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Federal Institute of Metrology METAS



A milestone in the evolution of the International System of Units

Beat Jeckelmann

Metrology: the science and application of measurement

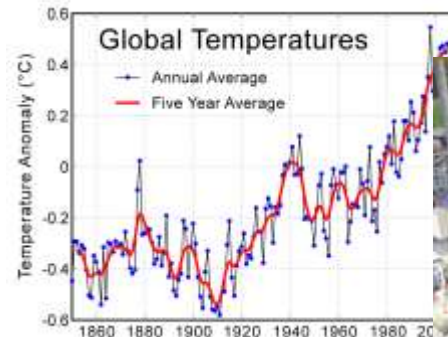
secure measurements
and reference data

do you get
what you
pay for?



safely below
radiation limits?

data quality for
policy making
sufficient?



underpin
basic science



do constants
of nature drift with time?

drive innovation



efficiency?
quality control
in production?

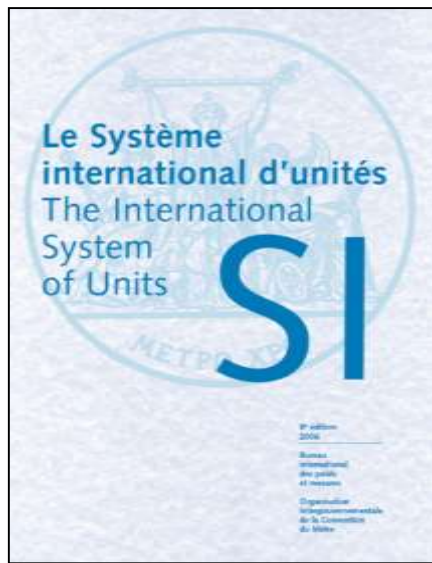
precise enough
to be
competitive?



Worldwide comparability

Metre Convention

- Basis for the International System of Units (SI)
- And a global measurement infrastructure



Mutual recognition arrangement:

→ 97.6 % of the world economy

→ The cornerstone of international Quality Infrastructure (QI)

→ Basis and condition sine qua non for world wide trade

Versailles, 16. November 2018.....

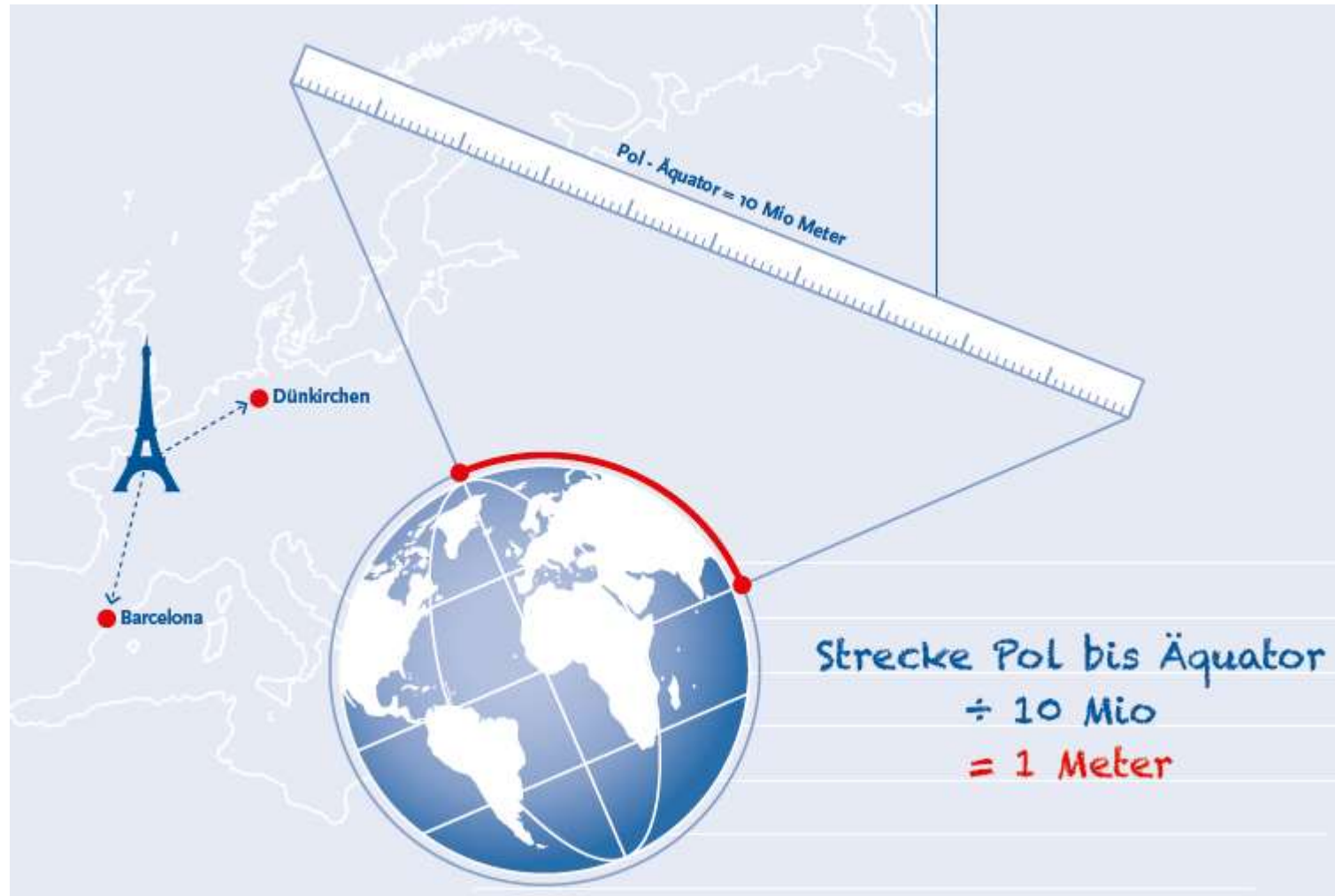


Outline

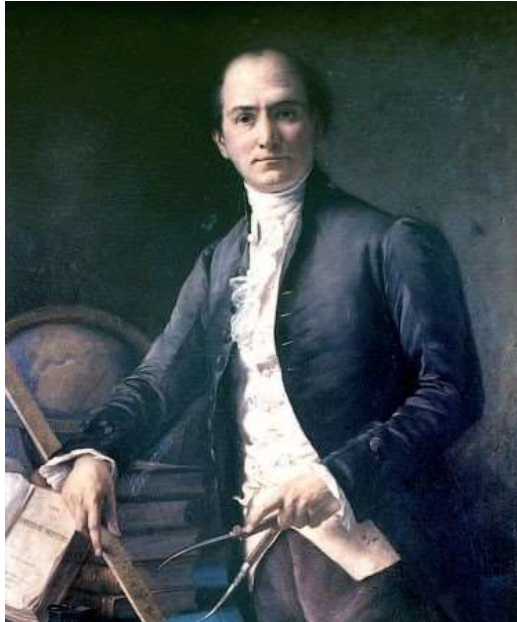
- The International System of Units SI: short history; structure
- Shortcomings of the present SI
- The concept of the revised SI
- Challenges:
 - Replacing the kg: Determination of the Planck constant
 - Boltzmann constant
- Conclusions



1791: Metric system



Meridian expedition



Jean-Baptiste J. Delambre
(northern part)



Pierre Méchains
(southern part)

1792 - 1798

Measurement of the
length of the meridian arc
through triangulation

Mètre des archives 1799



1801: Helvetic metre (iron)

A tous les temps, à tous les peuples....



France, loi du 4 juillet 1837

"...à partir du **1 janvier 1840**, tous poids et mesures autres que les poids établis par les lois des 18 germinal an III et 19 frimaire an VIII, constitutives du système métrique décimal, seront interdits sous les peines portées par l'article 479 du code pénal".

The metre convention 1875



CGPM: General Conference of Weights and Measures

BIPM: International Bureau of Weights and Measures

CCs: Consultative Committees

CCU: Consultative Committee for Units

1889: Prototype metre and kilogram

World Geodetic System 1984
meridian arc between pole and equator:
10001.966 km
Deviation: 200 ppm



