

FP Release Modeling

Presented by Larry L. Humphries <u>llhumph@sandia.gov</u> SAND2018-4214 C



MELCOR Radionuclides and Decay Heat Release Models

- Four basic release models, with options
 - -CORSOR, fractional release rate = A exp(B T)
 - -CORSOR-M, fractional release rate = $k_0 \exp(-Q/RT)$
 - Extended in MELCOR 1.8.5 from the original form to include release of classes 7 (Mo), 9 (La), and 11 (Cd)
 - -CORSOR-Booth, based on Cesium diffusion
 - $D_0 \exp(-Q'/RT)$
 - Options for high- or low-burn-up fuel
 - -Modified CORSOR-Booth
 - Refit of coefficients based on VERCORS and FPT
- Generalized Release Model
- Triso Fuel Release Model
- Structural Release of Ag-In-Cd control rod material



CORSOR Booth – Diffusion Implementation

- The Booth scaling factors were derived to fit diffusion coefficients to be used for independent diffusion calculations for each class, not as multipliers to the CS release rates as it has been applied in the past. Consequently, a new Booth model has been developed
- New ORNL-Booth Model (to be released)
 - Proposed implementation (ICRLSE=7,-7)
 - Diffusion coefficients scaled to Cs diffusion coefficients and independent diffusion calculation for each class
 - Consistent with how diffusion coefficients were fit to data

$$= \int_0^{D't} \frac{f'_k}{1 - f_k} \cdot (1 - f_k)$$

- Existing Modified ORNL-Booth Model
 - Modified Booth Implementation(ICRLSE=5,-5)
 - Scale RN releases rates to Cs Booth diffusion release rate

$$= \int_0^{D't} S_k \frac{f_{Cs}'}{1 - f_{Cs}} \cdot (1 - f_k)$$

Corrected Booth Modeling Using Scaled Diffusion Coefficients

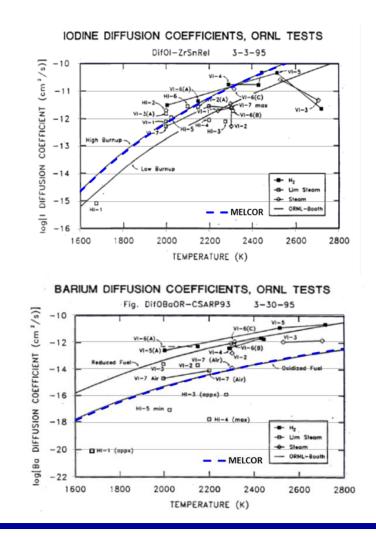
• MELCOR diffusion coefficients scaled with ORNL Data

$$- D_{class} = S_{class} D_0 e^{-\frac{Q}{RT}}$$

• Perform diffusion calculation for each class using appropriate diffusion coefficient

- ICRLSE=7,-7

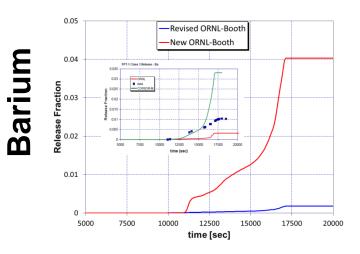
Diffusion Coeff, D ₀	1x10 ⁻⁶ m ² /sec	
Activation Energy, Q	3.814x10 ⁵ Joule/mole	
S _{XE}	1	
S _{Cs}	1	
S _{Ba}	4x10 ⁻⁴	
SI	0.64	
S _{Te}	0.64	
S _{Ru}	0.0025	
S _{Mo}	0.2	
S _{Ce}	4x10 ⁻⁸	
S _{La}	4x10 ⁻⁸	
S _{Cd}	.25	
S _{Sn}	.16	

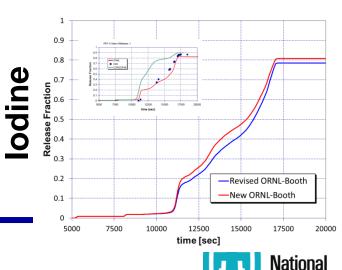




Validation of corrected Booth-Diffusion Model

- Validation of model against FPT-1 experiment
 - 'New ORNL-Booth' (ICRLSE=-7,7) represents modeling based on scaling diffusion coefficients
 - Release may be significantly changed for those RNs with diffusion scale factors much less than unity
 - Comparison of Barium release
 - Diffusion coefficient is orders of magnitude smaller than Cs (scale factor = $4x10^{-4}$)
 - Predicts larger release fraction than observed
 - » May need to be adjusted
 - Predicts much larger release fraction than 'Revised ORNL-Booth' (ICRLSE=-5,5) though relative error is about the same
 - Comparison of I₂ Release
 - Very Slight differences
 - Diffusion coefficient scale factor close to unity (0.65)





