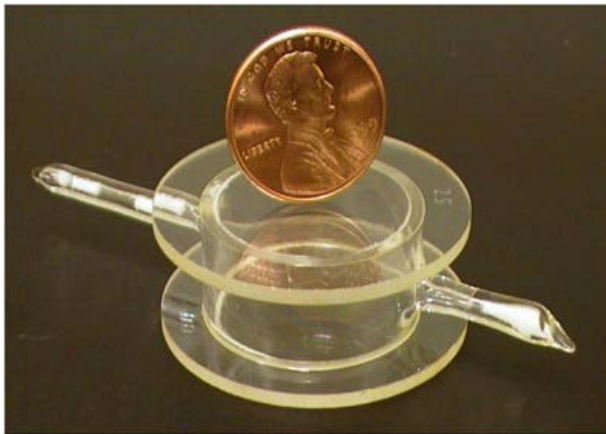
16 NOVEMBER 1987

New Constraints on Time-Reversal Asymmetry from a Search for a Permanent Electric Dipole Moment of ^{199}Hg

S. K. Lamoreaux, J. P. Jacobs, B. R. Heckel, F. J. Raab, and N. Fortson
Physics Department, University of Washington, Seattle, Washington 98195
(Received 17 August 1987)

A search for a permanent electric dipole moment of ^{199}Hg atoms has yielded the null result $d(^{199}\text{Hg}) = (0.7 \pm 1.5) \times 10^{-26} e \cdot \text{cm}$, which improves by an order of magnitude the limits on several possible interactions that violate time-reversal symmetry. The experiment was performed with the use of optically pumped atomic oscillators to measure any shift in the NMR frequency of ^{199}Hg ($I = \frac{1}{2}$) produced by an external electric field.

5000-fold improvement
over 30 years !



PHYSICAL REVIEW LETTERS

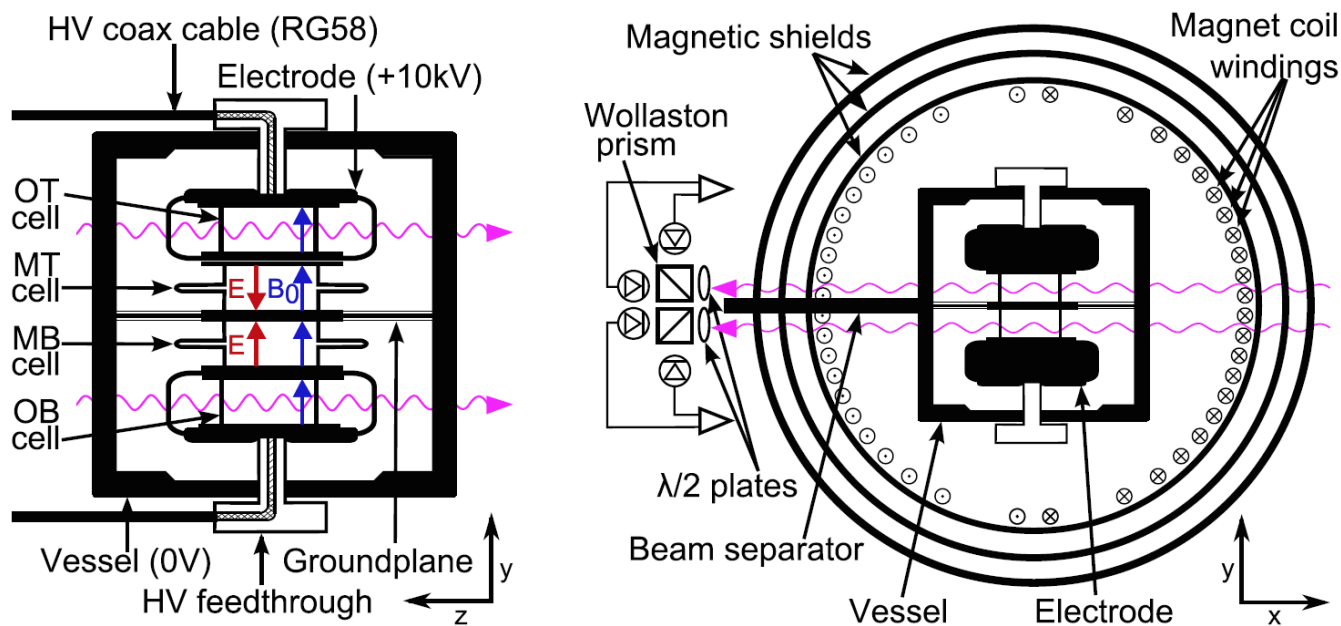
week ending
22 APRIL 2016

Reduced Limit on the Permanent Electric Dipole Moment of ^{199}Hg

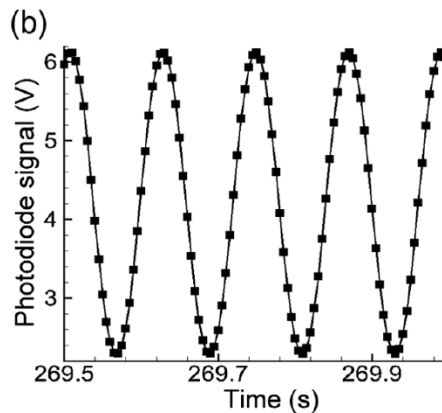
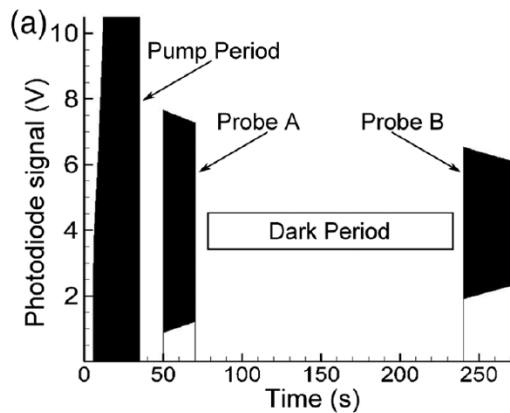
B. Graner,* Y. Chen (陳宜), E. G. Lindahl, and B. R. Heckel
Department of Physics, University of Washington, Seattle, Washington 98195, USA
(Received 19 January 2016; revised manuscript received 8 March 2016; published 18 April 2016)

This Letter describes the results of the most recent measurement of the permanent electric dipole moment (EDM) of neutral ^{199}Hg atoms. Fused silica vapor cells containing enriched ^{199}Hg are arranged in a stack in a common magnetic field. Optical pumping is used to spin polarize the atoms orthogonal to the applied magnetic field, and the Faraday rotation of near-resonant light is observed to determine an electric-field-induced perturbation to the Larmor precession frequency. Our results for this frequency shift are consistent with zero; we find the corresponding ^{199}Hg EDM $d_{\text{Hg}} = (-2.20 \pm 2.75_{\text{stat}} \pm 1.48_{\text{syst}}) \times 10^{-30} e \cdot \text{cm}$. We use this result to place a new upper limit on the ^{199}Hg EDM $|d_{\text{Hg}}| < 7.4 \times 10^{-30} e \cdot \text{cm}$ (95% C.L.), improving our previous limit by a factor of 4. We also discuss the implications of this result for various CP -violating observables as they relate to theories of physics beyond the standard model.

The Washington Hg-199 experiment



Phys. Rev. Lett. 116, 161601 (2016)



Quantity	Expression	Limit	Ref.
d_n	$S_{\text{Hg}}/(1.9 \text{ fm}^2)$	$1.6 \times 10^{-26} e \text{ cm}$	[21]
d_p	$1.3 \times S_{\text{Hg}}/(0.2 \text{ fm}^2)$	$2.0 \times 10^{-23} e \text{ cm}$	[21]
\tilde{g}_0	$S_{\text{Hg}}/(0.135 e \text{ fm}^3)$	2.3×10^{-12}	[5]
\tilde{g}_1	$S_{\text{Hg}}/(0.27 e \text{ fm}^3)$	1.1×10^{-12}	[5]
\tilde{g}_2	$S_{\text{Hg}}/(0.27 e \text{ fm}^3)$	1.1×10^{-12}	[5]
θ_{QCD}	$\tilde{g}_0/0.0155$	1.5×10^{-10}	[22,23]
$(\tilde{d}_u - \tilde{d}_d)$	$\tilde{g}_1/(2 \times 10^{14} \text{ cm}^{-1})$	$5.7 \times 10^{-27} \text{ cm}$	[25]
C_S	$d_{\text{Hg}}/(5.9 \times 10^{-22} e \text{ cm})$	1.3×10^{-8}	[15]
C_P	$d_{\text{Hg}}/(6.0 \times 10^{-23} e \text{ cm})$	1.2×10^{-7}	[15]
C_T	$d_{\text{Hg}}/(4.89 \times 10^{-20} e \text{ cm})$	1.5×10^{-10}	see text



PRL 114, 233002 (2015)

PHYSICAL REVIEW LETTERS

week ending
12 JUNE 2015

First Measurement of the Atomic Electric Dipole Moment of ^{225}Ra

R. H. Parker,^{1,2} M. R. Dietrich,^{1,3} M. R. Kalita,^{1,4} N. D. Lemke,^{1,*} K. G. Bailey,¹ M. Bishof,¹ J. P. Greene,¹
R. J. Holt,¹ W. Korsch,⁴ Z.-T. Lu,^{1,2,†} P. Mueller,¹ T. P. O'Connor,¹ and J. T. Singh^{1,5}

¹Physics Division, Argonne National Laboratory, Argonne, Illinois 60439, USA

²Department of Physics and Enrico Fermi Institute, University of Chicago, Chicago, Illinois 60637, USA

³Department of Physics and Astronomy, Northwestern University, Evanston, Illinois 60208, USA

⁴Department of Physics and Astronomy, University of Kentucky, Lexington, Kentucky 40506, USA

⁵National Superconducting Cyclotron Laboratory and Department of Physics and Astronomy,
Michigan State University, East Lansing, Michigan 48824, USA

