

# MELCOR 1.8.6 Volatile iodine release challenges in case of external reactor vessel cooling combined with filtered vent - Effect of pH

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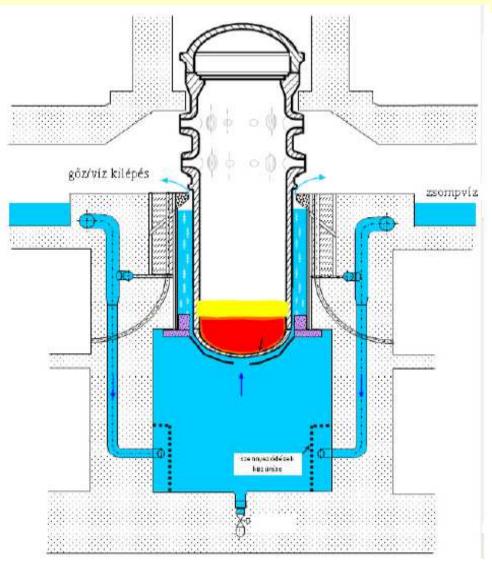


# Accident conditions for dominating PSA-2 case: PDS-05C

Initiating event	SBLOCA d=11mm
ECCS	No
Cont. Init. State	Intact
Spray	No
Sec. Side depressurisation	Yes
Sec. Side FW	No
Prim. Side depressurisation	Yes
Early cont failure	No
Ex-vessel cooling	Yes
Filtered vent	Yes
Late phase cont. Failure	No

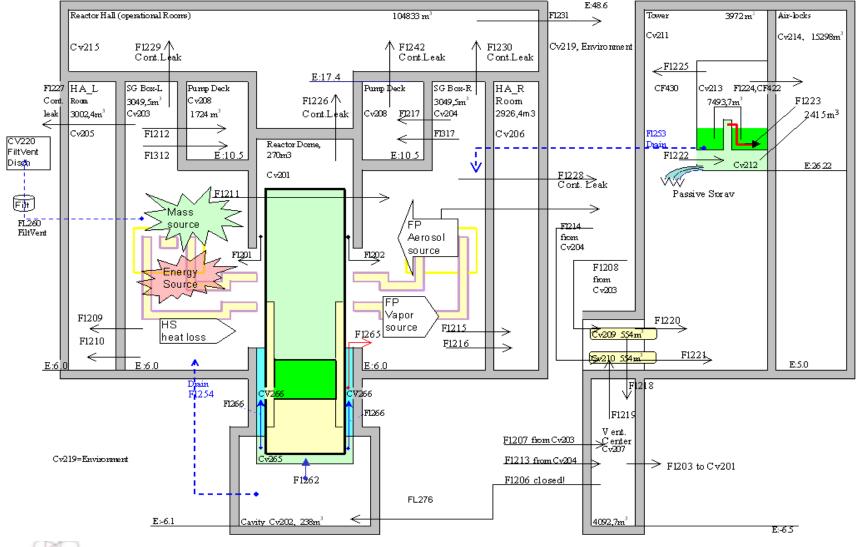


# Plant solution: External Vessel Cooling





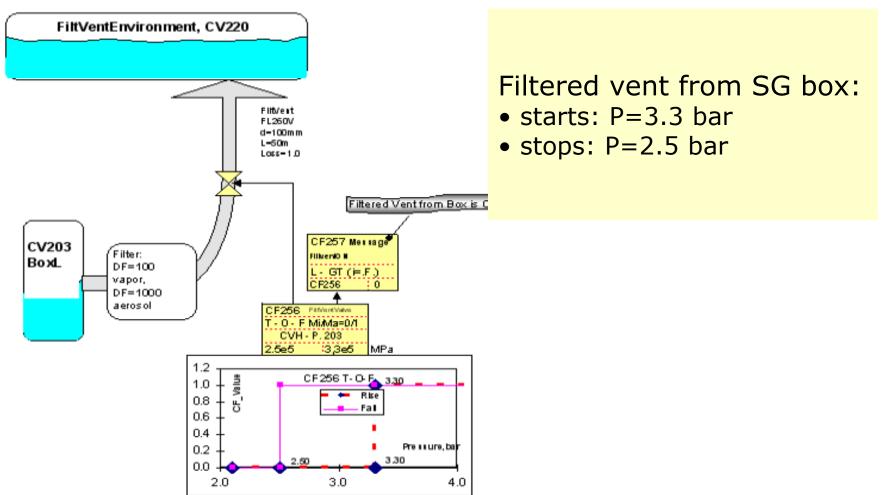
## VVER-440 Simplified Stand Alone Containment: Sources: Primary circ + Ex-vessel cooling + BC drain+Fvent