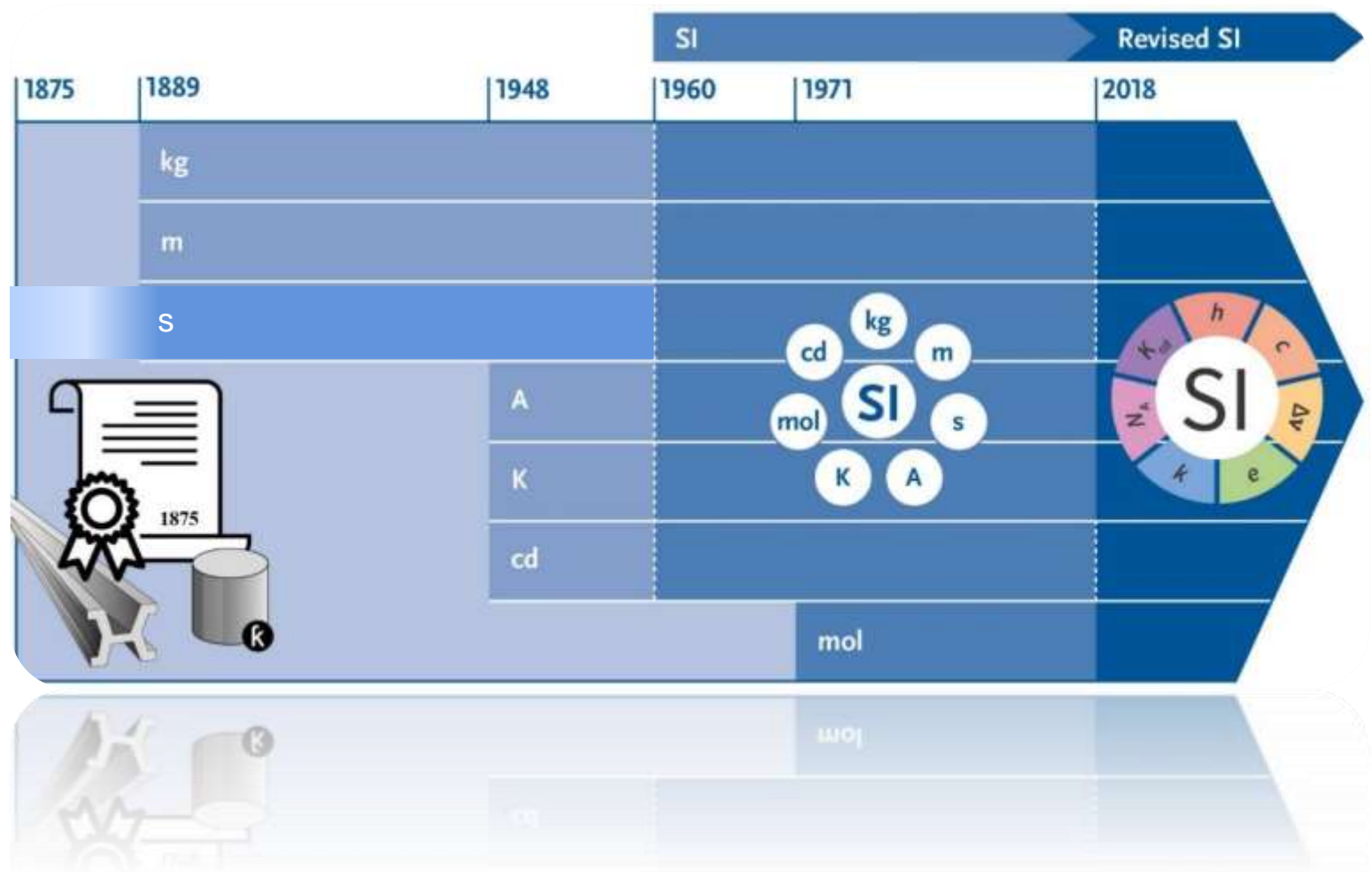
Photo NIST

90% Platin, 10% Iridium

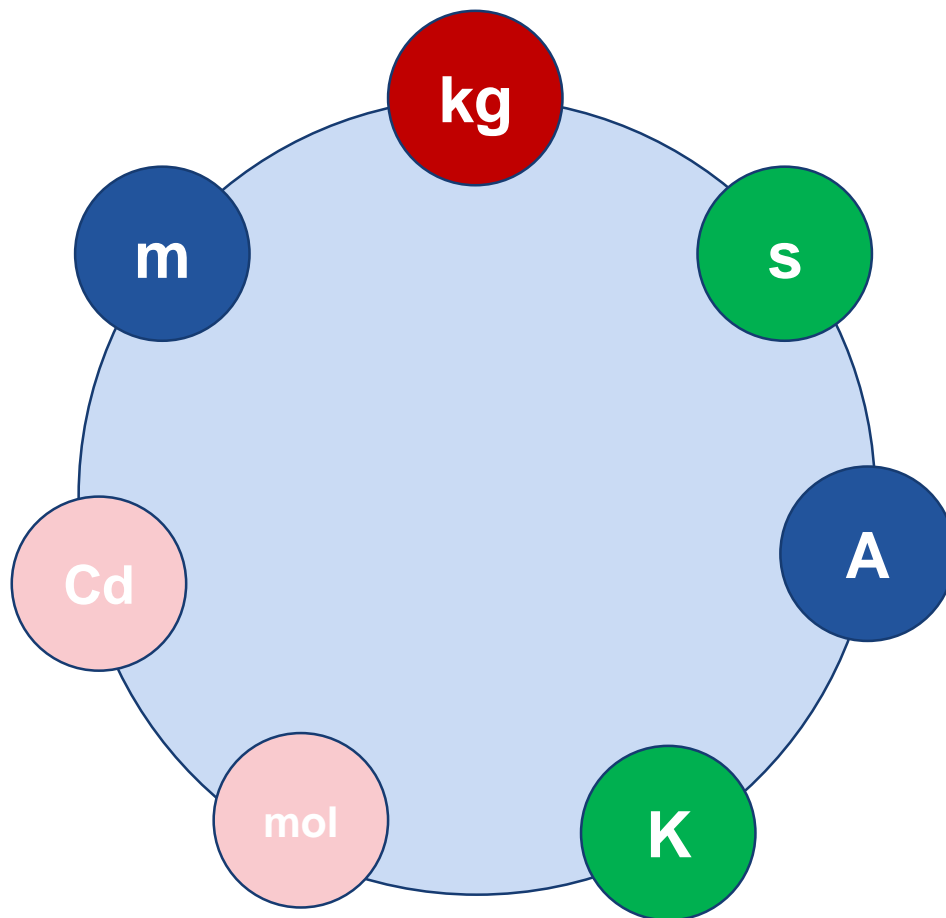


1 kg = 1 dm³ water at 4°C

Short history of the metric system



The SI today



- **7 Base units**
- Values chosen by convention
- Derived units

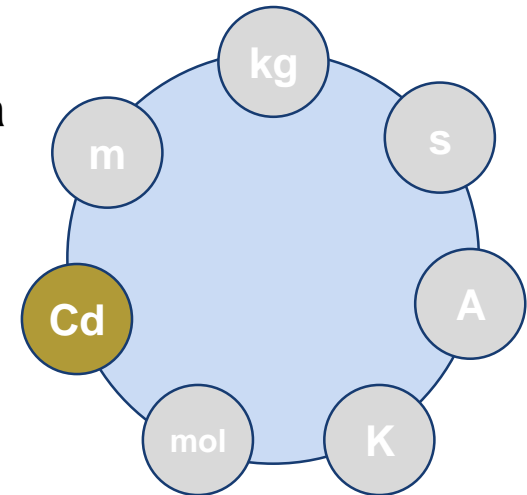
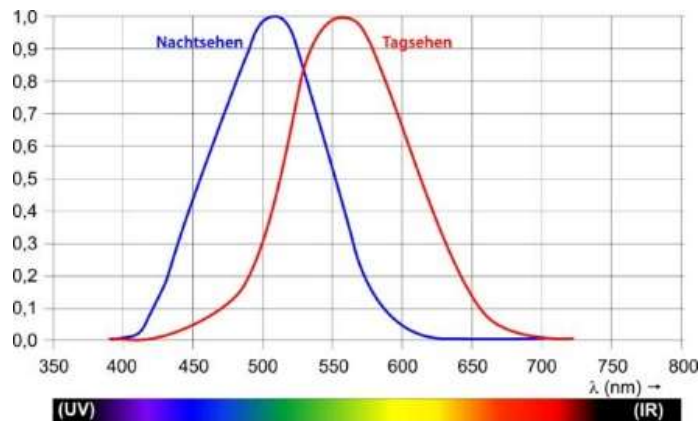
Various types of unit definitions, based on:

- **Artefact**
- **Physical state**
- **Fundamental constant**
- **“Technical” constant**

Practical units

Photometric unit

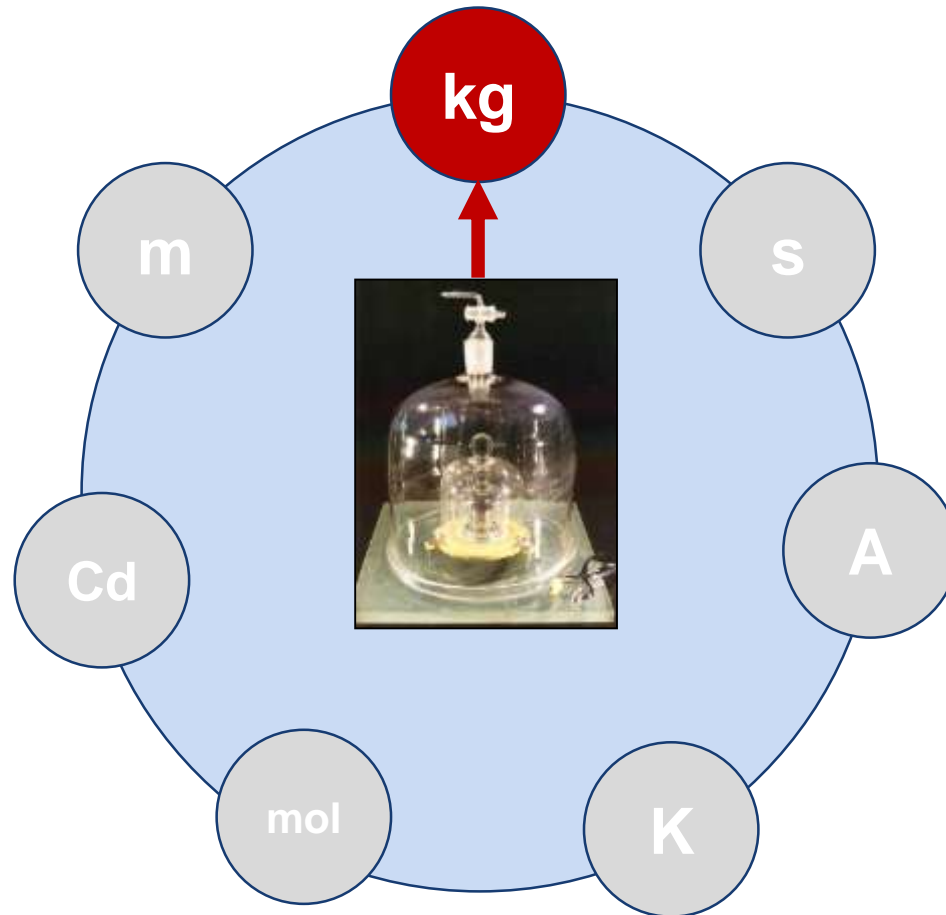
- 1 Candela = Luminous intensity of a source in a given direction,
- monochromatic radiation 540 THz,
- Radiant intensity = $1/683 \text{ W/sr}$



Measurement of light adapted to the human eye

Unit definitions and shortcomings

The kilogram



The kilogram: present definition

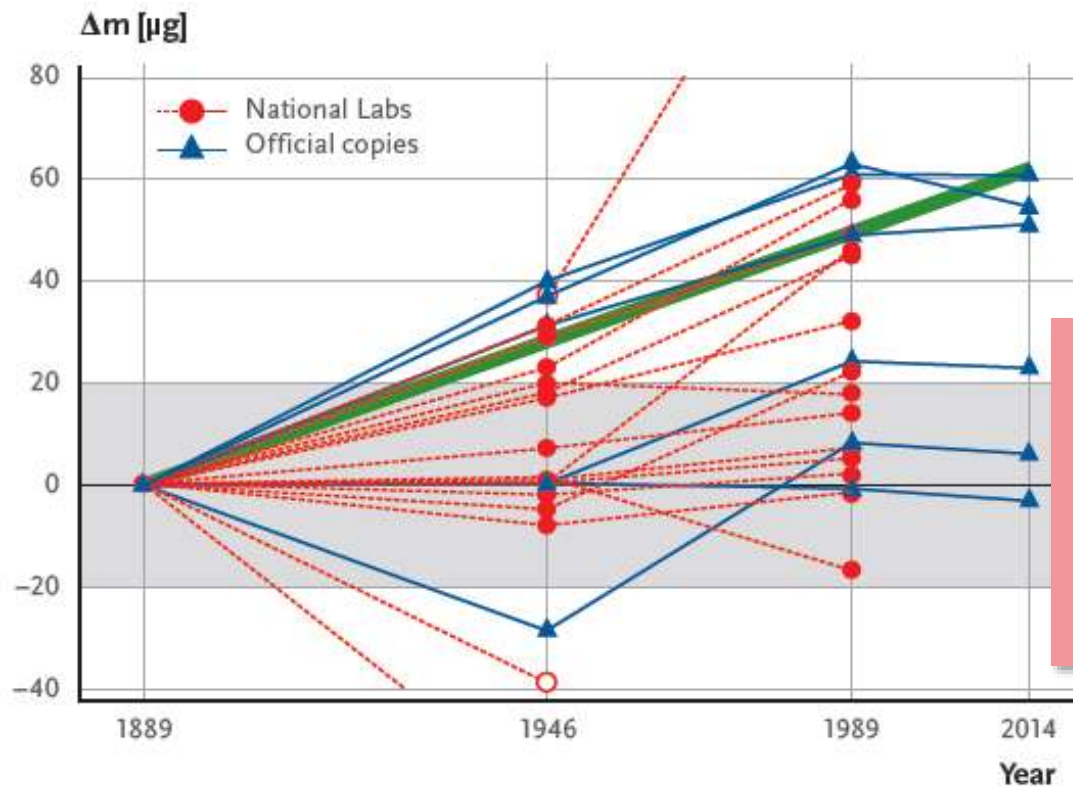
"The kilogram is the unit of mass; it is equal to the mass of the international prototype of the kilogram." (CGPM, 1901)

... immediately after cleaning and washing by a specified method (*mise en pratique*, CIPM 1989).