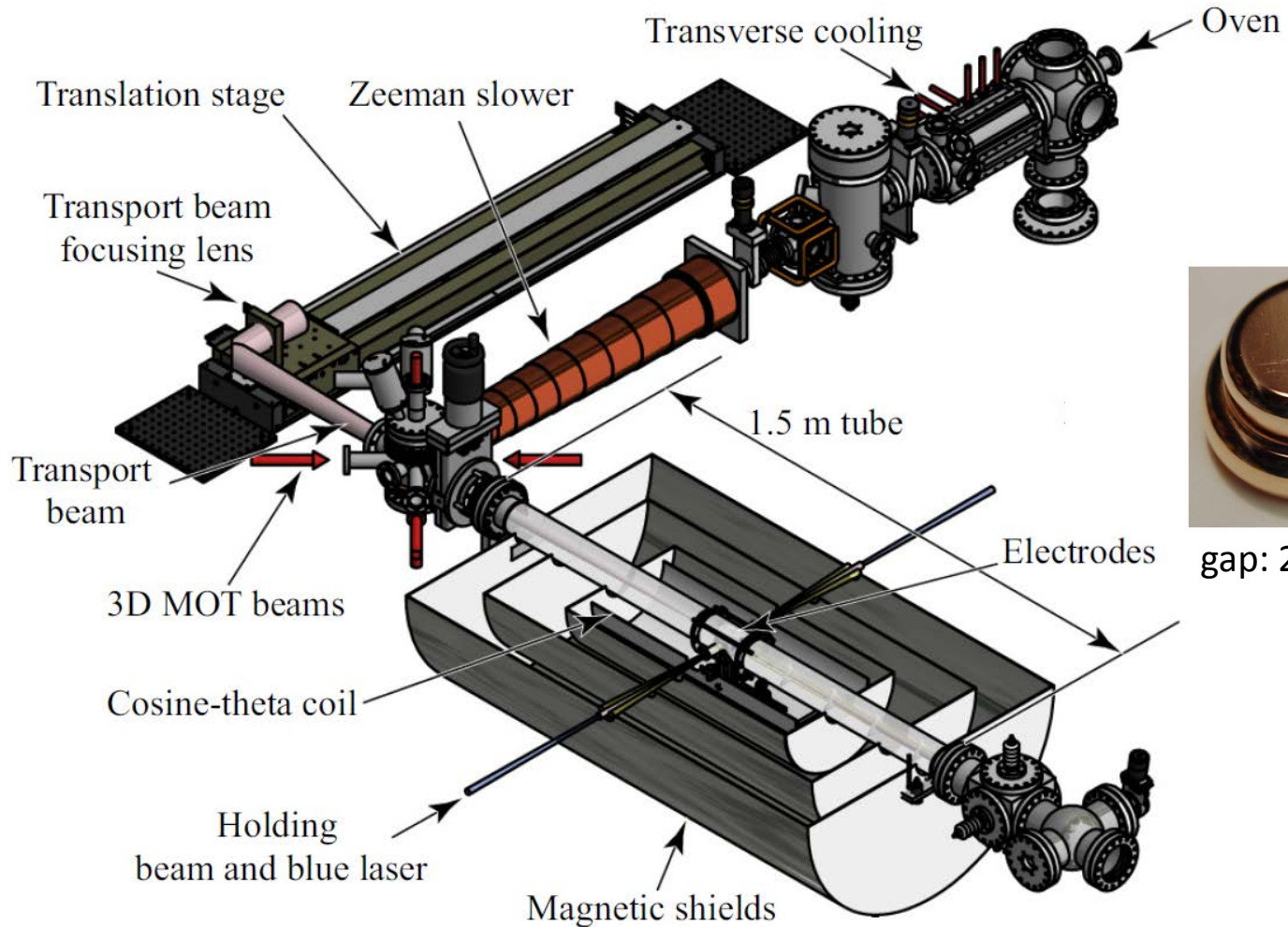
(Received 3 March 2015; published 9 June 2015)

The radioactive radium-225 (^{225}Ra) atom is a favorable case to search for a permanent electric dipole moment. Because of its strong nuclear octupole deformation and large atomic mass, ^{225}Ra is particularly sensitive to interactions in the nuclear medium that violate both time-reversal symmetry and parity. We have developed a cold-atom technique to study the spin precession of ^{225}Ra atoms held in an optical dipole trap, and demonstrated the principle of this method by completing the first measurement of its atomic electric dipole moment, reaching an upper limit of $|d(^{225}\text{Ra})| < 5.0 \times 10^{-22} e \text{ cm}$ (95% confidence).

DOI: 10.1103/PhysRevLett.114.233002

PACS numbers: 32.10.Dk, 11.30.Er, 24.80.+y, 37.10.Gh

The Argonne Ra-225 experiment: laser-cooled atoms!



MOT: 1000 atoms @ $40 \mu\text{K}$
Dipole trap: 50 atoms at $50 \mu\text{K}$



Our approach



QUANTUM METROLOGY

quMercury

- atomic EDM measurements with (non-radioactive) laser-cooled atoms
- find ways to improve sensitivity by using quantum phenomena