Laboratories

5

CORSOR-Booth

Gaseous mass transport

$$\dot{m}_{k} = \left[Nu \frac{D_{k,gas}}{D_{fuel}} \right] \cdot A_{fuel} \cdot \left(\frac{P_{k,eq}}{RT} - \rho_{k,free} \right)$$

• CORSOR-Booth Model (ICRLSE=3,-3)

Gas-phase mass transfer from heat transfer analogy (free stream density of all RNs assumed zero)

– Mass transfer term removed from effective release rate before scaling other RNs

$$ReleaseRate_{Cs} = \frac{\left[f(\sum_{t+\Delta t} D'\Delta t) - f(\sum_{t} D'\Delta t)\right]}{(1 - RFRAC)\Delta t} V\rho$$
$$DIFF_{Cs} = \left[\frac{1}{ReleaseRate_{Cs}} - \frac{1}{\dot{m}_{Cs}}\right]^{-1}$$
$$DIFF_{k} = DIFF_{CS} \cdot S_{k}$$

- Revised CORSOR-Booth Model (ICRLSE=5,-5)
 - Mass transfer term not removed from Cs release rate
 - Assumes ReleaseRate only accounts for diffusion

$DIFF_k = ReleaseRate_{CS} \cdot S_k$

• Mass transfer term added back to each RN

$$\dot{m}_k = [1/DIFF_k + 1/\dot{m}_k]^{-1}$$



Gaseous Transport of RNs relative to Moles of UO_2 in Volume

- Note that RM indicates Cs release rate is fractional release multiplied by ρV
 - Where V is volume and ρ is UO₂ molar density
 - New model uses actual class inventory
- Note however that release rate is divided by pV before multiplying by RN inventory to obtain mass released
- Even so, scaling by ρV is important in comparing diffusion release rate with gas release rate

Old Modeling

$$ReleaseRate_{Cs} = \frac{\left[f(\sum_{t+\Delta t} D'\Delta t) - f(\sum_{t} D'\Delta t)\right]}{(1 - RFRAC_{Cs})\Delta t}\rho V$$
$$\dot{f}_{k} \text{ (fraction/s)} = \frac{\dot{m}_{tot,k}}{\rho V} \left[F - \frac{P_{k,bulk}}{P_{k,eq}}\right]$$

$$\dot{m}_k = [1/DIFF_k + 1/\dot{m}_k]^{-1}$$

New Modeling

$$ReleaseRate_{Cs} = \frac{[f(\sum_{t+\Delta t} D'\Delta t) - f(\sum_{t} D'\Delta t)]}{(1 - RFRAC_k)\Delta t} Mass_{k,component}$$

$$\dot{f}_{k}(fraction/s) = \frac{\dot{m}_{tot,k}}{Mass_{k}} \left[F - \frac{P_{k,bulk}}{P_{k,eq}} \right]$$
$$\dot{m}_{k} = \left[1/DIFF_{k} + 1/\dot{m}_{k} \right]^{-1}$$



Modification for Te Release

- The presence of unoxidized zirconium can lead to a reduction in release rate of the tellurium class.
 - -Te reacts with Zr forming products with low vapor pressure
 - -The release rate of Te is reduced by a release rate multiplier (with a default value of 1/40 = 0.025) until the mass of unoxidized intact metal cladding falls below a cut-off fraction (default value of 0.7) of the total mass of intact cladding (including the oxide mass).
- The parameters are sensitivity coefficients
 - -Array 7105 for CORSOR and CORSOR-M
 - -Array 7107 for CORSOR-Booth

T. Nakamura and R. A. Lorenz, "Effective Diffusion Coefficients Calculated from ORNL FP Release Test Results," Oak Ridge National Laboratory Research Paper (April 1989).