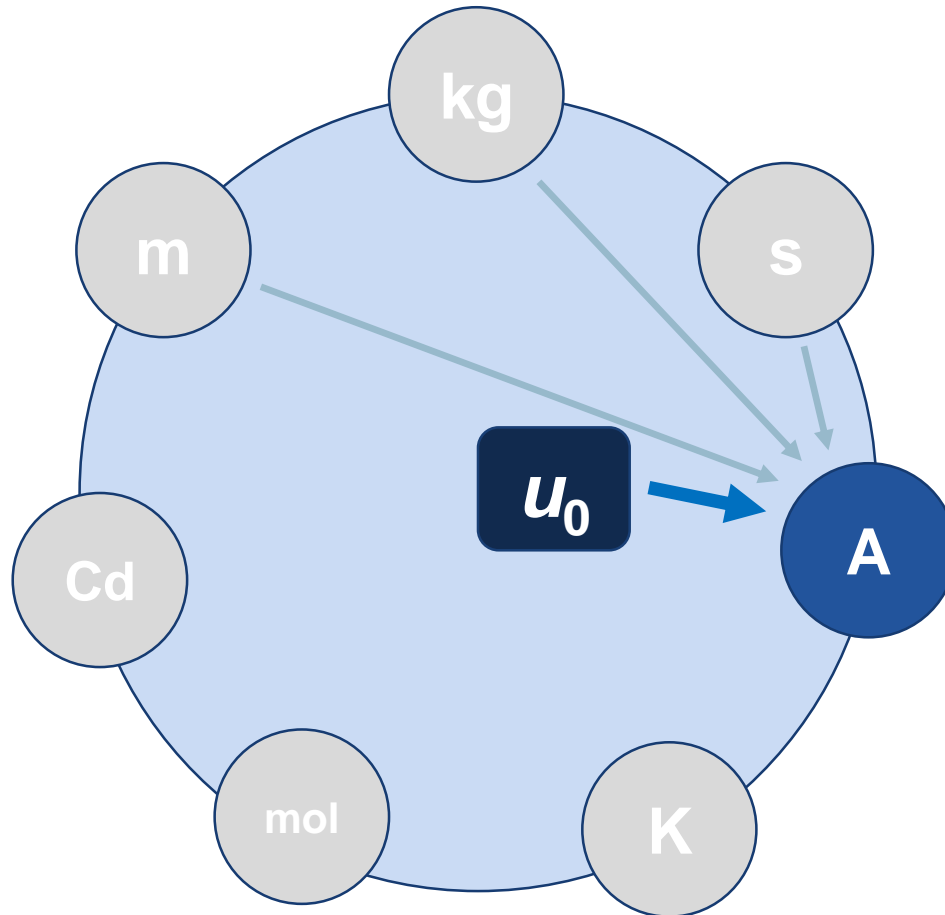
Periodical verification of the kg

~ 50 μg / 100 years

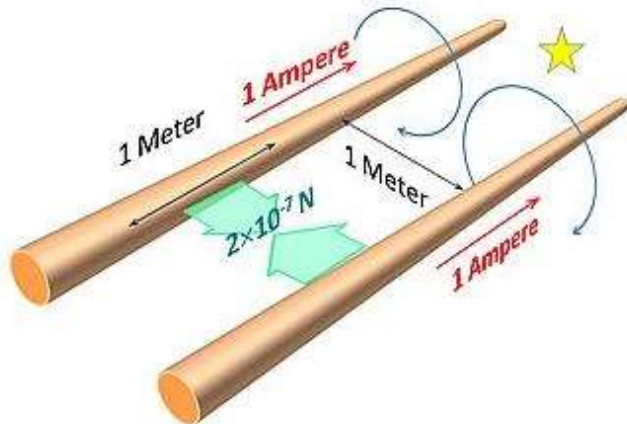


- unique
- local
- instable
- could be contaminated, damaged, stolen,.....

The ampere



Ampere definition



$$\frac{F}{l} = 2 \cdot 10^{-7} \frac{\text{N}}{\text{m}}$$

Ampère law for the chosen ideal case:

$$\frac{F}{l} = \mu_0 \frac{I^2}{2\pi r}$$

Using the ampere definition and equating mechanical and electrical power (1 Nm/s = 1 VA) leads to:

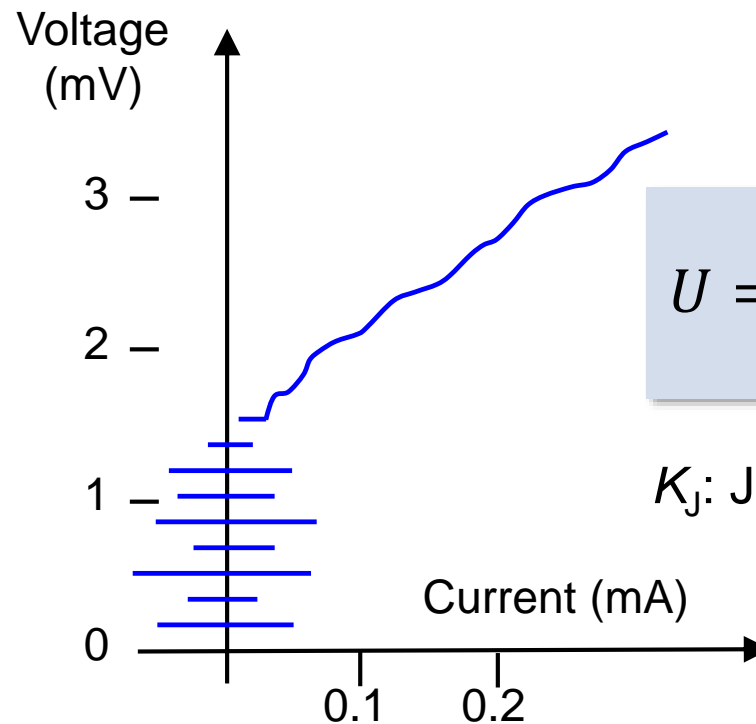
$$\mu_0 = 2\pi \frac{F}{I^2} \frac{r}{l} = 4\pi \cdot 10^{-7} \frac{\text{N}}{\text{A}^2}$$

Electrical quantum standards

- Quantum mechanical effects allow the realization of highly reproducible electrical standards:



B. Josephson predicts quantized voltage steps in superconductors (1962) → voltage standard



$$U = \frac{h}{2e} i \cdot f_J = \frac{i}{K_J} f_J$$

K_J : Josephson constant

Quantum voltmeter

Commercially available

PROGRAMMABLE JOSEPHSON VOLTAGE

70'000 junctions
in series!

The centre piece of the AC Quantum Voltmeter is a 10 Volt programmable Josephson voltage standard circuit

- Number of Josephson junctions: 69632
- Maximum output voltage: ± 10.1 V
- Operating frequency: 70 GHz
- Zero & first order Shapiro step: 1 mA
- Bias current: ± 6 mA
- Voltage increment: 145 μ V

$V = n \times f / K_{J50}$			
V	Josephson voltage	K_{J50}	Josephson constant
n	programmable integer	f	microwave frequency

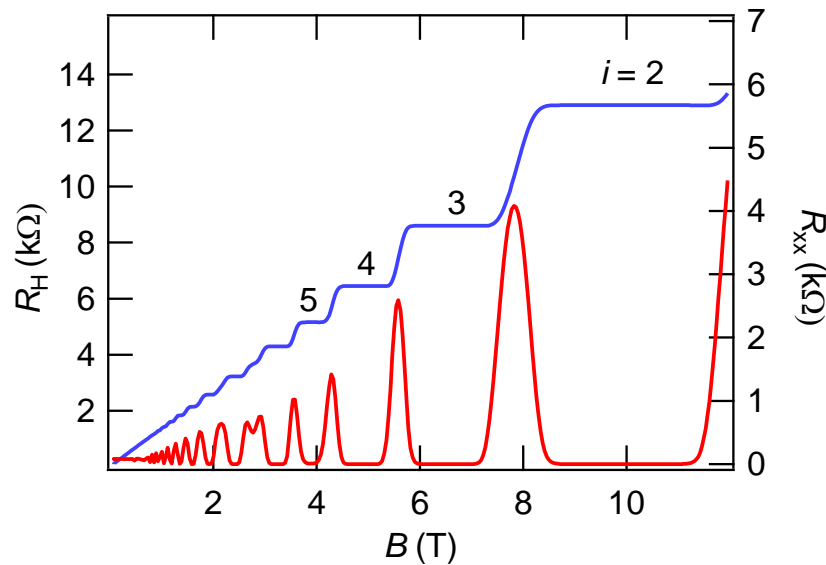


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Electrical quantum standards (2)

K. Von Klitzing discovers the quantum Hall effect in 1980 → Resistance standard



2D electron gas in high magnetic field

$$R_H = \frac{h}{i \cdot e^2} = \frac{R_K}{i}$$

R_K : von Klitzing constant



Practical electrical units

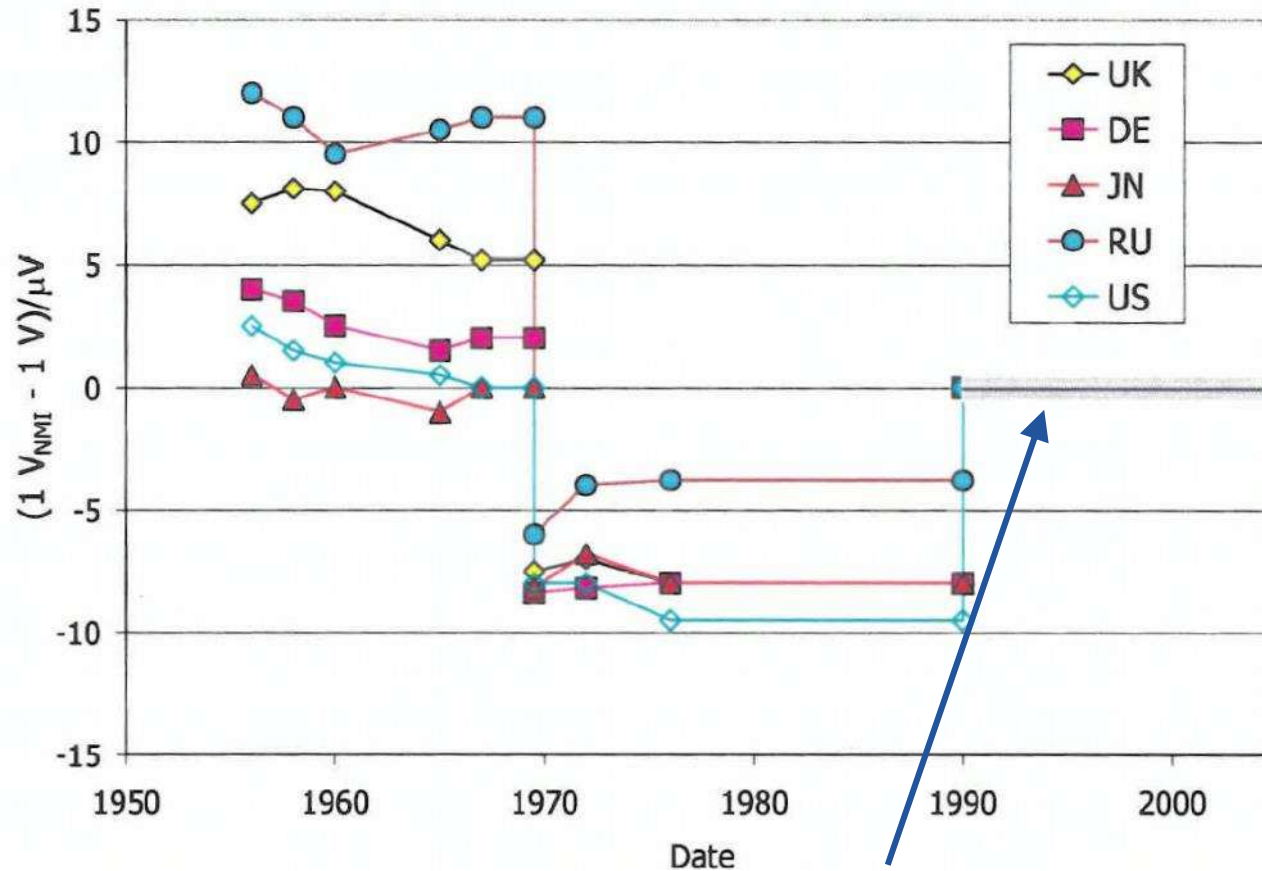
- To make best use of the good reproducibility and worldwide availability of the quantum standards, the conventional values for the Josephson- and von Klitzing constants were introduced as of January 1, 1990.

$$K_{J-90} = 483\,597.9 \text{ GHz/V} \rightarrow \text{rel. uncertainty in the SI : } 0.4 \text{ ppm}$$
$$R_{K-90} = 25\,812.807 \text{ } \Omega \quad 0.1 \text{ ppm}$$

- Worldwide uniformity and improvement of electrical calibrations as a consequence of the conventional units.
- The uncertainty of the constants does only apply if electrical units are linked with mechanical units.