Choice of the atomic species

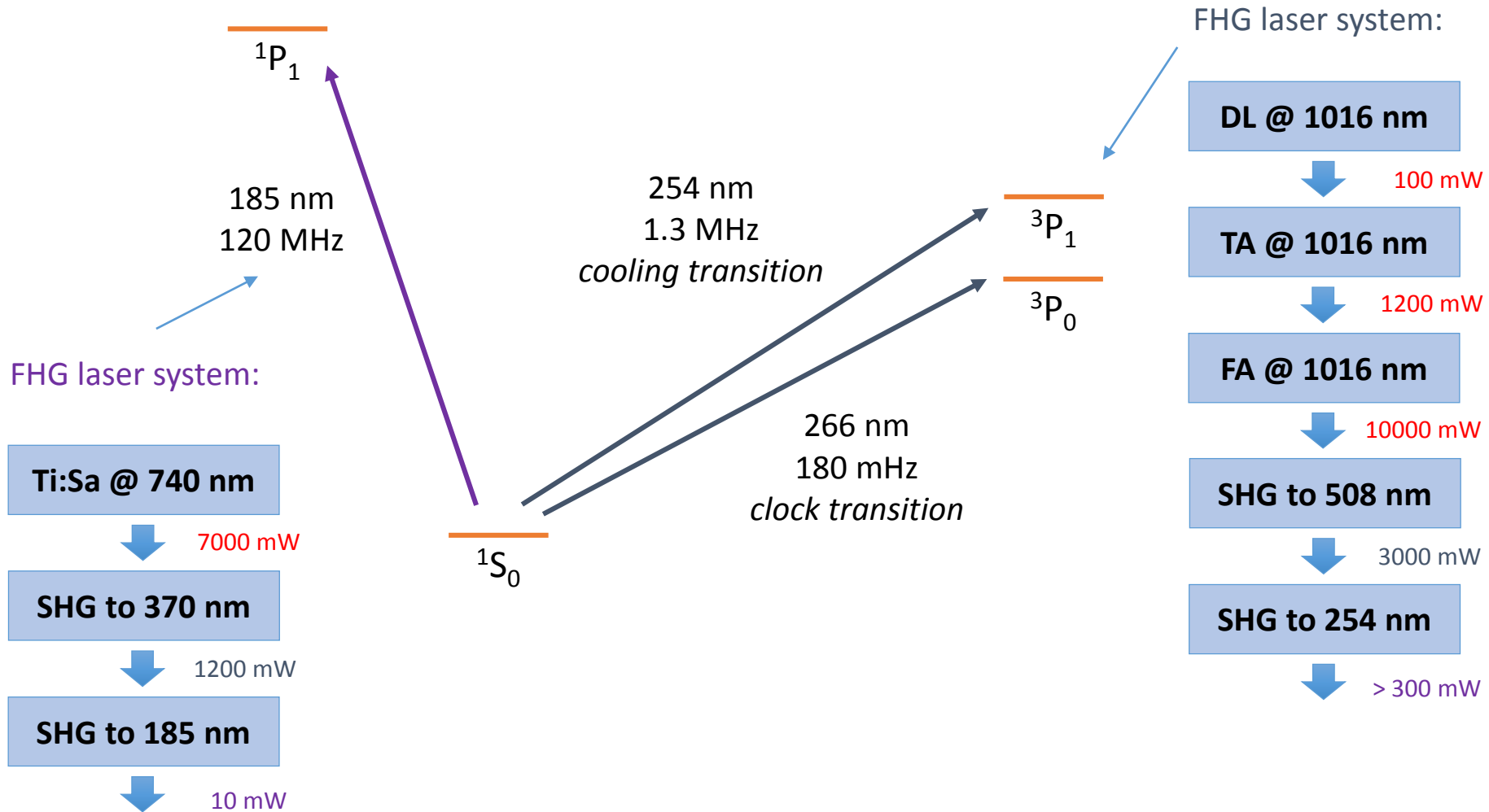


$$\text{EDM} \sim Z^3$$

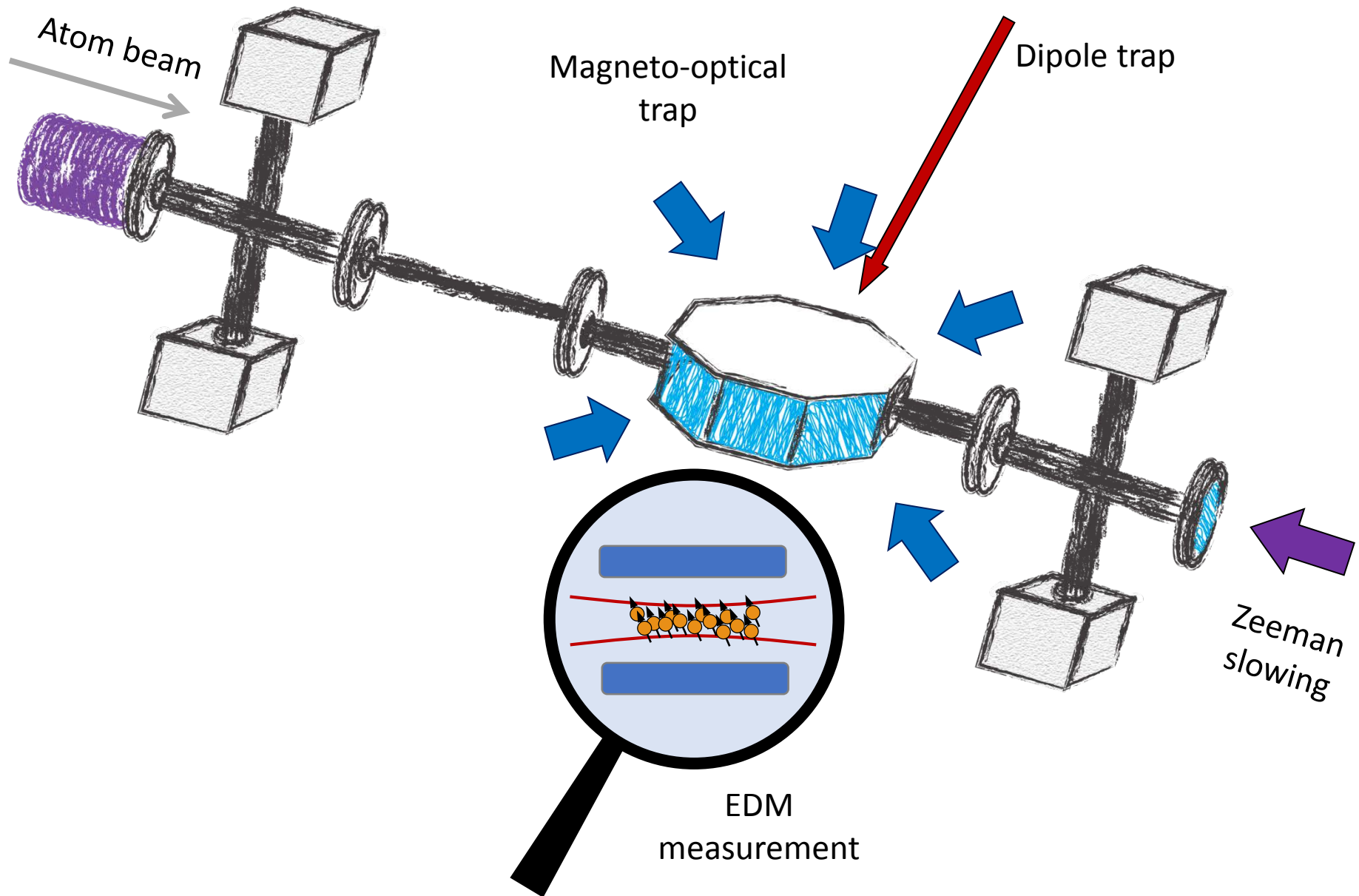
hydrogen 1 H 1.0079																			helium 2 He 4.0026
lithium 3 Li 6.941	beryllium 4 Be 9.0122											boron 5 B 10.811	carbon 6 C 12.011	nitrogen 7 N 14.007	oxygen 8 O 15.999	fluorine 9 F 18.998	neon 10 Ne 20.180		
sodium 11 Na 22.990	magnesium 12 Mg 24.305											aluminium 13 Al 26.982	silicon 14 Si 28.086	phosphorus 15 P 30.974	sulfur 16 S 32.065	chlorine 17 Cl 35.453	argon 18 Ar 39.948		
potassium 19 K 39.098	calcium 20 Ca 40.078	scandium 21 Sc 44.956	titanium 22 Ti 47.867	vanadium 23 V 50.942	chromium 24 Cr 51.996	manganese 25 Mn 54.938	iron 26 Fe 55.845	cobalt 27 Co 58.933	nickel 28 Ni 58.693	copper 29 Cu 63.546	zinc 30 Zn 65.39	gallium 31 Ga 69.723	germanium 32 Ge 72.61	arsenic 33 As 74.922	selenium 34 Se 78.96	bromine 35 Br 79.904	krypton 36 Kr 83.80		
rubidium 37 Rb 85.468	strontium 38 Sr 87.62	yttrium 39 Y 88.906	zirconium 40 Zr 91.224	niobium 41 Nb 92.906	molybdenum 42 Mo 95.94	technetium 43 Tc [98]	ruthenium 44 Ru 101.07	rhodium 45 Rh 102.91	palladium 46 Pd 106.42	silver 47 Ag 107.868	cadmium 48 Cd 112.41	indium 49 In 114.82	tin 50 Sn 118.71	antimony 51 Sb 121.76	tellurium 52 Te 127.60	iodine 53 I 126.90	xenon 54 Xe 131.29		
caesium 55 Cs 132.91	barium 56 Ba 137.33	57-70 *	lutetium 71 Lu 174.97	hafnium 72 Hf 178.49	tantalum 73 Ta 180.95	tungsten 74 W 183.84	rhenium 75 Re 186.21	osmium 76 Os 190.23	iridium 77 Ir 192.22	platinum 78 Pt 195.08	gold 79 Au 196.967	mercury 80 Hg 200.59	thallium 81 Tl 204.38	lead 82 Pb 207.2	bismuth 83 Bi 208.98	polonium 84 Po [209]	astatine 85 At [210]	radon 86 Rn [222]	
francium 87 Fr [223]	radium 88 Ra [226]	89-102 * *	lawrencium 103 Lr [262]	rutherfordium 104 Rf [261]	dubnium 105 Db [262]	seaborgium 106 Sg [266]	bohrium 107 Bh [264]	hassium 108 Hs [269]	meitnerium 109 Mt [268]	unnilium 110 Uun [271]	ununium 111 Uuu [272]	ununbium 112 Uub [277]	ununtrium 113 Uut [284]	flerovium 114 Fl [289]	moscovium 115 Mc [288]	livermorium 116 Lv [293]	tennessine 117 Ts [294]	oganeson 118 Og [294]	

lanthanum 57 La 138.91	cerium 58 Ce 140.12	praseodymium 59 Pr 140.91	neodymium 60 Nd 144.24	promethium 61 Pm [145]	samarium 62 Sm 150.36	europium 63 Eu 151.96	gadolinium 64 Gd 157.25	terbium 65 Tb 158.93	dysprosium 66 Dy 162.50	holmium 67 Ho 164.93	erbium 68 Er 167.26	thulium 69 Tm 168.93	ytterbium 70 Yb 173.04
actinium 89 Ac [227]	thorium 90 Th 232.04	protactinium 91 Pa 231.04	uranium 92 U 238.03	neptunium 93 Np [237]	plutonium 94 Pu [244]	americium 95 Am [243]	curium 96 Cm [247]	berkelium 97 Bk [247]	californium 98 Cf [251]	einsteinium 99 Es [252]	fermium 100 Fm [257]	mendelevium 101 Md [258]	nobelium 102 No [259]

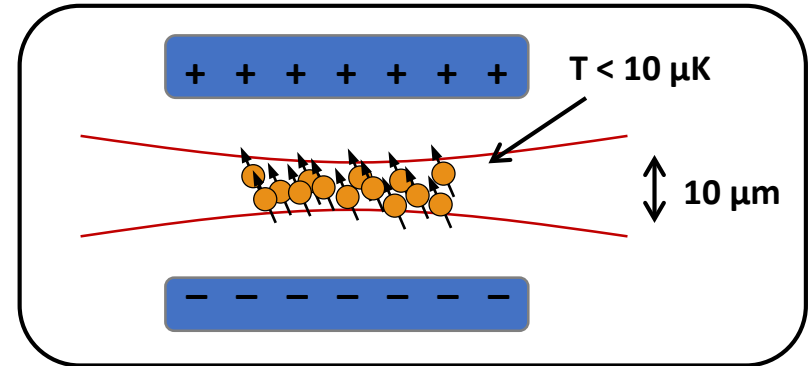
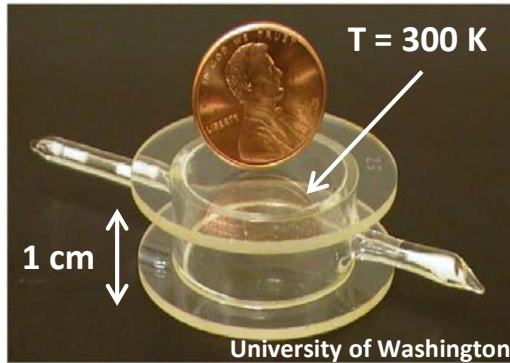
Mercury: a 2-electron atom



The experimental setup



Taking a classical experiment into the quantum world



2016 Seattle Hg-199 EDM experiment:

$$|d_{\text{Hg}}| < 7.4 \times 10^{-30} \text{ e cm}$$

quMercury Hg-199 EDM experiment

- improve control:
- no spin decoherence from collisions with the walls
 - no $E \times v$ effect from movement of the atoms
 - size reduces from 1 cm to a few 10 μm : homogeneous fields
 - in-vacuum electrodes: no leakage currents, much higher fields

improve sensitivity by at least a factor 10, **conservative estimate: $\delta d = 7 \times 10^{-31} \text{ e cm}$**

$$\delta d = \frac{\hbar}{2E \sqrt{2N\tau T}}$$

field E :	30 keV on a 0.5-mm gap
atom number N :	10^8 atoms
vacuum lifetime τ :	300 s
measurement time T :	3 months

- explore benefits of quantum nature:
- **Heisenberg scaling** through entanglement?
 - state engineering (squeezing, decoherence-free subspace)?
 - matter-wave interference?