CORSOR-Booth Strictly Valid in Fuel

- CORSOR-Booth release rate is dependent on the release history (diffusion dependence) since concentration differentials vary over time.
 - Release fraction for fuel is well characterized and does not involve any material that has moved
 - All other components associated with 'transported' fuel material as a result of melting (conglomerate) or loss of structural support PD
 - MELCOR 'blends' release fractions after transport but the results may not be meaningful
 - Consider blending material which has had no release with material that has almost complete release. Averaging over mass makes little sense.
 - Would require tracking distributions of release fractions for each component
- Alternative would be to allow CORSOR-Booth in fuel material but other release models for other components
 - -CORSOR-M and CORSOR release rates are strictly temperature dependent.
 - Include resistance from gas-phase mass transport.



Additional Burst Release Modes

- Diffusion dominates while fuel matrix remains intact, however:
 - -UO2/ZrO2 Interaction
 - Can now be modeled in eutectic model
 - -Liquefaction of UO2 from molten Zircalloiy
 - Can now be modeled in eutectic model
 - -Melting of UO2
- Can lead to burst of fission product release
- Currently not captured in CORSOR/BOOTH models
- Generalized fission product release model



Generalized Fission Product Release Model

• A cumulative burst fission product release fraction is described by the following equation:

$$FB_{j,i} = a_burst_j (c_0 + c_1 * T_i + c_2 * T_i^2 + c_3 * T_i^3)$$

Where

T_i is the fuel temperature that existed during the time interval Dt_i

 c_0, c_1, c_2, c_3 are constant coefficients provided in user input

a_j is a constant class dependent coefficient provided in user input.

• A cumulative diffusive fission product release fraction is described by the following equation:

$$FD_{j,i} = b_{diff_{j}} (FD_{j,i-1} + (1 - FB_{j,i-1} - FD_{j,i-1}) \cdot [1 - e^{-kd_{j,i} \cdot \Delta t_{i}}])$$

Where

FD_{j,i} is the cumulative fraction of diffusive fission product released up to time t_i B_diff_j is a constant class dependent coefficient provided in user input FD_{j,i-1} is the cumulative fraction of diffusive fission product released up to time t_{i-1} FB_i is the cumulative fraction of burst fission product released up to time t_i $[1 - e^{-kd_{j,i}\cdot\Delta t_i}]$ is the fractional release due to diffusion during the time interval Dt_i $kd_{j,i}$ is the release rate coefficient for fission product class j calculated using the temperature, Ti, that existed during the time interval Dt_i

$$kd_{j,i} = A_j e^{-B_j/(RT_i)}$$

Where A_j and B_j are class dependent coefficients provided in user input.



Generalized Fission Product Release Model

• The total cumulative fission product release fraction at time t_i for fission product j is determined by:

$$F_{j,i} = d_total_j \cdot (FB_{j,i} + FD_{j,i})$$

• The cumulative release fraction cannot exceed the amount of fission product available

 $FB_{j,i} = FB_{j,i-1}$ and $FD_{j,i} = FD_{j,i-1}$ when $FD_{j,i} \ge 1.0$

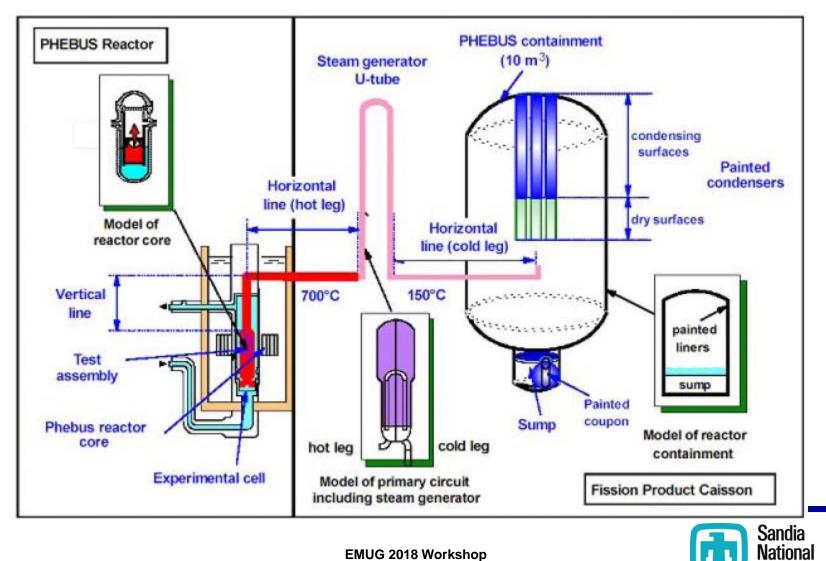
• The derivative of the cumulative burst release with respect to time cannot be less than zero; if the temperature decreases, the cumulative burst release remains constant.