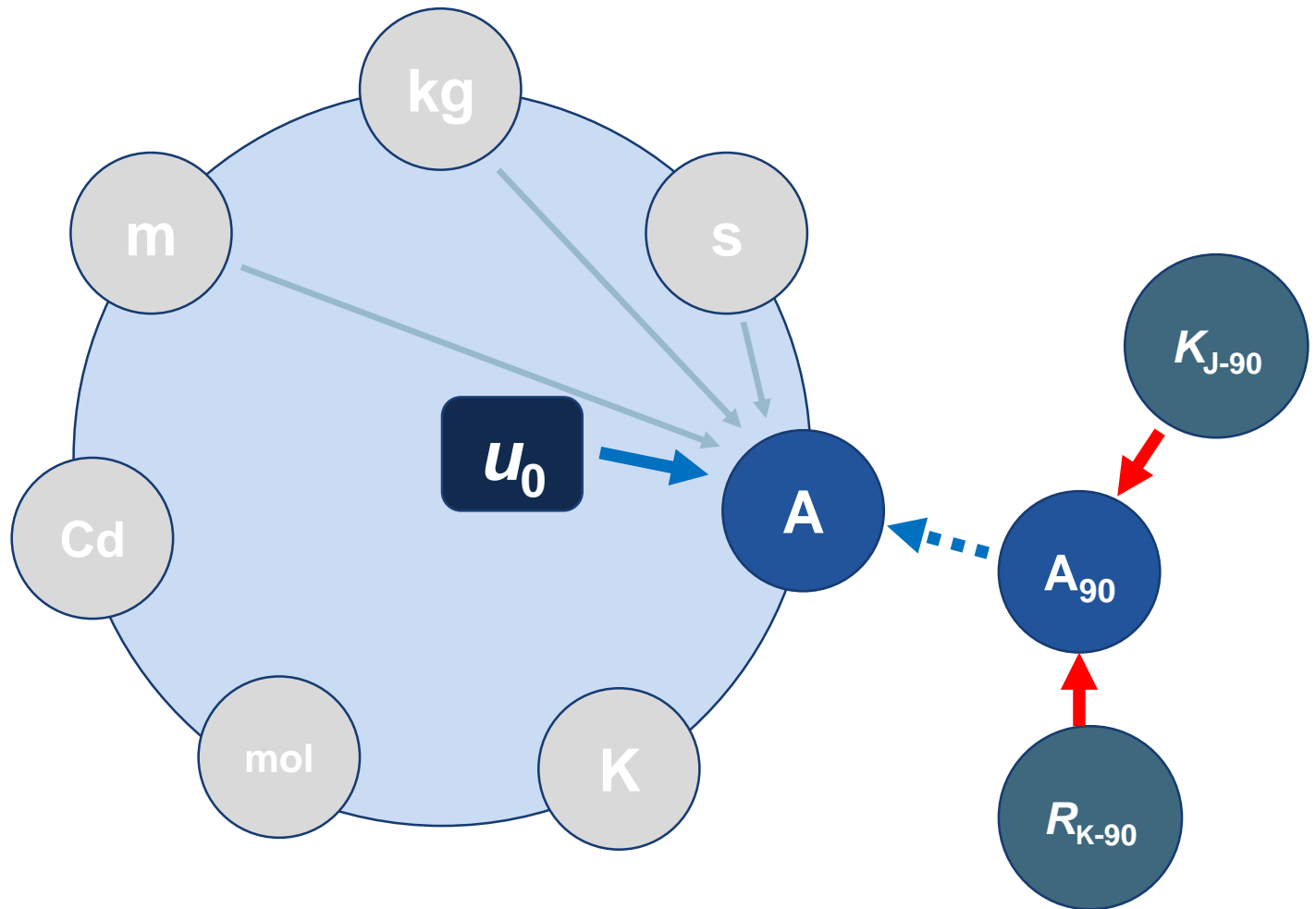
The effect of the 1990 conventional values

Differences in unit realization in different countries

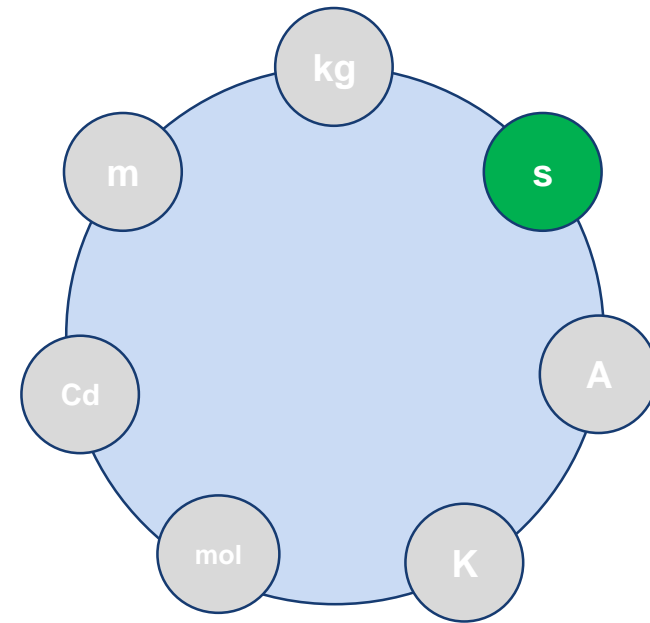
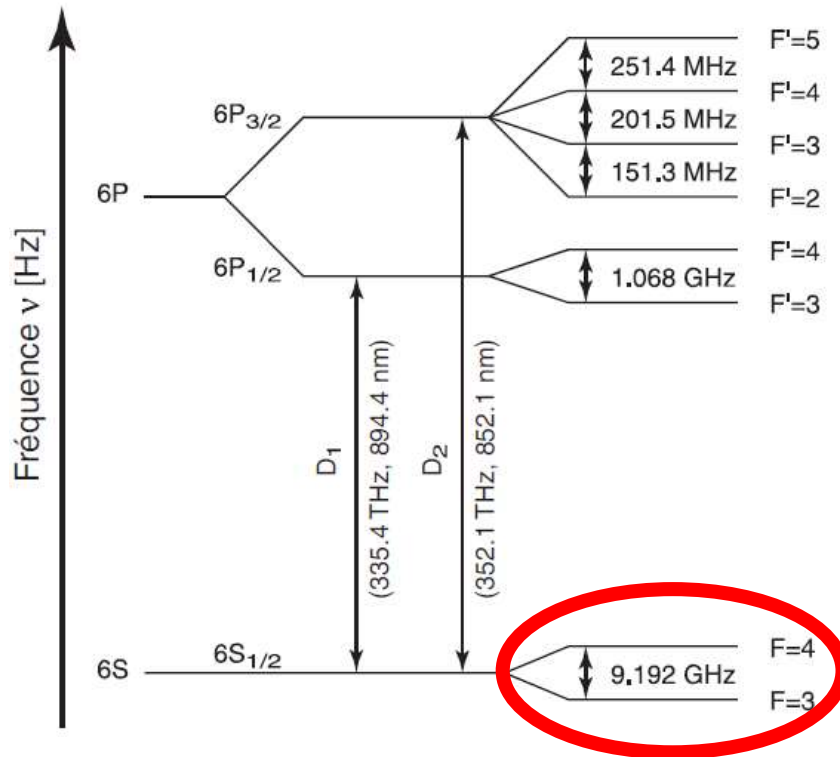


Discovery of a new physical effect allows an improvement by a factor of 100

The practical electrical subsystem

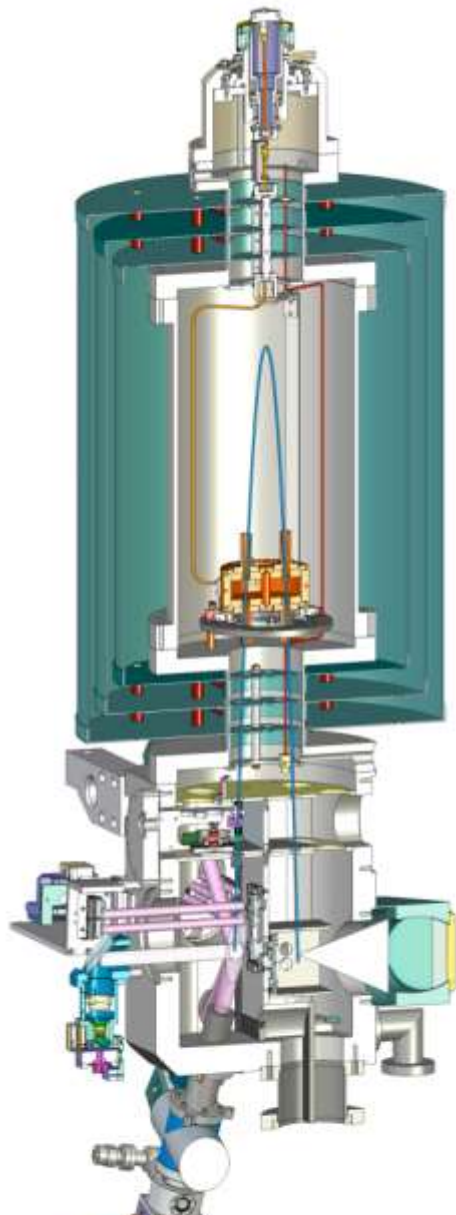


Unit definition: the second



$$1 \text{ s} = \frac{9\,192\,631\,770}{\Delta \nu (^{133}\text{Cs})_{\text{hfs}}}$$

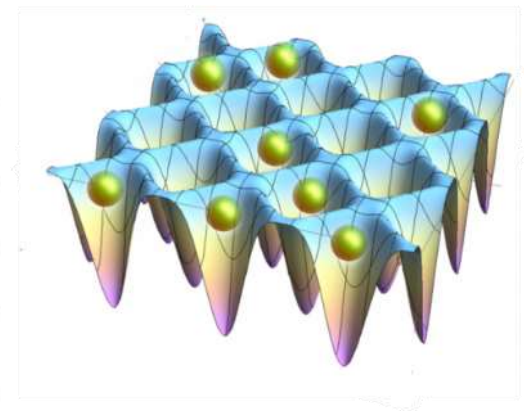
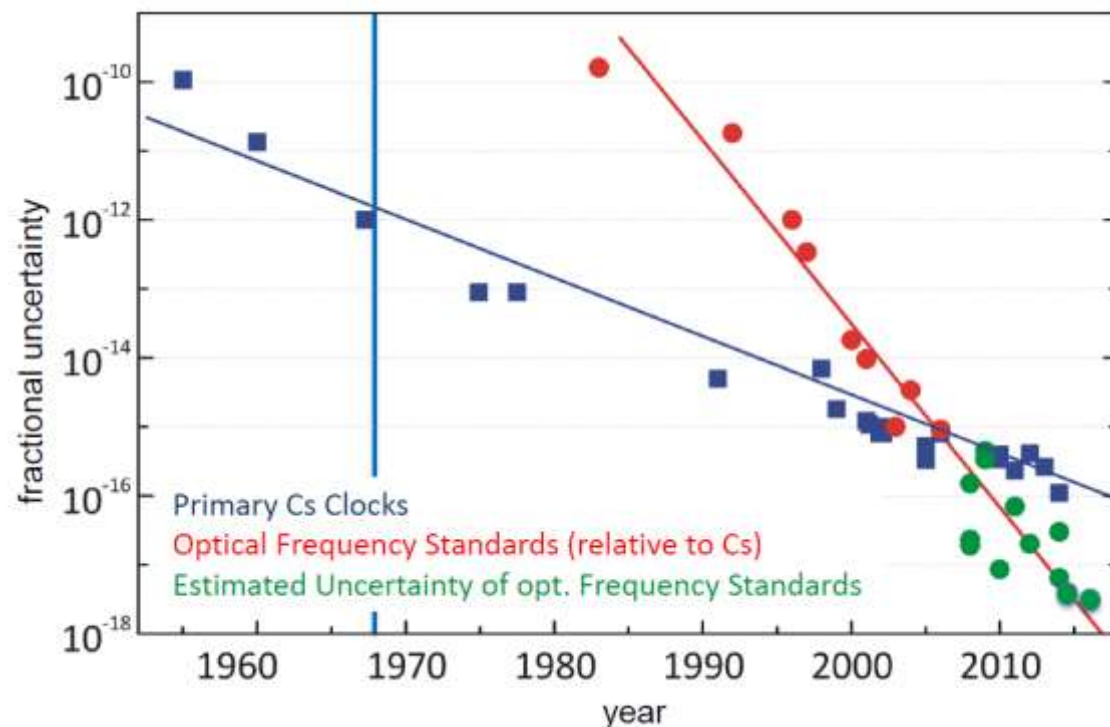
Realisation of the s: Cs fountain