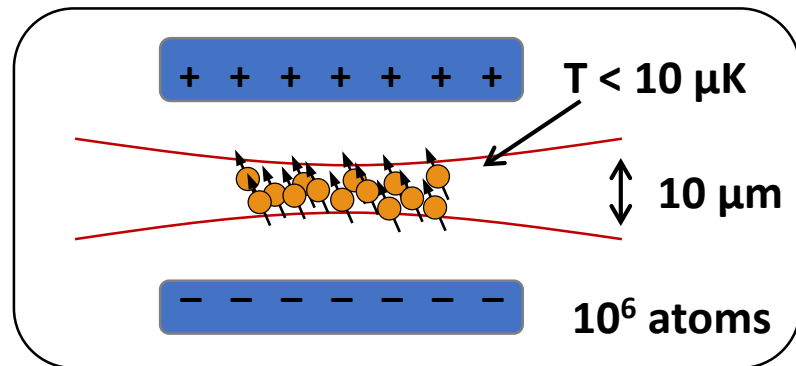
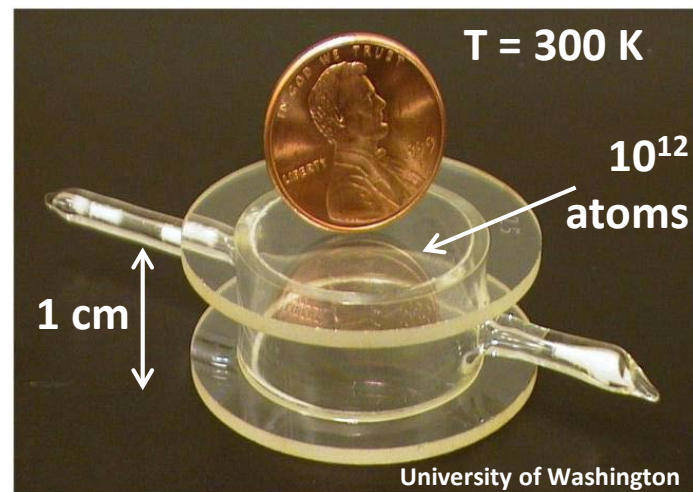
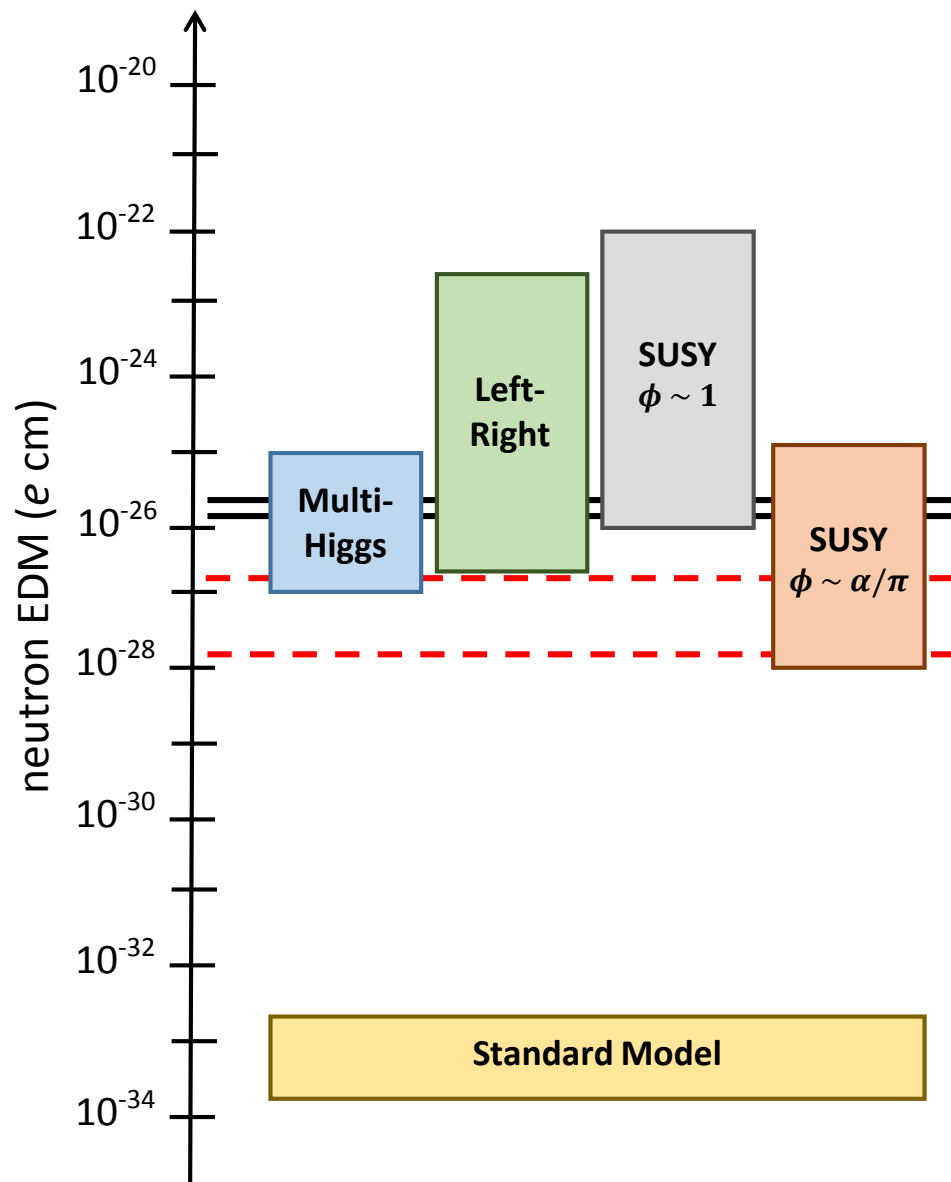
Expected sensitivity of the quMercury experiment



$$\delta d = \frac{\hbar}{2E \sqrt{2N\tau T}}$$

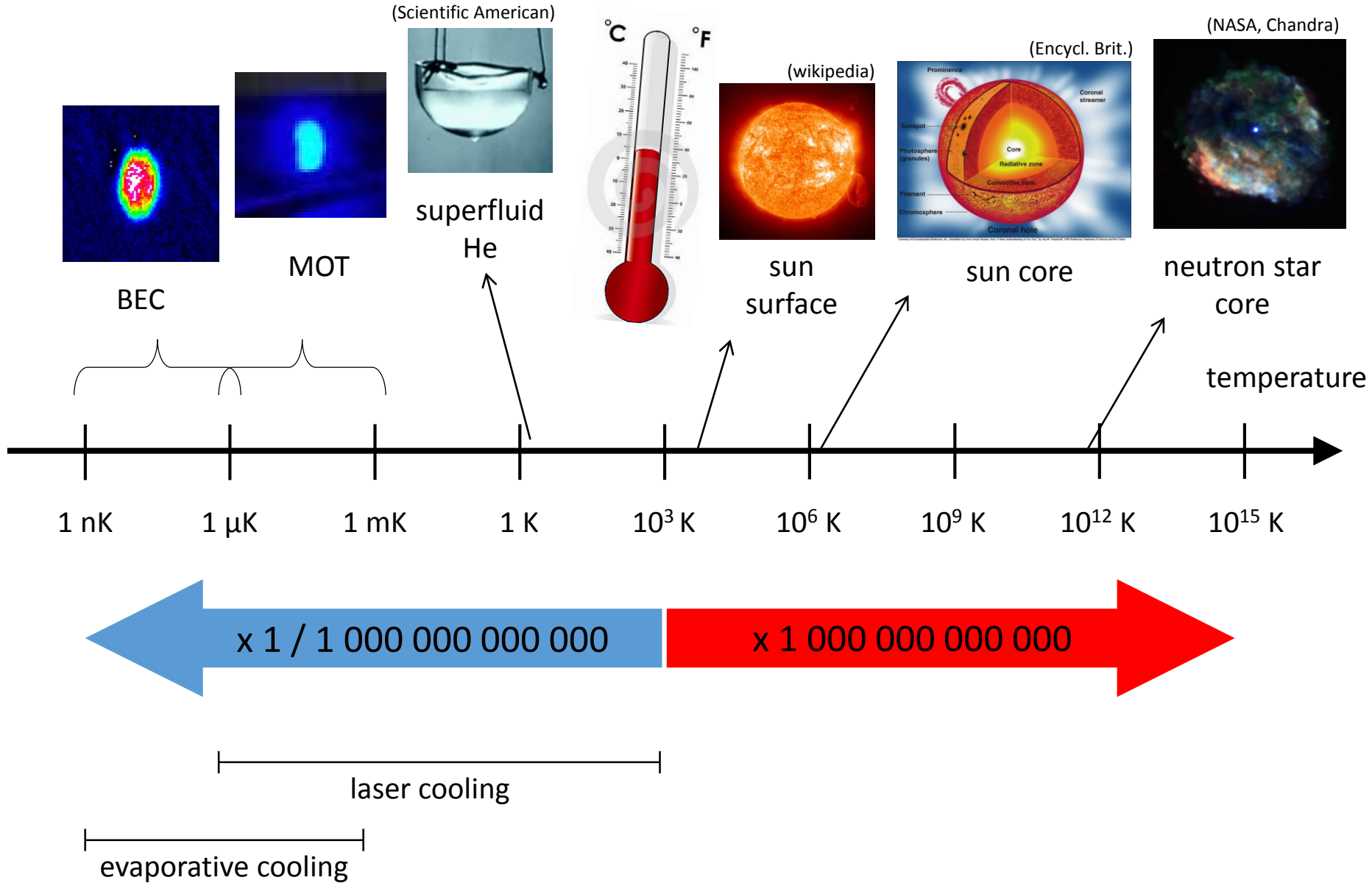
	Seattle (2016)	quMercury			
		first try	after 5 years	2nd generation	Heisenberg
voltage	15 kV	10 kV	30 kV	30 kV	30 kV
electrode gap	1.1 cm	0.5 mm	0.5 mm	0.2 mm	1 mm
E-field	13.6 kV/cm	200 kV/cm	600 kV/cm	1500 kV/cm	300 kV/cm
spin decoherence τ	100 s	100 s	300 s	300 s	100 s
atom number N	2×10^{14}	10^6	10^8	10^9	10^6
measurement time T	10 months	3 months	3 months	6 months	3 months
sensitivity δd	$7.4 \times 10^{-30} e \text{ cm}$	$1.5 \times 10^{-28} e \text{ cm}$	$7.6 \times 10^{-31} e \text{ cm}$	$7.1 \times 10^{-32} e \text{ cm}$	$2.6 \times 10^{-32} e \text{ cm}$
comments	Nominal δd of $2 \times 10^{-33} e \text{ cm}$ not reached because of leakage currents.	Already six orders of magnitude better than Rn-221 and Ra-225 experiments in a dipole trap.	Improves Seattle experiment by factor 10.	Improves Seattle experiment by factor 100.	