 $FB_{j,i} = FB_{j,i-1}$ when $T_{i-} \ge T_{B-max}$ or $T_{melting}$

• The cumulative burst release reaches its maximum when the fuel temperature reaches T_{B-max} or $T_{melting}$ whichever is lower $FB_{j,i} = FB_{j,i}$ when $T_i \ge T_{B-max}$ or $T_{melting}$

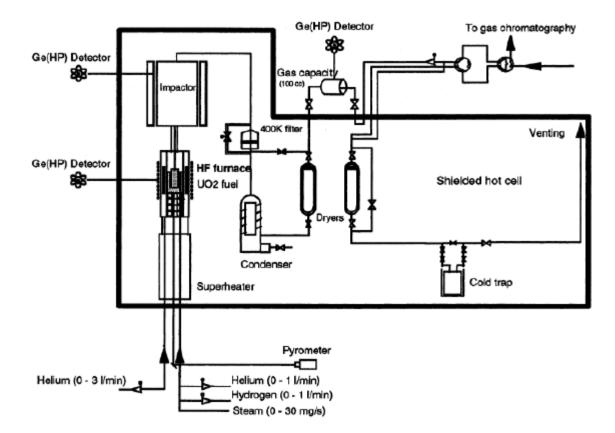


PHEBUS Facility





VERCORS Facility

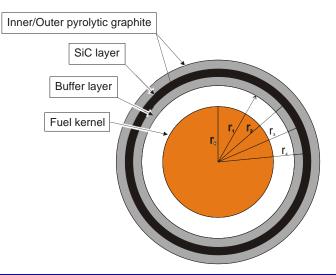


G. Ducros et al. / Nuclear Engineering and Design 208 (2001) 191-203



HTGR Fission Product Release

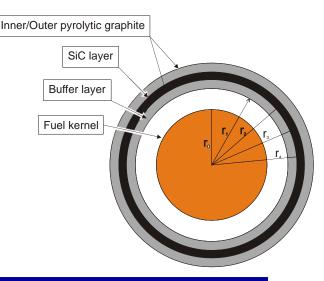
- HTGR release includes phenomena not in LWRs
 - -Failure and release is spread out over time
 - Low level release during operation
 - Circulating activity
 - Accident release time is much longer: 50-100 hrs
 - Localized release vs gap release
 - –Dust present in primary system
- Phenomena considered
 - -TRISO particle failure
 - -Release
 - -Dust generation and transport
 - -U contamination of matrix





MELCOR Approach for FP Model

- Input required from other codes
 - -FP inventory ORIGEN
 - -Power shape, axial and radial profiles
 - -Reactivity feedback parameters for point kinetics
- For HTGR, require some other initial input
 - -Some from experimental data, ie
 - Initial particle failure fraction
 - Dust generation rate
 - –Distribution of dust and FP
 - In particle, kernel vs buffer
 - In primary system, dust distribution





MELCOR Initial Operation Phase

- Do "accelerated" normal operation run with MELCOR to get FP and dust distribution
 - -This problem would be run long enough to establish trends and/or equilibrium for burnup cycle (~3 yrs)
 - -Some FP release during operation
 - -Dust generated at given rate in core
 - -Transported and deposited using MELCOR CVH/RN packages
- Scale to desired operating time (~10 yrs)
- •Use as starting conditions for accident scenario



MELCOR Transient Phase

- Use steady state results and provided inventory, neutronic results, etc. at start of accident scenario
- Transient calculation using
 - -FP release model
 - -Transport of FP and resuspended dust via MELCOR CVH/RN packages
 - -User input Failure Curve
 - Particle failure fraction vs temperature, or
 - Failure surface from PARFUME