$\sim 2 \times 10^{-16}$

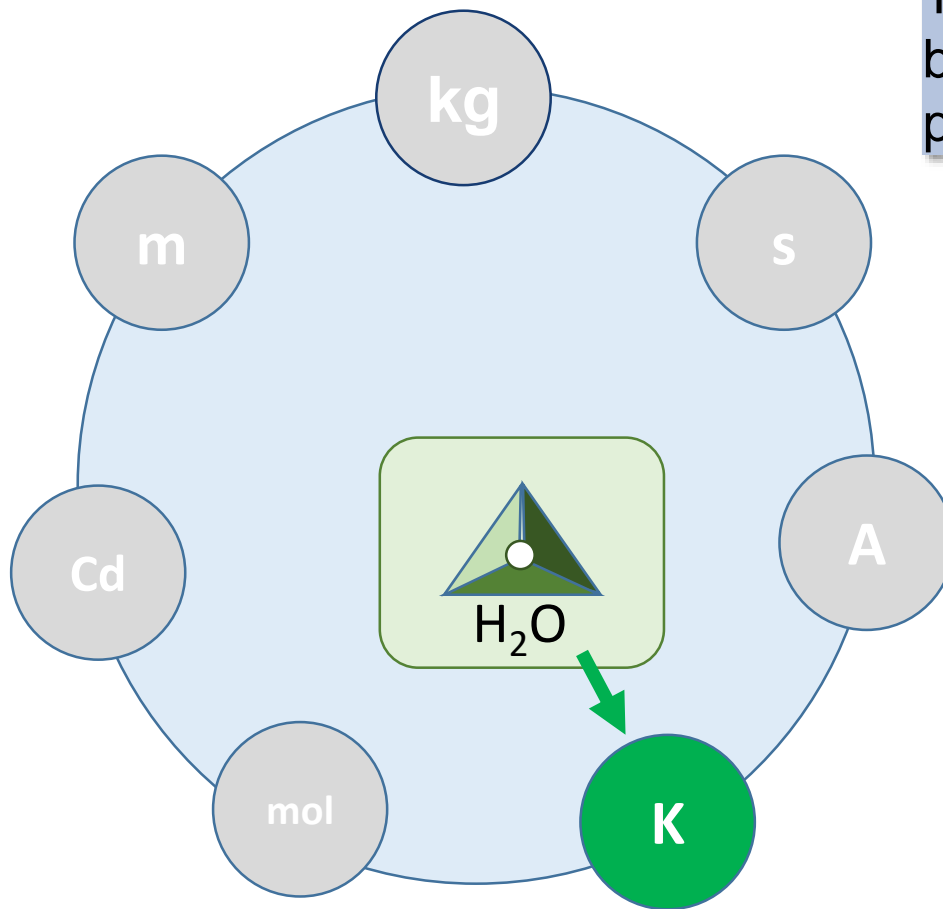
New definition for the second?

- Properties of the selected Cs transition limit the accuracy of the realization
- Next definition: Frequency of an optical transition (GHz \rightarrow THz)



Optical lattice clock, RIKEN

The kelvin



Temperature scale defined by absolute zero and triple point of water

Limitations:

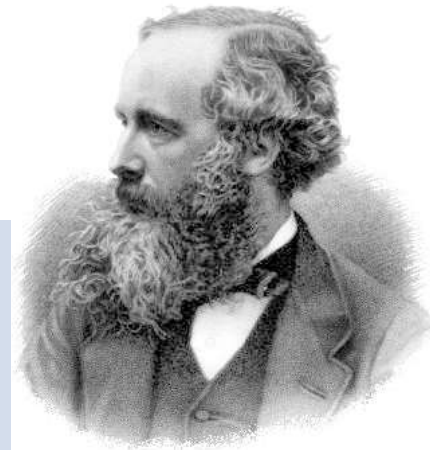
- Impurities in cell
- Isotopic composition of water

Concept of the revised SI

- Determination of the Planck constant
- Boltzmann constant

James Clerk Maxwell (1870)

« If we wish to obtain standards of length, time and mass which shall be absolutely permanent, we must seek them not in the dimensions, or motion or the mass of our planet, but in the wavelength, the period of vibration, and absolute mass of these imperishable and unalterable and perfectly similar molecules. »



Units and fundamental constants

Concept for the revised SI

- Choose seven (fundamental) constants and fix their values.
- presentation of the definition of the entire system using a single statement of the values of the seven chosen constants.
⇒ the constants set the scale for the system.
- With the fixed constants and with the help of the laws of physics, all units in the SI may be realized.
- A distinction between base and derived units is no longer necessary. (for reason of continuity with the past however, the concept of base units will be kept).

Challenge 1: Replacing the kilogram prototype

- Link the unit of mass to fundamental constants
- Experiments needed, that are able to establish this link with $\Delta m/m \leq 10^{-8}$.

- Approach A

“From microscopic to macroscopic world...”



count atoms

- Approach B

“establish link between electrical and mechanical quantities via the force laws and benefit from the electrical quantum standards...”

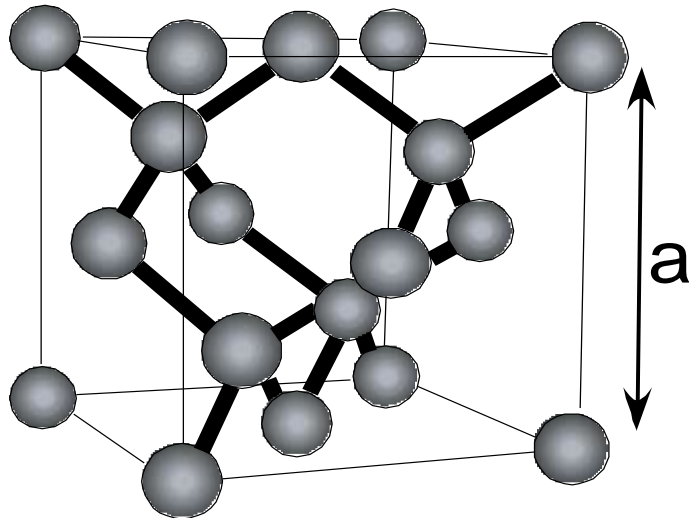


quantum Hall effect , Josephson effect

“Counting” atoms

Step 1: Link between microscopic and macroscopic mass

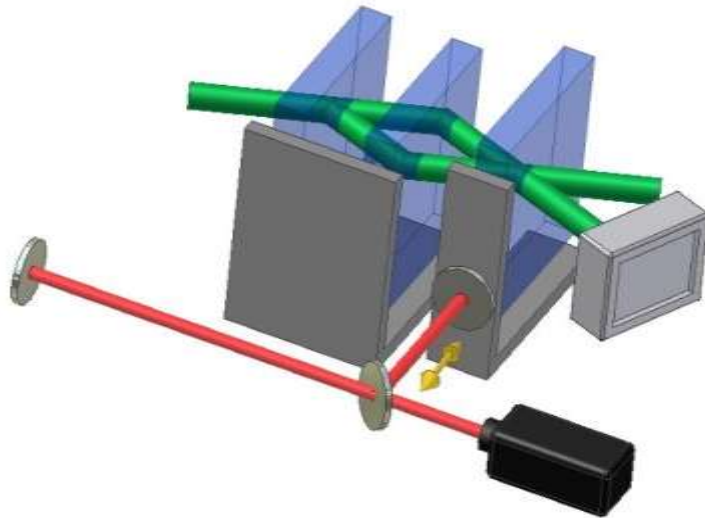
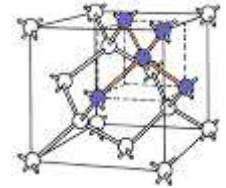
➔ Density measurement in a Si single crystal



$$n_x = 8 \frac{V}{a^3},$$

$$m = n_x \cdot m_{Si}$$

X-ray crystal density experiment (XRDC)

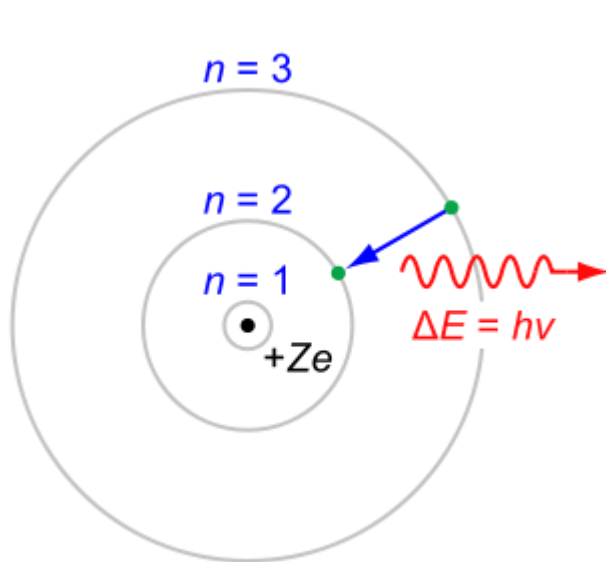


- **$M(\text{Si})$** , molar mass:
isotopic content (^{28}Si , ^{29}Si , ^{30}Si)
measured using mass spectrometry
- lattice constant **a** : X-ray interferometer
- **Volume**: interferometric measurement of the sphere
- **$n=8$** ? : lattice defects, surface effects

Uncertainty using ^{28}Si : 2×10^{-8}

Link to the Planck constant

Bohr Model of the hydrogen atom

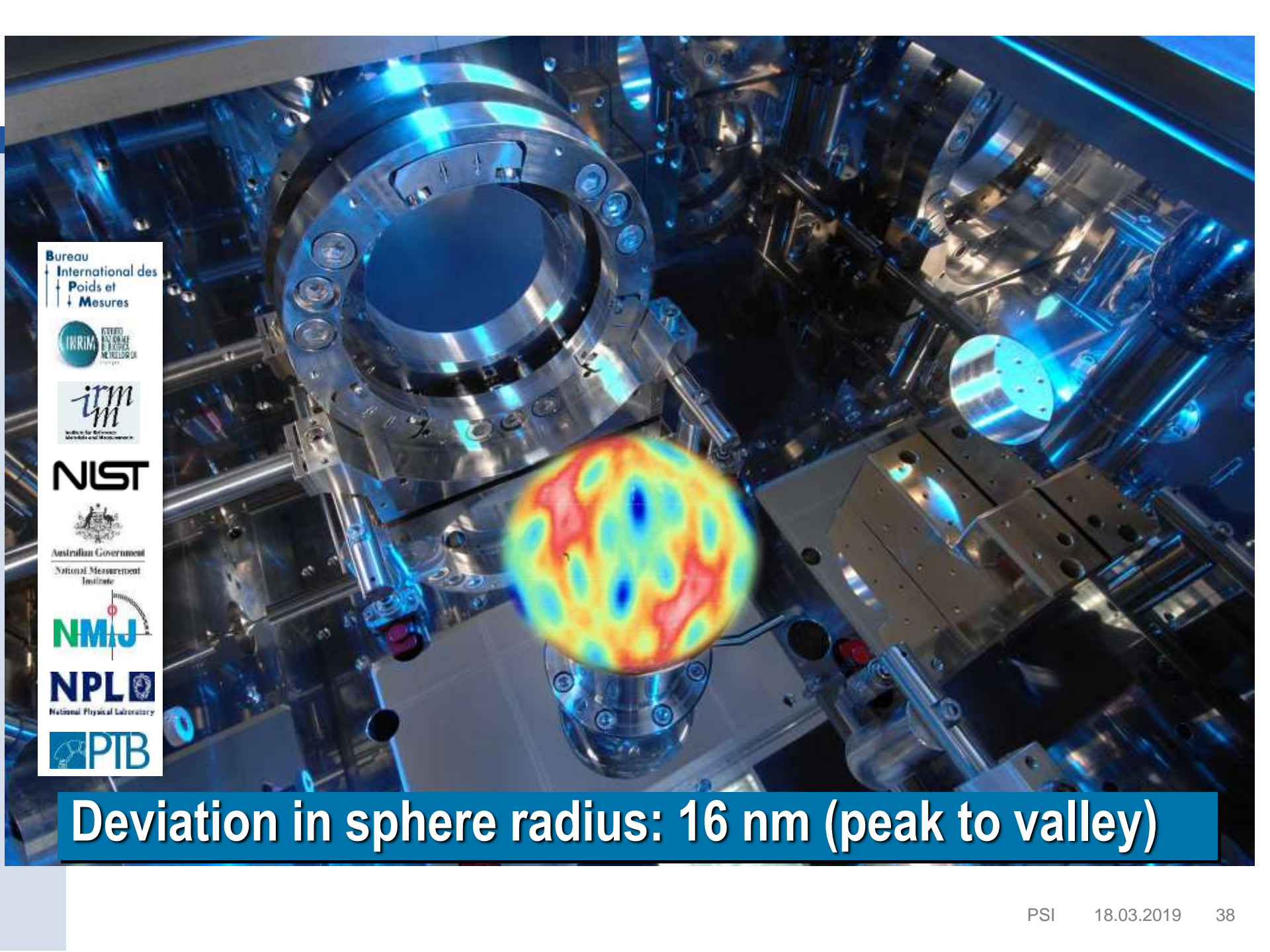


$$\frac{1}{\lambda} = R_{\infty} \left(\frac{1}{n_1^2} - \frac{1}{n_2^2} \right)$$