Expected sensitivity of the quMercury experiment

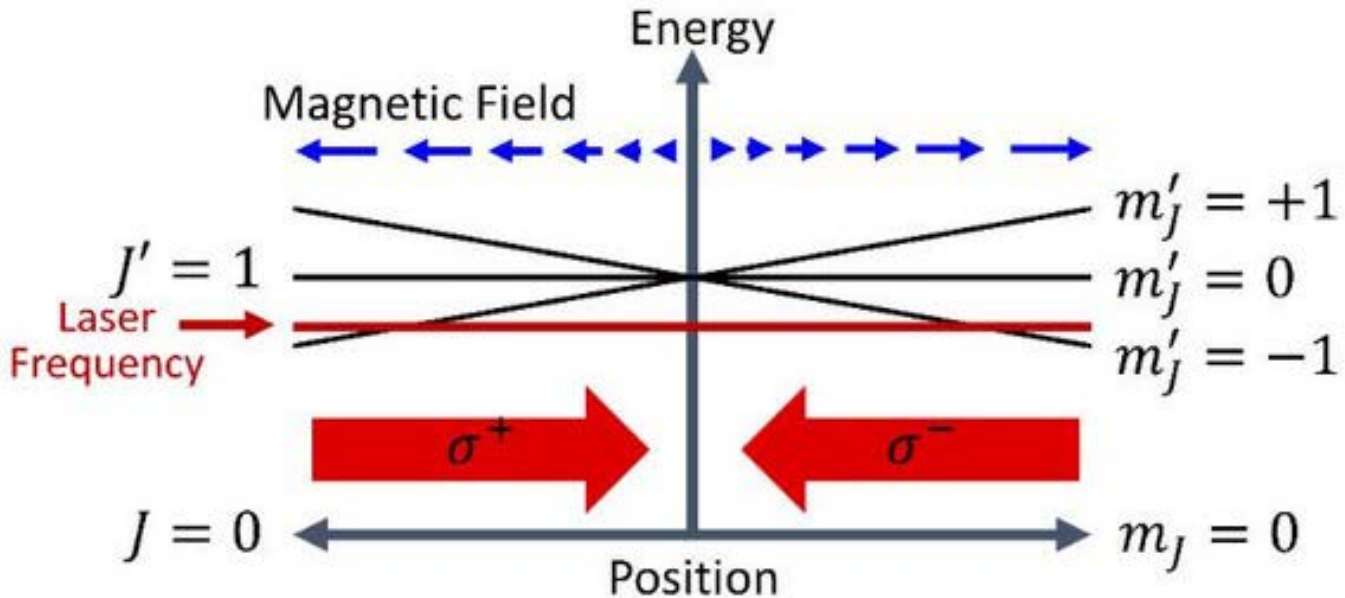
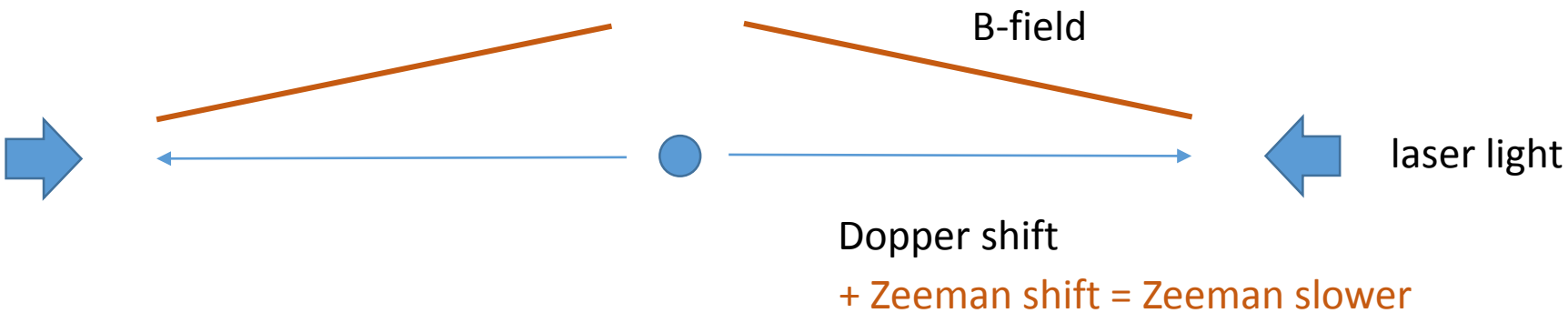