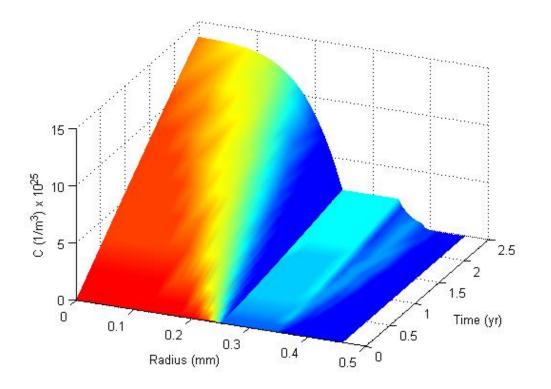


MELCOR FP Model - Release

- Kernel release to buffer region
 - -Diffusion
 - -Recoil
- Release from intact particles -controlled by SiC layer
- Release from failed particles
 - -FP in buffer layer + further release from kernel
 - -Can be further holdup in matrix
- U contamination of matrix in manufacture



Cesium Diffusion in Intact TRISO





- Dust generation
 - -No verified analytic models available currently
 - -Only experimental evidence from GT-MHR, AVR
 - -Parametric model in MELCOR
- Dust transport and distribution
 - -Need initial distribution in primary system for accident
 - MELCOR transport calculation for operation
 - Provided by user
 - -Resuspension
 - Force balance/parametric liftoff model
 - -FP adsorption on dust



Total Release from Failed Particles

Particles fail at different times during accident

 Convolution integral of failure rate and release
 fraction

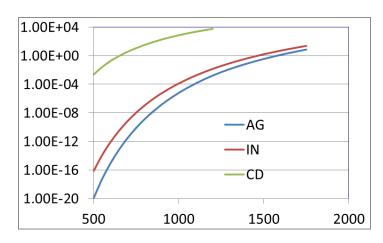
$$F_{tot}(t) = \int_0^t \frac{dF_W(\tau)}{d\tau} F_R(t-\tau)d\tau$$

where F_{tot} = Total release fraction F_W = Failure fraction F_R = Release fraction of particle



Ag/In/CD Release From Control Rods

- FPT-1 Post-test analysis demonstrated the importance of predicting structural aerosol release of CR materials
- MELCOR has a single alloy material to represent all three component materials
 - Vapor pressure of alloy taken as the component with smallest vapor pressure
 - Composition of remaining unvaporized alloy does not change
 - In reality, cadmium would vaporize quickest, followed by indium, then silver, and the composition of the molten alloy would change.
- MELCOR assumes vaporization controlled by mass diffusion in the gas
 - Cooling by vaporization ignored
 - Condensation of vapor not allowed



$$\dot{m}_{vap} = h_m A_c M_{Ag} \left(\frac{P_v(T_c)}{RT_c} - \frac{P_{Ag}(T_b)}{RT_b} \right)$$

Mass transfer analogy



Ag/In/CD Release From Control Rods

• Model is not enabled by default COR_CR record

COR_CR ACTC

• User must also specify three new RN classes AG-CR, IN-CR, CD-CR

-Vapor pressure (C7110), diffusion coefficients (C7111), and molecular weights (C7120) must also be defined (1) IAICON

Turns on the silver release model

(a) 0 or NACT Model is not active. No additional fields are required.

(b) 1 or ACTC Model is active, vaporization is allowed f

Model is active, vaporization is allowed from candling material only

(c) 2 or ACTDC Model is active, vaporization is allowed from both candling material and conglomerate

(type = integer / character*5, default = 0, units = none)

The following fields are required if and only if IAICON = ACTC or 1, ACTDC or 2. If they are not set, then the default values will be used:

(2) ARATIO

The area ratio of break area to control rod internal cross-sectional area. This cannot be greater than 1.

(type = real, default = 0.1, units = none)

(3) AKFRCT

The flow loss coefficient used in the release velocity calculation. If input, this will override the value calculated according to the formula given in the reference manual.

(type = real, default = 0.32*(1.0 - ARATIO), units = none)



RN1_CRCL Record

RN1_CRCL ! N COR Mat RN Mat Fraction 1 POISON BO2 1.0

Rarely (if ever) should be changed Mapping of nonradioactive core mass to the RN classes

ICRMT	ICLSS	Fraction
1	UO2	1.0
2	Ce	1.0
3	Ce	1.0
4	Mo	1.0
5	Mo	1.0
6	BO2	1.0 (BWR)
6	Cd	0.05 (PWR)
6	Ag	0.95 (PWR)