Rydberg constant

$$R_{\infty} = \frac{m_e \alpha^2 c}{2h}$$

Si sphere $\rightarrow m_e = \left(\frac{m_e}{m_{Si}} \right)_r m_{Si} \rightarrow h = \frac{c \alpha^2}{R_{\infty}} \left(\frac{m_e}{m_{Si}} \right)_r m_{Si}$



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National Measurement
Institute

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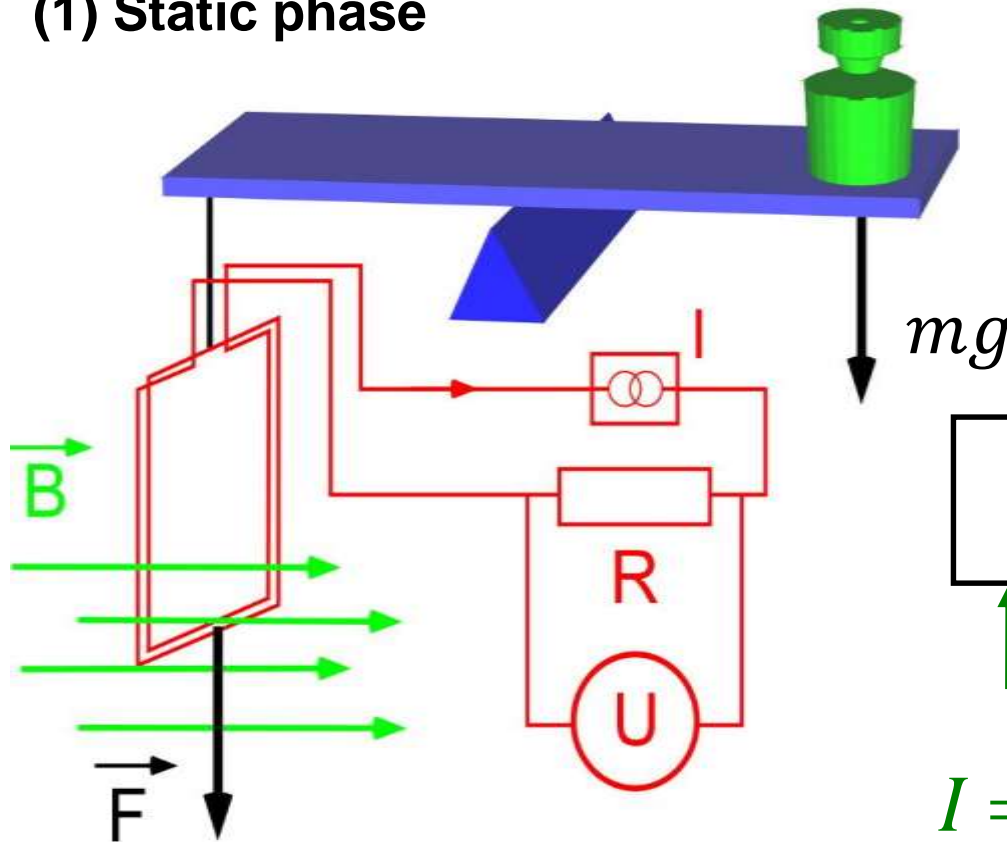
PTB

Deviation in sphere radius: 16 nm (peak to valley)

Electric approach: Watt balance

Two phase experiment (B. Kibble, 1976)

(1) Static phase



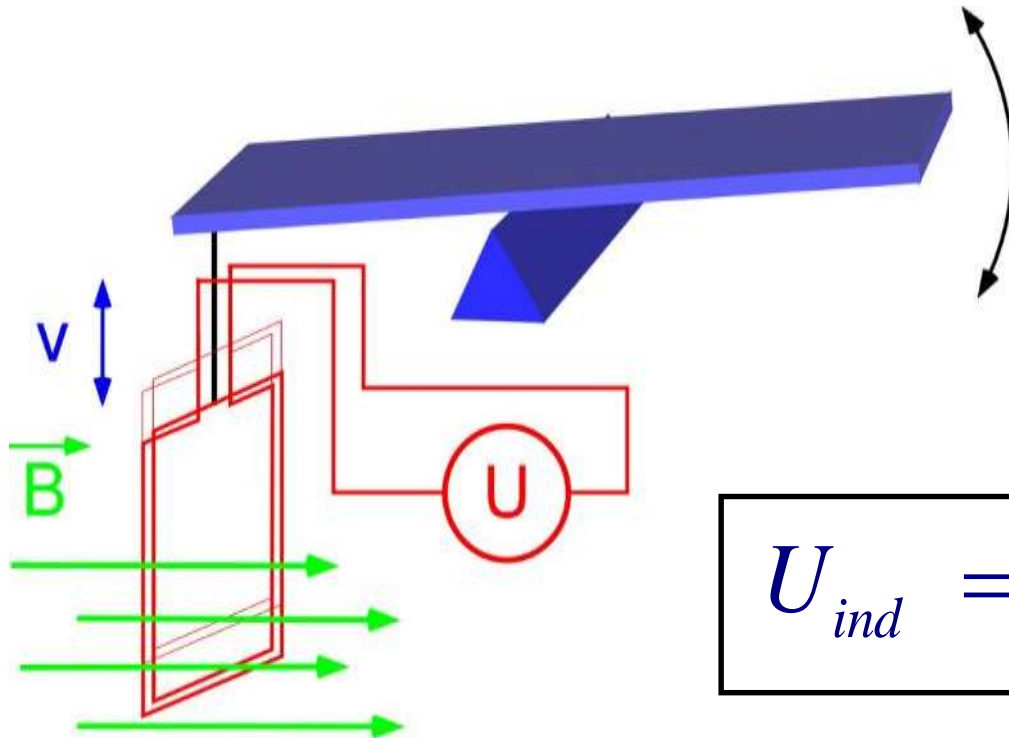
B. Kibble

$$I \cdot \int d\ell \times B = mg$$

$$I = \frac{U_j}{R}$$

(QHE & JVS)

Watt balance (2): Dynamic phase



$$U_{ind} = v \cdot \int d\ell \times B$$

Combination of the phases

$$UI = mgv$$

Electrical
power

Mechanical
power

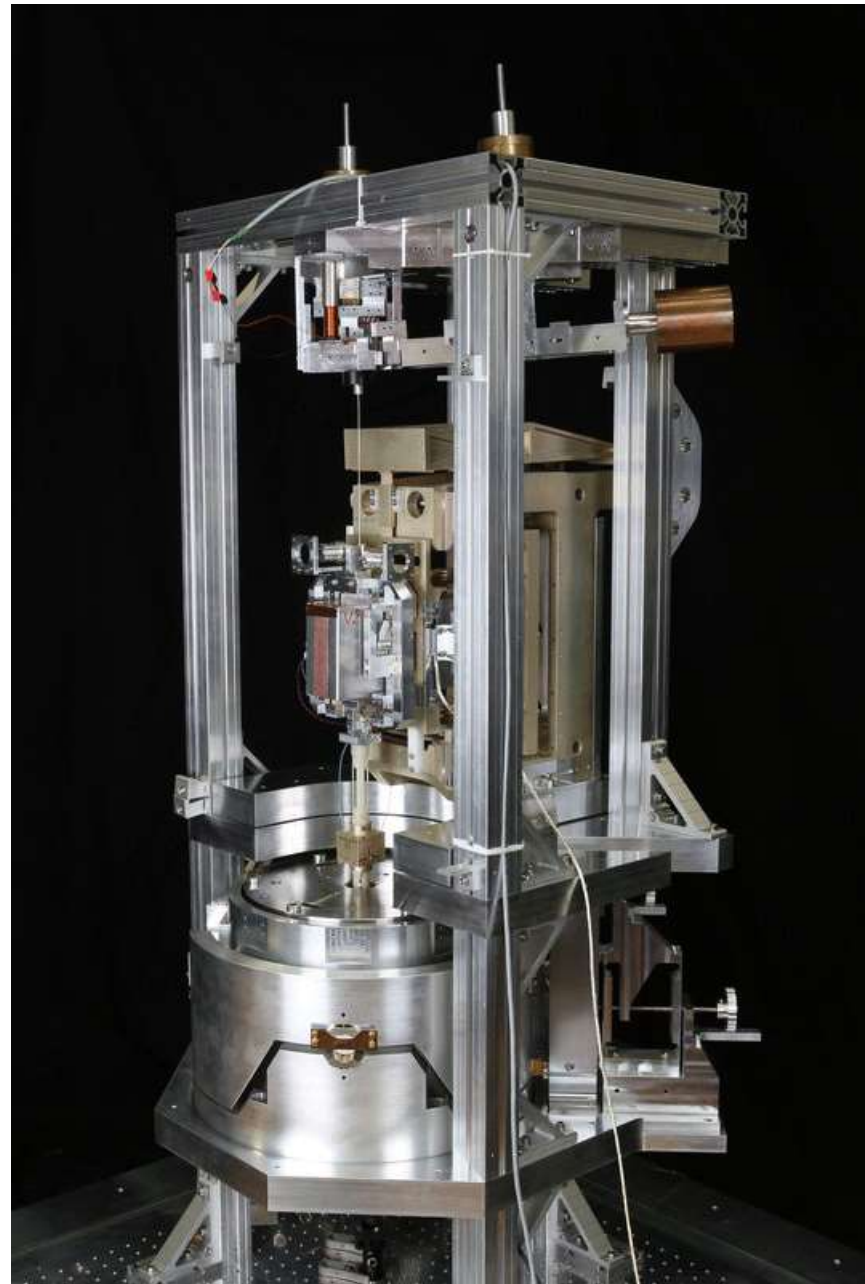
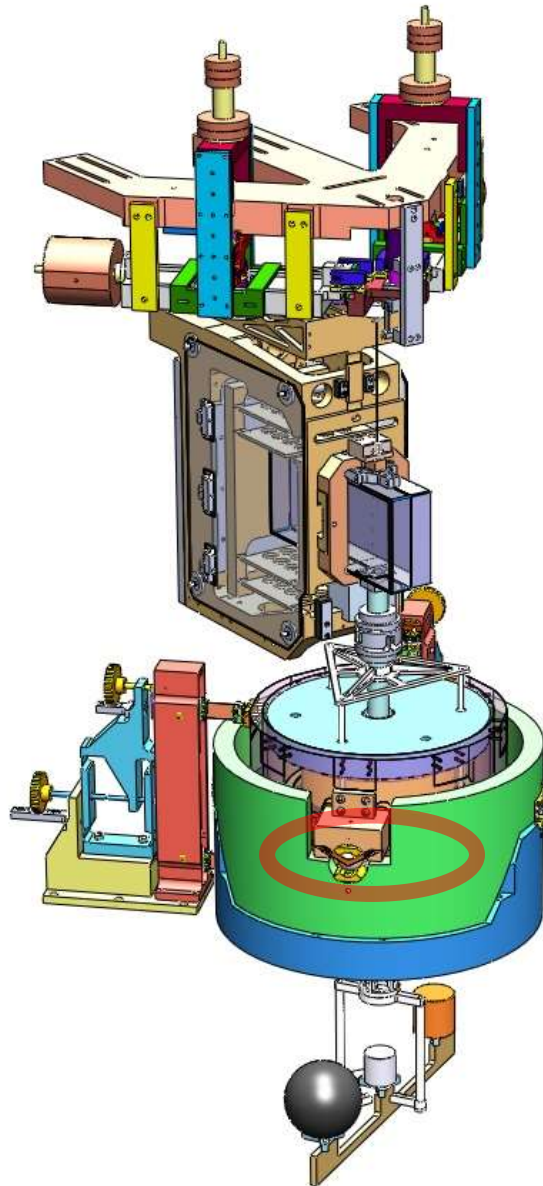