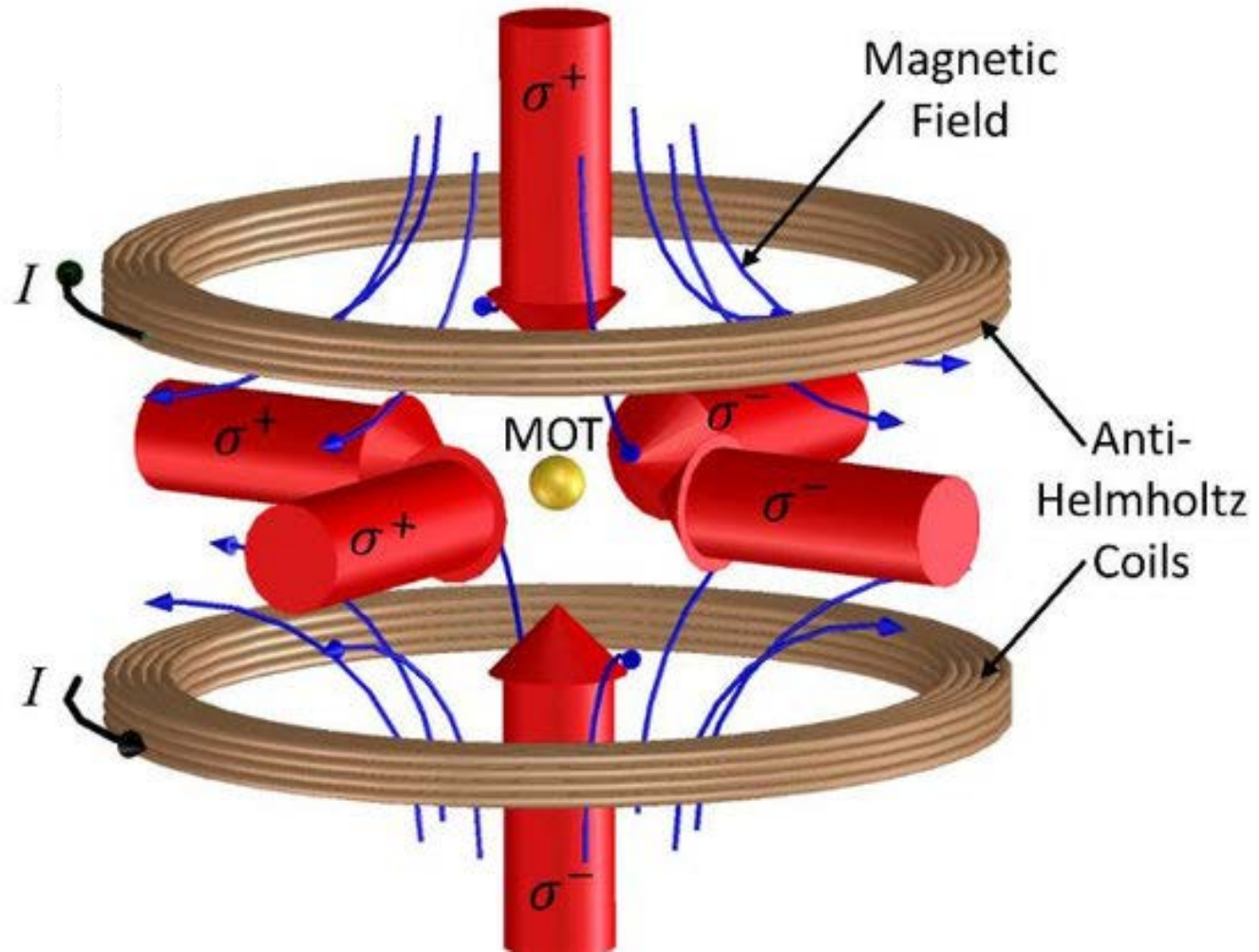


Laser cooling is a powerful tool



Laser cooling

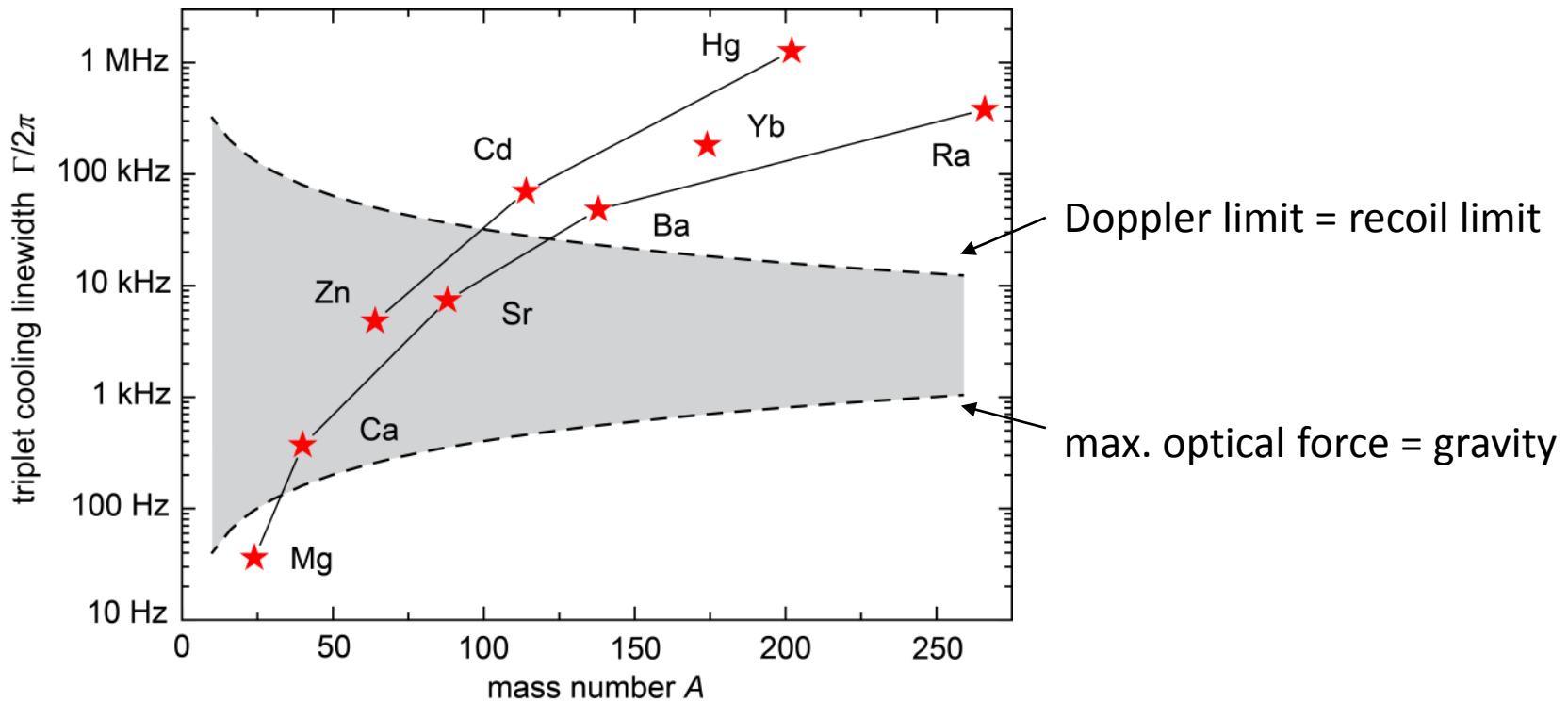




$$T_D = \hbar\Gamma / (2k_B)$$

... but we can do better: sub-Doppler cooling

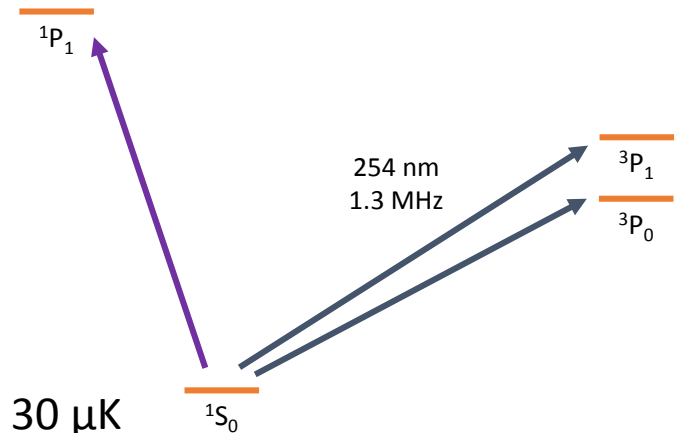
Narrow-line cooling



$$T_D = \hbar\Gamma / (2k_B)$$

$$T_r = \hbar^2 k^2 / (k_B m)$$

Hg Doppler temperature: 30 μ K





week ending
8 FEBRUARY 2008

PRL 100, 053001 (2008)

PHYSICAL REVIEW LETTERS

Trapping of Neutral Mercury Atoms and Prospects for Optical Lattice Clocks

H. Hachisu,^{1,2} K. Miyagishi,¹ S. G. Porsev,^{3,4} A. Derevianko,^{4,5} V. D. Ovsiannikov,⁶
V. G. Pal'chikov,⁷ M. Takamoto,^{1,2} and H. Katori^{1,2}

¹Department of Applied Physics, Graduate School of Engineering, The University of Tokyo, Bunkyo-ku, Tokyo 113-8656, Japan

²CREST, Japan Science and Technology Agency, 4-1-8 Honcho Kawaguchi, Saitama, Japan

³Petersburg Nuclear Physics Institute, Gatchina, Leningrad District, 188300, Russia

⁴Physics Department, University of Nevada, Reno, Nevada 89557, USA

⁵Laboratoire Aimé Cotton, Bâtiment 505, Campus d'Orsay, 91405 Orsay Cedex, France

⁶Physics Department, Voronezh State University, Universitetskaya ploschad 1, 394006, Voronezh, Russia

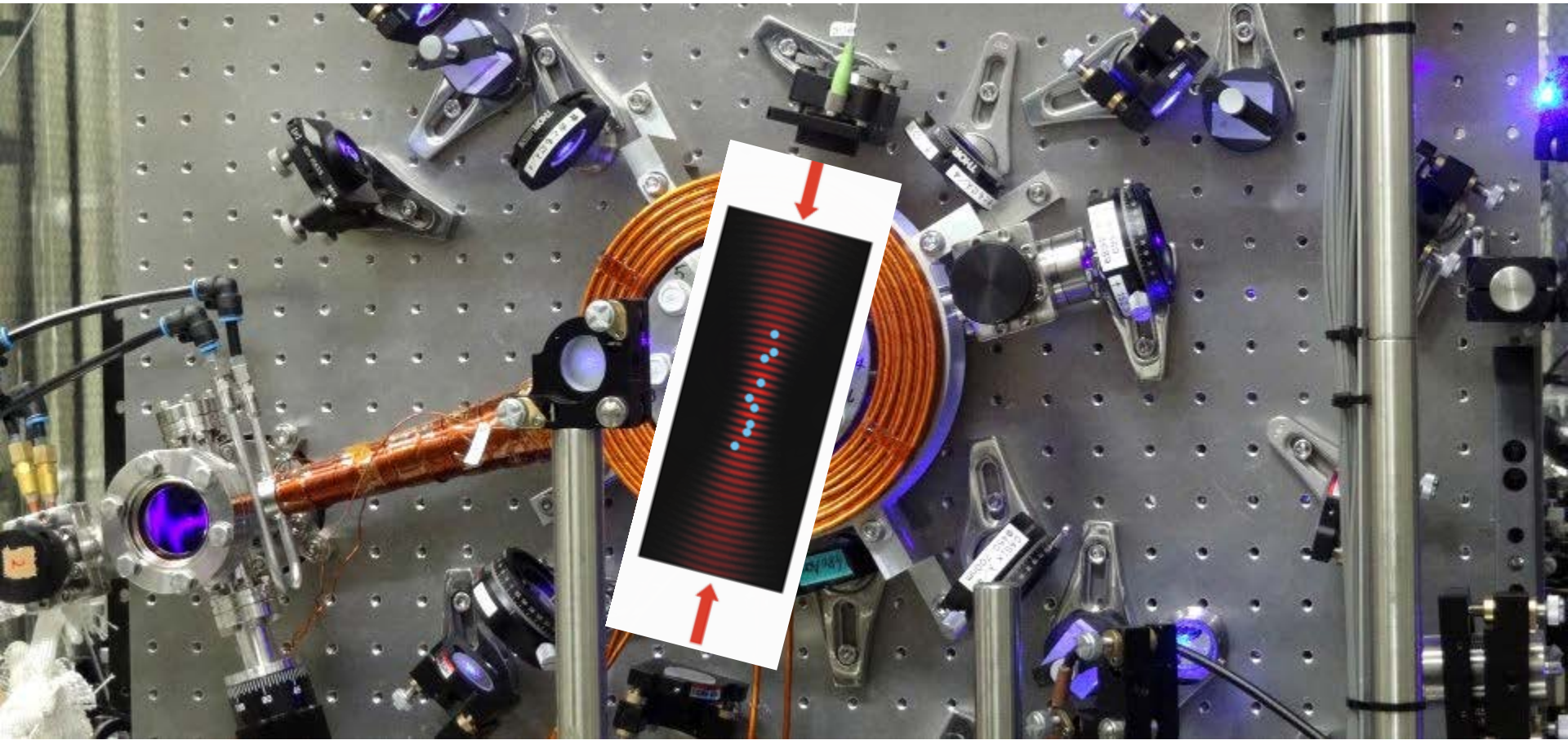
⁷Institute of Metrology for Time and Space at National Research Institute for Physical-Technical and Radiotechnical Measurements, Mendeleevo, Moscow Region, 141579, Russia

(Received 16 October 2007; revised manuscript received 28 November 2007; published 8 February 2008)

We report vapor-cell magneto-optical trapping of Hg isotopes on the 1S_0 - 3P_1 intercombination transition. Six abundant isotopes, including four bosons and two fermions, were trapped. Hg is the heaviest nonradioactive atom trapped so far, which enables sensitive atomic searches for “new physics” beyond the standard model. We propose an accurate optical lattice clock based on Hg and evaluate its systematic accuracy to be better than 10^{-18} . Highly accurate and stable Hg-based clocks will provide a new avenue for the research of optical lattice clocks and the time variation of the fine-structure constant.

Katori group in 2008:

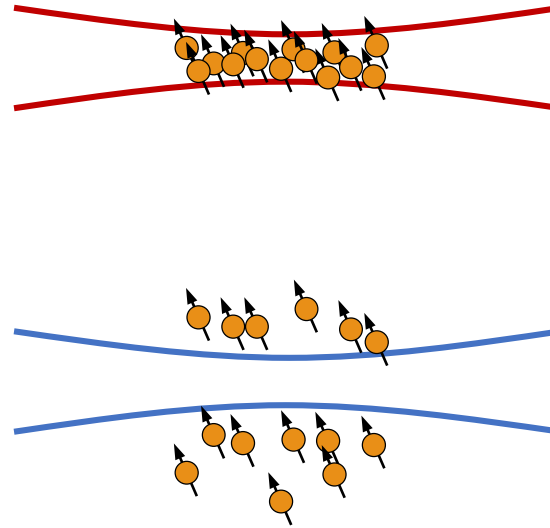
500,000 atoms of Hg-199 at 50 μ K



Katori group, RIKEN (Sr lattice clock)



NP 2018:
Arthur Ashkin
optical tweezers



$$U_{\text{dip}}(\mathbf{r}) = -\frac{3\pi c^2}{2\omega_0^3} \left(\frac{\Gamma}{\omega_0 - \omega} + \frac{\Gamma}{\omega_0 + \omega} \right) I(\mathbf{r})$$

resonance frequency
of the atom

frequency of the
laser

Scattering properties are important (and unknown)

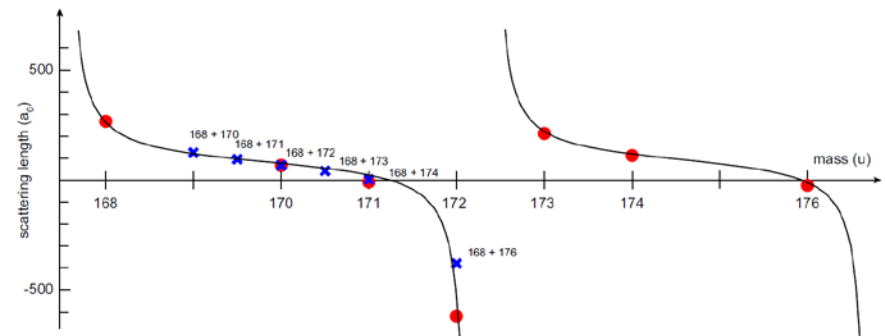
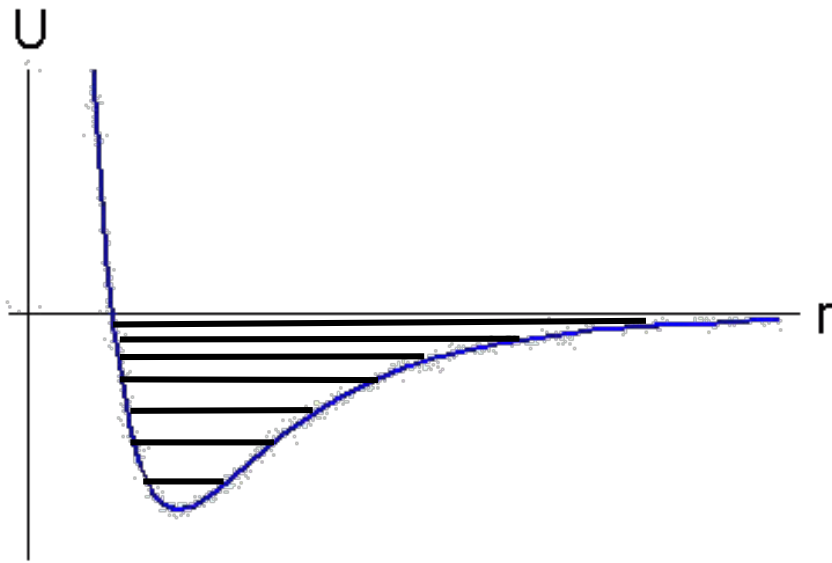


Evaporative cooling...

- is an efficient way to increase phase space density
- relies on re-thermalization of the atoms
- works only within a certain range of scattering properties

scattering cross section: $\sigma = 4\pi a_s^2$

isotope	character	I	abundance
^{196}Hg	boson	0	0.15 %
^{198}Hg	boson	0	9.97 %
^{199}Hg	fermion	$1/2$	16.87 %
^{200}Hg	boson	0	23.10 %
^{201}Hg	fermion	$3/2$	13.18 %
^{202}Hg	boson	0	29.86 %
^{204}Hg	boson	0	6.87 %



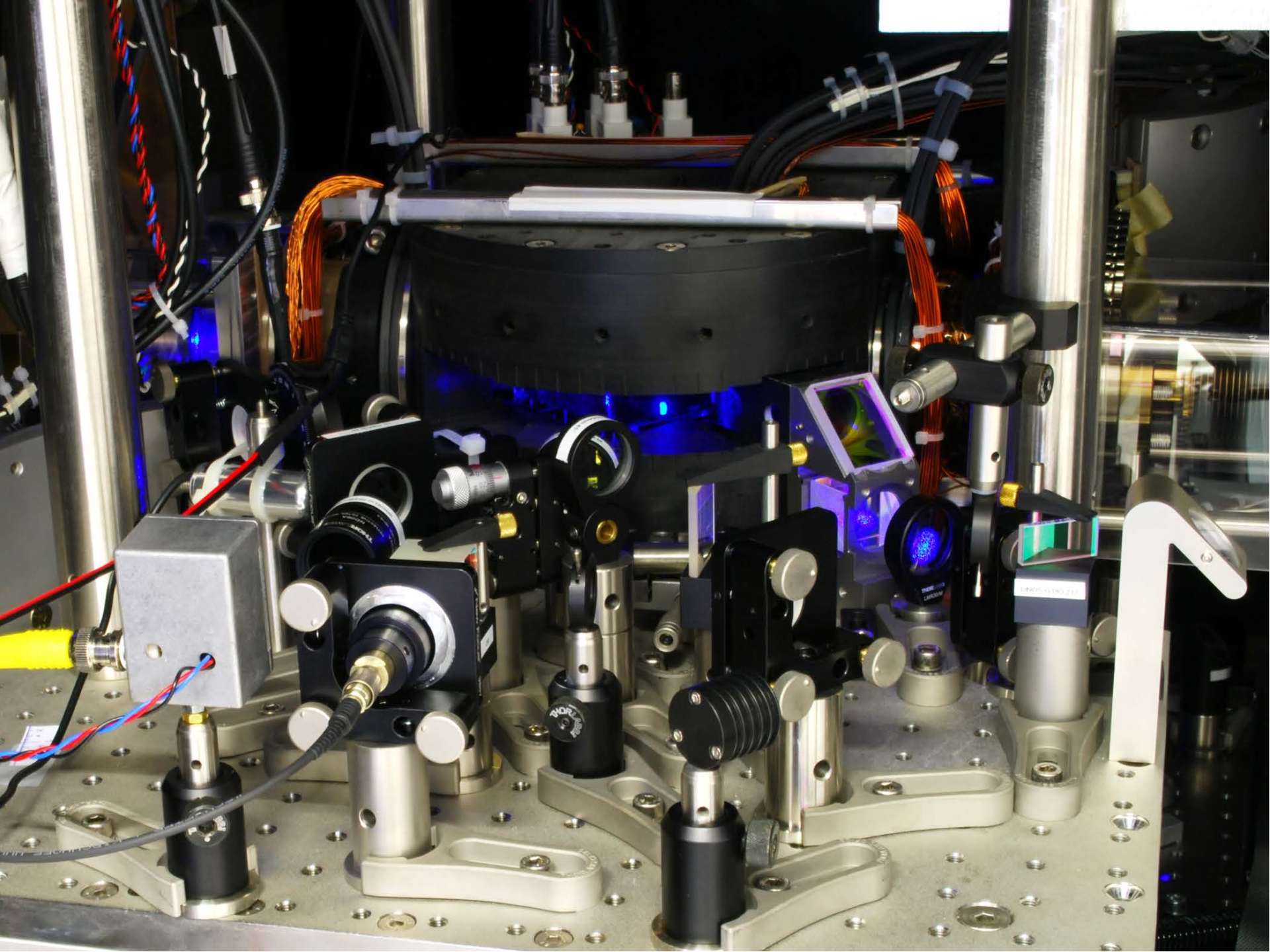
mass scaling in Yb

Strontium, very similar to mercury



hydrogen 1 H 1.0079																	helium 2 He 4.0026						
lithium 3 Li 6.941	beryllium 4 Be 9.0122																	boron 5 B 10.811	carbon 6 C 12.011	nitrogen 7 N 14.007	oxygen 8 O 15.999	fluorine 9 F 18.998	neon 10 Ne 20.180
sodium 11 Na 22.990	magnesium 12 Mg 24.305																	aluminium 13 Al 26.982	silicon 14 Si 28.086	phosphorus 15 P 30.974	sulfur 16 S 32.065	chlorine 17 Cl 35.453	argon 18 Ar 39.948
potassium 19 K 39.098	calcium 20 Ca 40.078	scandium 21 Sc 44.956	titanium 22 Ti 47.867	vanadium 23 V 50.942	chromium 24 Cr 51.996	manganese 25 Mn 54.938	iron 26 Fe 55.845	cobalt 27 Co 58.933	nickel 28 Ni 58.693	copper 29 Cu 63.546	zinc 30 Zn 65.39	gallium 31 Ga 69.723	germanium 32 Ge 72.61	arsenic 33 As 74.922	selenium 34 Se 78.96	bromine 35 Br 79.904	krypton 36 Kr 83.80						
rubidium 37 Rb 85.468	strontium 38 Sr 87.62	yttrium 39 Y 88.906	zirconium 40 Zr 91.224	niobium 41 Nb 92.906	molybdenum 42 Mo 95.94	technetium 43 Tc [98]	ruthenium 44 Ru 101.07	rhodium 45 Rh 102.91	palladium 46 Pd 106.42	silver 47 Ag 107.87	cadmium 48 Cd 112.41	indium 49 In 114.82	tin 50 Sn 118.71	antimony 51 Sb 121.76	tellurium 52 Te 127.60	iodine 53 I 126.90	xenon 54 Xe 131.29						
caesium 55 Cs 132.91	barium 56 Ba 137.33	57-70 *	lutetium 71 Lu 174.97	hafnium 72 Hf 178.49	tantalum 73 Ta 180.95	tungsten 74 W 183.84	rhenium 75 Re 186.21	osmium 76 Os 190.23	iridium 77 Ir 192.22	platinum 78 Pt 195.08	gold 79 Au 196.97	mercury 80 Hg 200.59	thallium 81 Tl 204.38	lead 82 Pb 207.2	bismuth 83 Bi 208.98	polonium 84 Po [209]	astatine 85 At [210]	radon 86 Rn [222]					
francium 87 Fr [223]	radium 88 Ra [226]	89-102 * *	lawrencium 103 Lr [262]	rutherfordium 104 Rf [261]	dubnium 105 Db [262]	seaborgium 106 Sg [266]	bohrium 107 Bh [264]	hassium 108 Hs [269]	meitnerium 109 Mt [268]	ununnilium 110 Uun [271]	ununnilium 111 Uun [272]	ununnilium 112 Uub [277]	ununnilium 113 Nh [284]	ununnilium 114 Fl [289]	ununnilium 115 Mc [288]	ununnilium 116 Lv [293]	ununnilium 117 Ts [294]	ununnilium 118 Og [294]					