Electrical quantum effects(QHE & JVS)



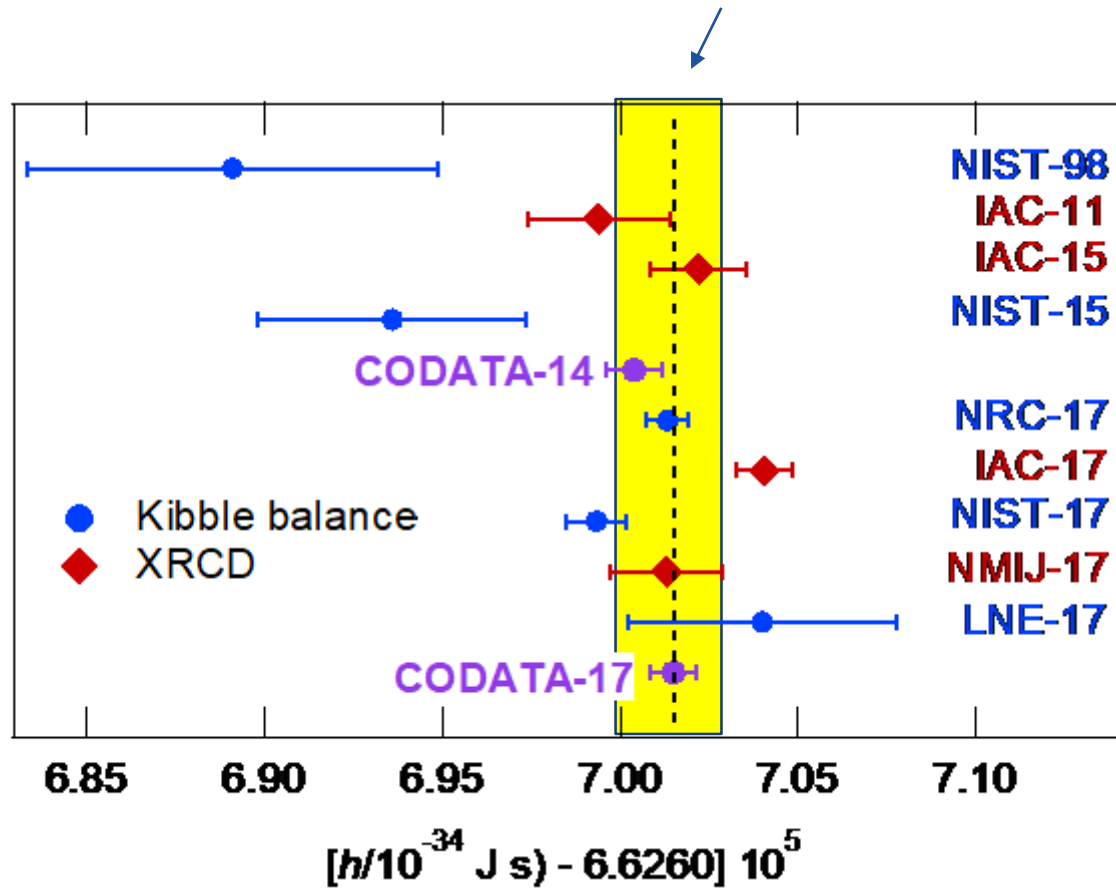
$$m = C \frac{f_{J1} f_{J2}}{gv} h$$

The METAS watt balance

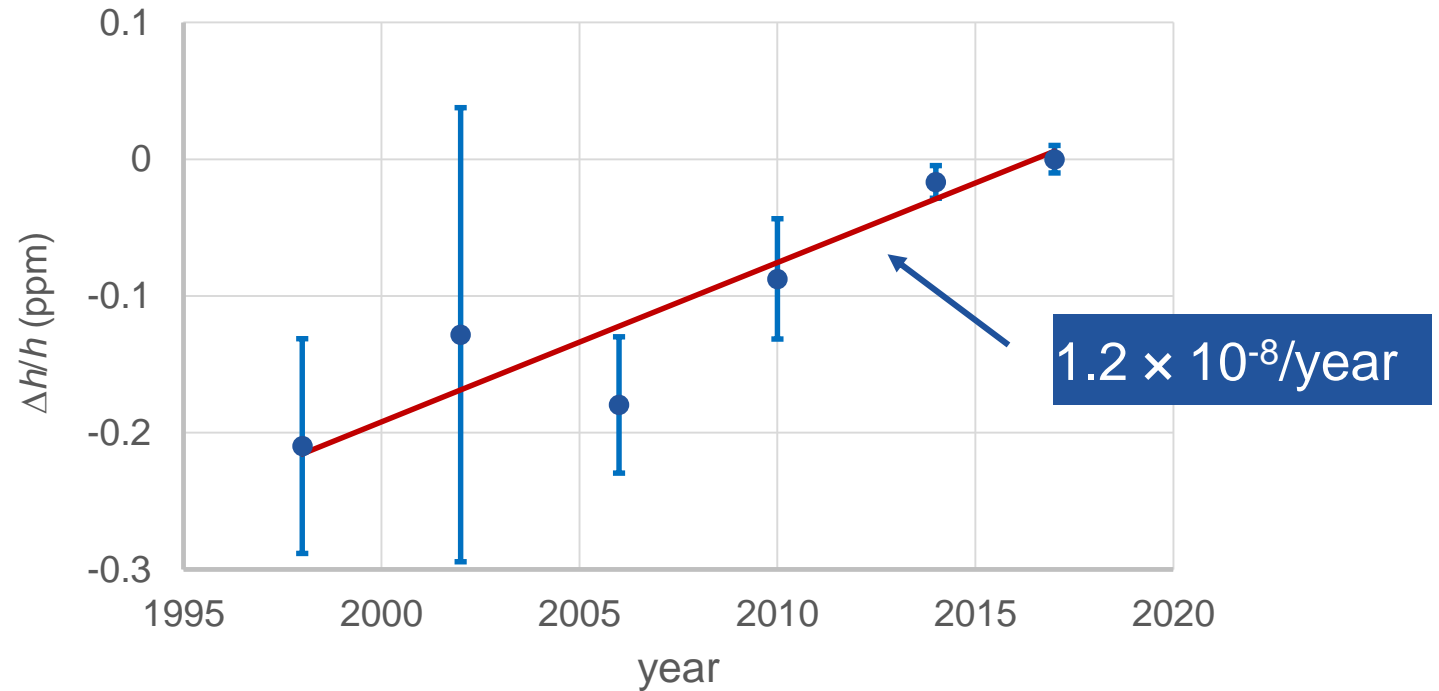


Status h -values

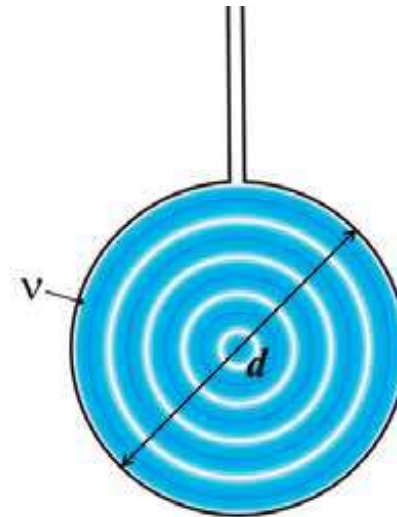
2×10^{-8} uncertainty requirement



CODATA- values for h



Challenge 2: Boltzmann constant



Acoustic gas thermometer

$$E = \frac{3}{2} kT$$

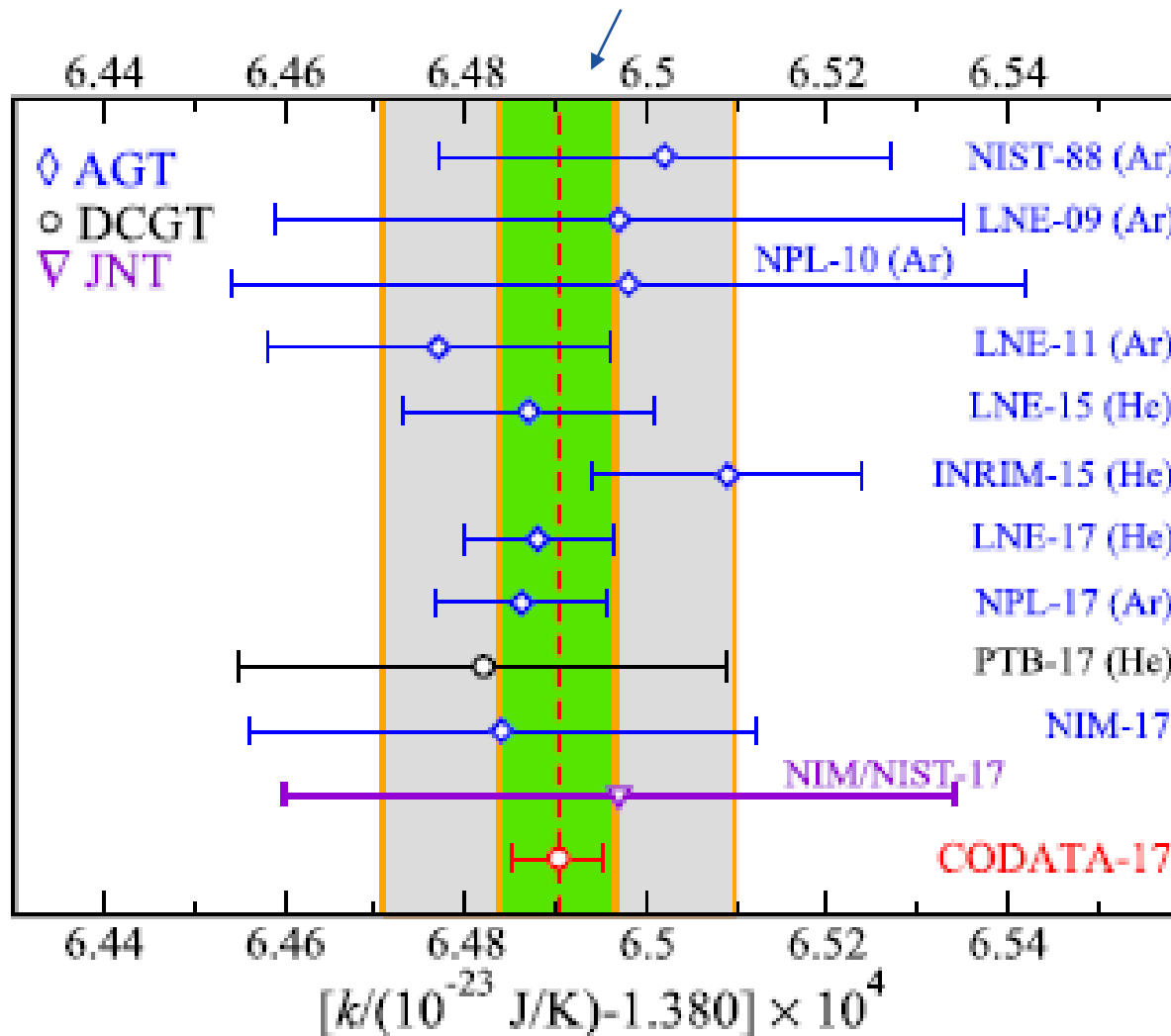
Sound velocity

$$\mu_0 = \sqrt{\frac{\gamma_0 N_A kT}{M}}$$

Specific heat ratio

Experiments Boltzmann constant

0.5×10^{-6} uncertainty requirement



AGT: acoustic gas thermometry;

DCGT: dielectric constant gas thermometry;

JNT: Johnson noise thermometry.