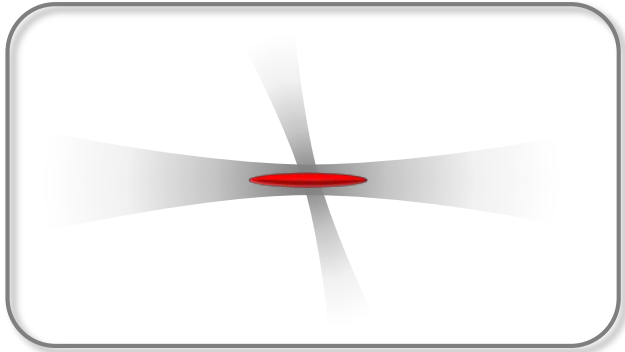
lanthanum 57 La 138.91	cerium 58 Ce 140.12	praseodymium 59 Pr 140.91	neodymium 60 Nd 144.24	promethium 61 Pm [145]	samarium 62 Sm 150.36	europium 63 Eu 151.96	gadolinium 64 Gd 157.25	terbium 65 Tb 158.93	dysprosium 66 Dy 162.50	holmium 67 Ho 164.93	erbium 68 Er 167.26	thulium 69 Tm 168.93	ytterbium 70 Yb 173.04
actinium 89 Ac [227]	thorium 90 Th 232.04	protactinium 91 Pa 231.04	uranium 92 U 238.03	neptunium 93 Np [237]	plutonium 94 Pu [244]	americium 95 Am [243]	curium 96 Cm [247]	berkelium 97 Bk [247]	californium 98 Cf [251]	einsteinium 99 Es [252]	fermium 100 Fm [257]	mendelevium 101 Md [258]	nobelium 102 No [259]



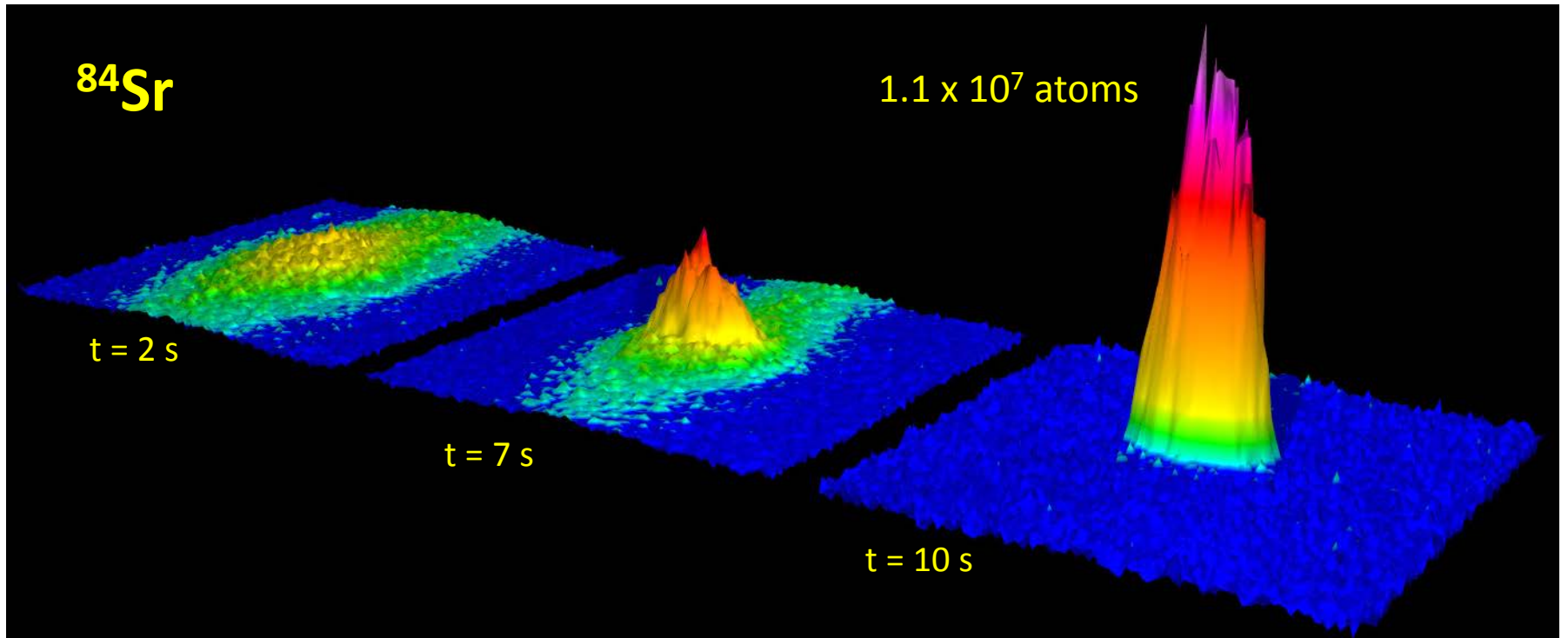
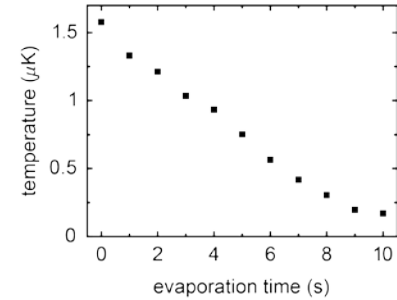
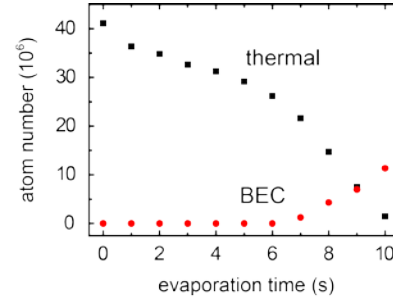
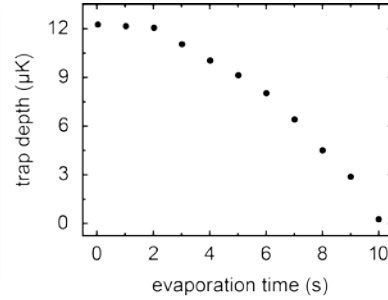
Bose-Einstein condensation of Strontium



loading of the dipole trap



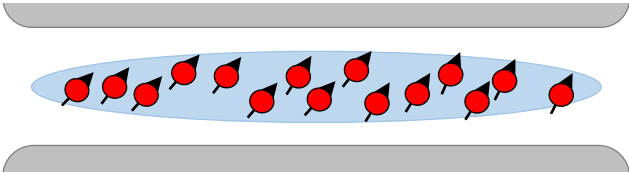
evaporation



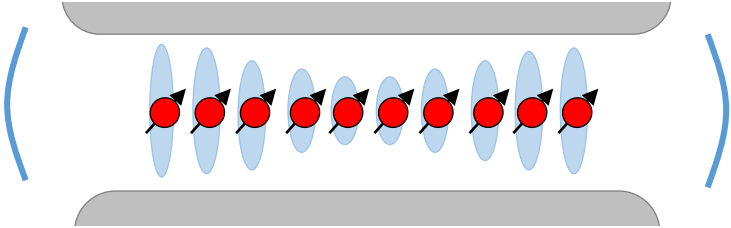
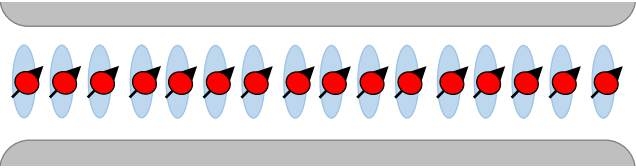
Is there an advantage of quantum-degenerate atoms?

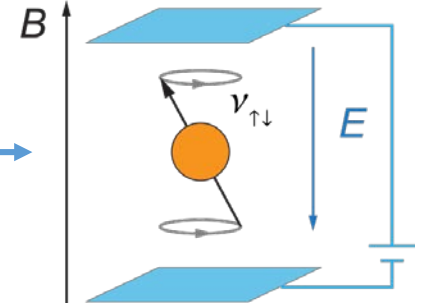
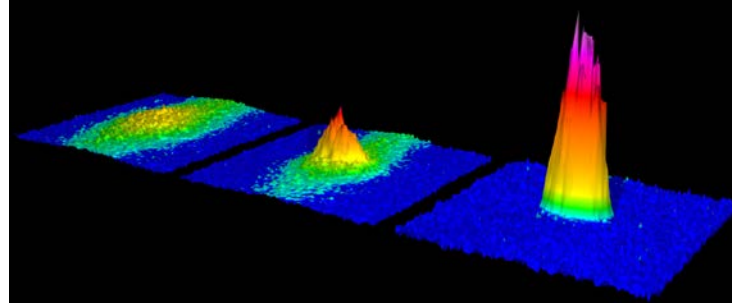


$$\delta d = \frac{\hbar}{2E \sqrt{2N\tau T}}$$



?





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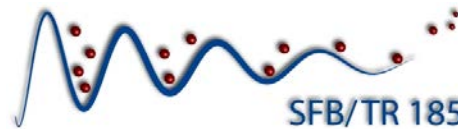
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