The seven constants

<i>Unit</i>		<i>SI old</i>
second	s	$\Delta \nu(^{133}\text{Cs})_{\text{hfs}}$
meter	m	c
kilogram,	kg	$m(K)$
ampere,	A	μ_0
kelvin,	K	T_{TPW}
mole,	mol	$M(^{12}\text{C})$
candela,	cd	K_{cd}

CODATA adjustment 2017

Metrologia 55 (2018) L13–L16

<https://doi.org/10.1088/1681-7575/aa950a>

Short Communication

The CODATA 2017 values of h , e , k , and N_A for the revision of the SI

D B Newell¹, F Cabiati, J Fischer, K Fujii, S G Karshenboim,
H S Margolis[✉], E de Mirandés, P J Mohr, F Nez, K Pachucki, T J Quinn,
B N Taylor, M Wang, B M Wood and Z Zhang

Committee on Data for Science and Technology (CODATA) Task Group on Fundamental Constants

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CrossMark

Abstract

Sufficient progress towards redefining the International System of Units (SI) in terms of exact values of fundamental constants has been achieved. Exact values of the Planck constant h , elementary charge e , Boltzmann constant k , and Avogadro constant N_A from the CODATA 2017 Special Adjustment of the Fundamental Constants are presented here. These values are recommended to the 26th General Conference on Weights and Measures to form the foundation of the revised SI.

Keywords: international system of units, fundamental constants, SI redefinition

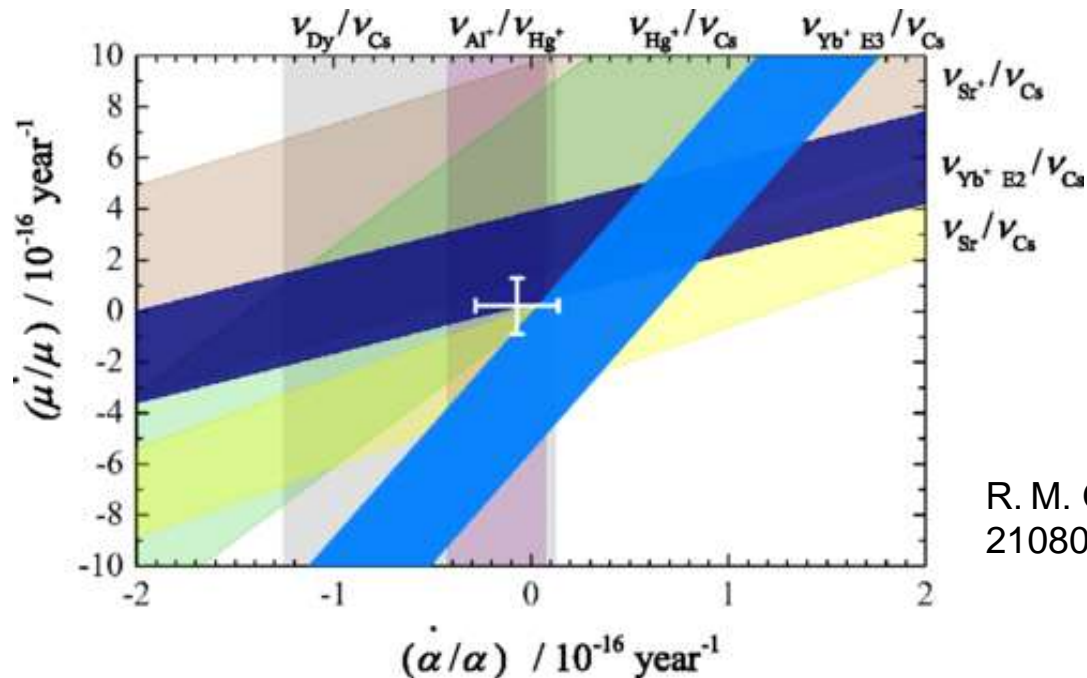
Definition of the revised SI

The International System of Units, the SI, is the system of units in which

- the ground state hyperfine splitting frequency of the caesium 133 atom $\Delta\nu(^{133}\text{Cs})_{\text{hfs}}$ is exactly **9 192 631 770 hertz**,
- the **speed of light in vacuum c** is exactly **299 792 458 metre per second**,
- the **Planck constant h** is exactly **$6.626\ 070\ 15 \times 10^{-34}$ joule second**,
- the **elementary charge e** is exactly **$1.602\ 176\ 634 \times 10^{-19}$ coulomb**,
- the **Boltzmann constant k** is exactly **$1.380\ 649 \times 10^{-23}$ joule per kelvin**,
- the **Avogadro constant N_A** is exactly **$6.022\ 140\ 76 \times 10^{23}$ reciprocal mole**,
- the **luminous efficacy K_{cd}** of monochromatic radiation of frequency **540×10^{12} Hz** is exactly **683 lumen per watt**.

Are constants constant?

Frequencies of different clocks depend differently on fundamental constants



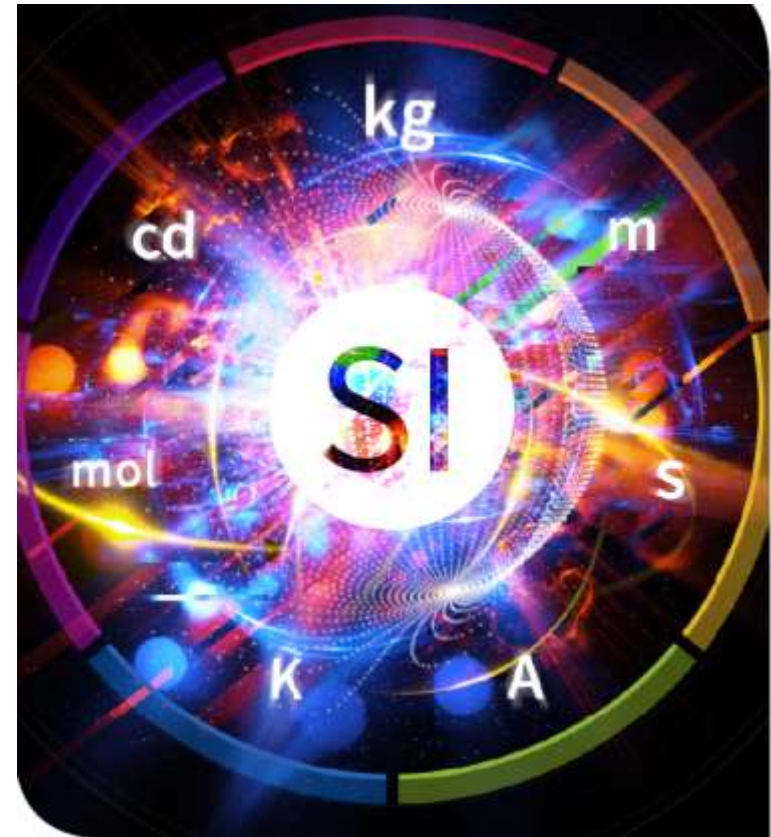
R. M. Godun et al., Phys. Rev. Lett. 113, 210801 – 2014

Constraints placed on time variation of the **fine structure constant α** and the **proton-to-electron mass ratio μ** , from frequency measurements in $^{171}\text{Yb}^+$, $^{88}\text{Sr}^+$, ^{87}Sr , $^{199}\text{Hg}^+$, Dy and $^{27}\text{Al}^+$.

Timeline for the revised SI

- 16. Nov. 2018: Decision at the 26. General Conference for Weights and Measures

- **20. May 2019:**
World metrology day;
Practical implementation of the revised SI



Implementation of the revised SI

- No changes for the units **second, metre and candela**
- The **kg** has to be realized through experiments like the watt balance and the XRCD; The kilogram prototype will have an uncertainty.
- The practical electrical subsystem becomes obsolete; the quantum effects realize the units directly in the SI. As a consequence small step-changes in results will occur.

Voltage: 1.07×10^{-7} (rel. change)

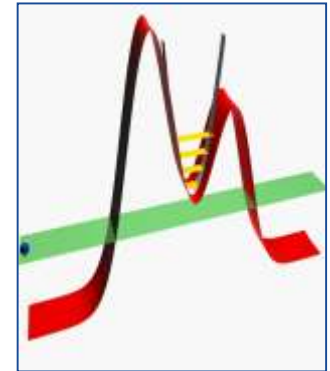
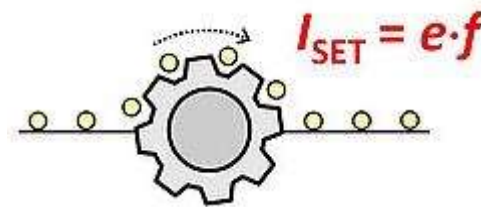
Resistance: 1.78×10^{-8}

Electrical units in the revised SI

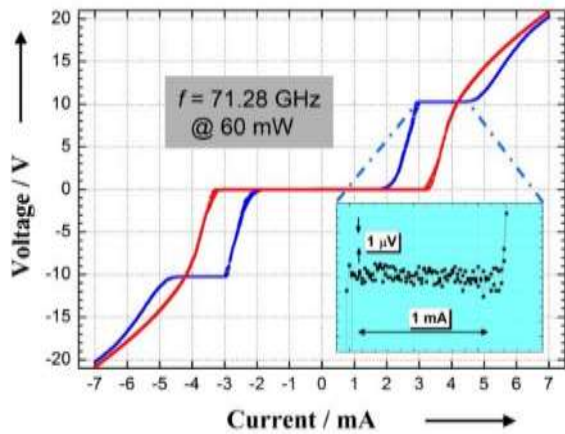
h and *e* fixed:

Electrical quantum effects realise the electrical units directly.

ampere

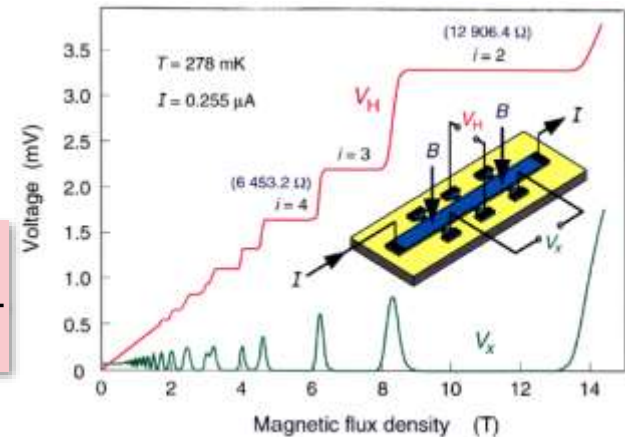


volt



$$U = \frac{h}{2e} i \cdot f_J$$

ohm



$$R_H = \frac{h}{i \cdot e^2}$$

Implementation (2)

- **Kelvin:** The experiments used for the Boltzmann constant measure now temperature directly. The scale will improve. The temp. of the triple point will have an uncertainty.
- **Mole:** No link to the kilogram anymore; the molar mass of ^{12}C has an uncertainty.

Conclusion



1. The planned revision changes and improves the SI fundamentally.
2. A set of "defining constants" sets-up the system.
3. **The system is independent of specific experiments** - improved insights and technologies make improved unit realizations possible.
4. The realization is possible everywhere.
5. According to current knowledge, the system is stable over time.





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Thank you very much for your attention