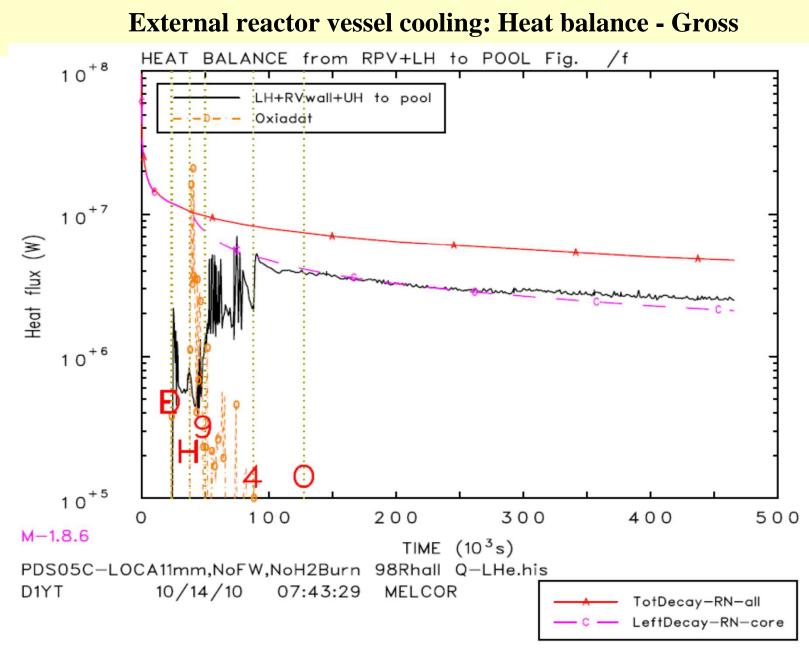


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## Filtered vent model for VVER-440/213 Stand Alone Containment







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## **External Reactor Vessel Cooling: Heat balance - Conclusions**

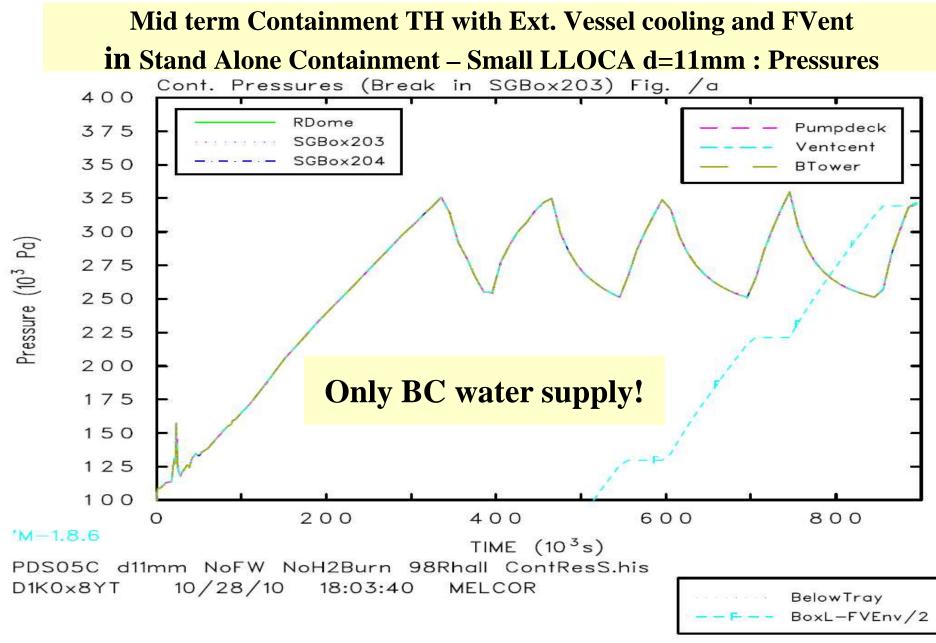
Heat removed from RPV:

- early stages: Vessel wall dominates
- late stages: Lower Head dominates
- Very late stages: Vessel wall and LH are similar

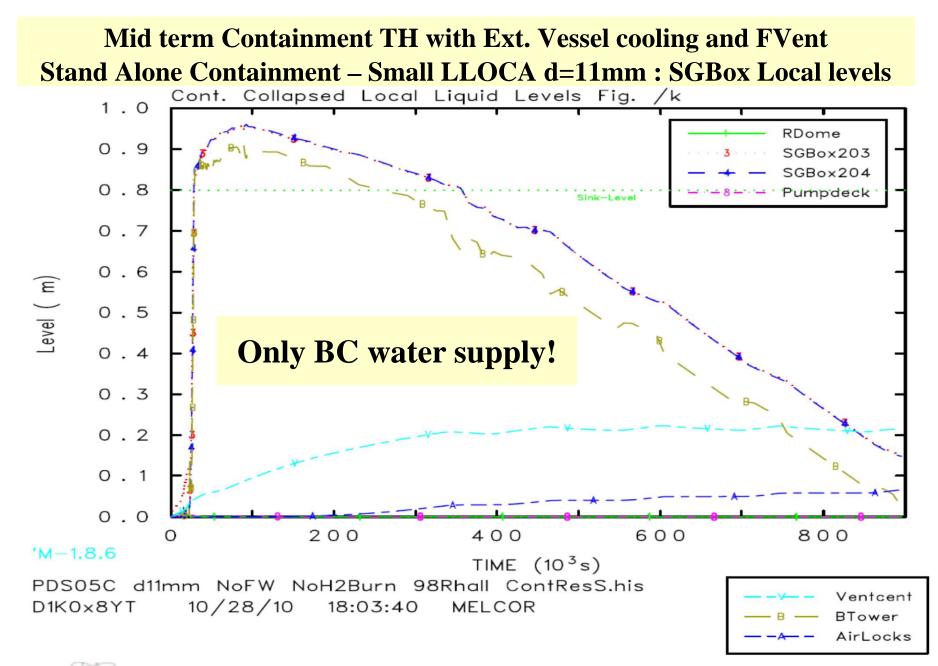
Unresolved:

- Presence of FOCUSING EFFECT of molten metallic layer on RPV wall
- •Effect of crust separating the molten metallic layer from the RPV wall





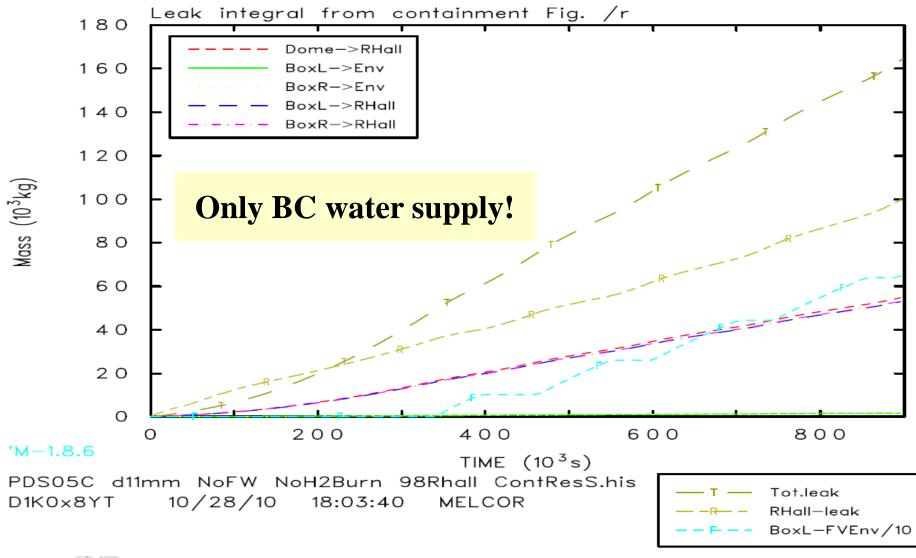




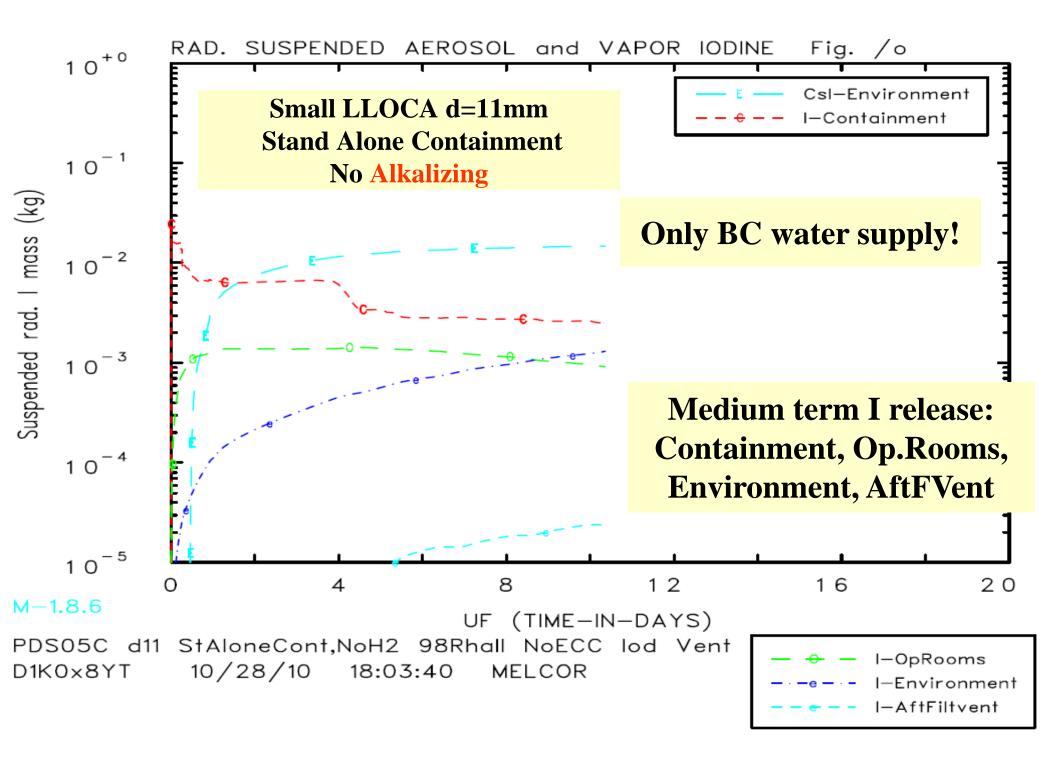
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### Mid term Containment TH with Ext. Vessel cooling and FVent in Stand Alone Containment – Small LLOCA d=11mm : Leak rates



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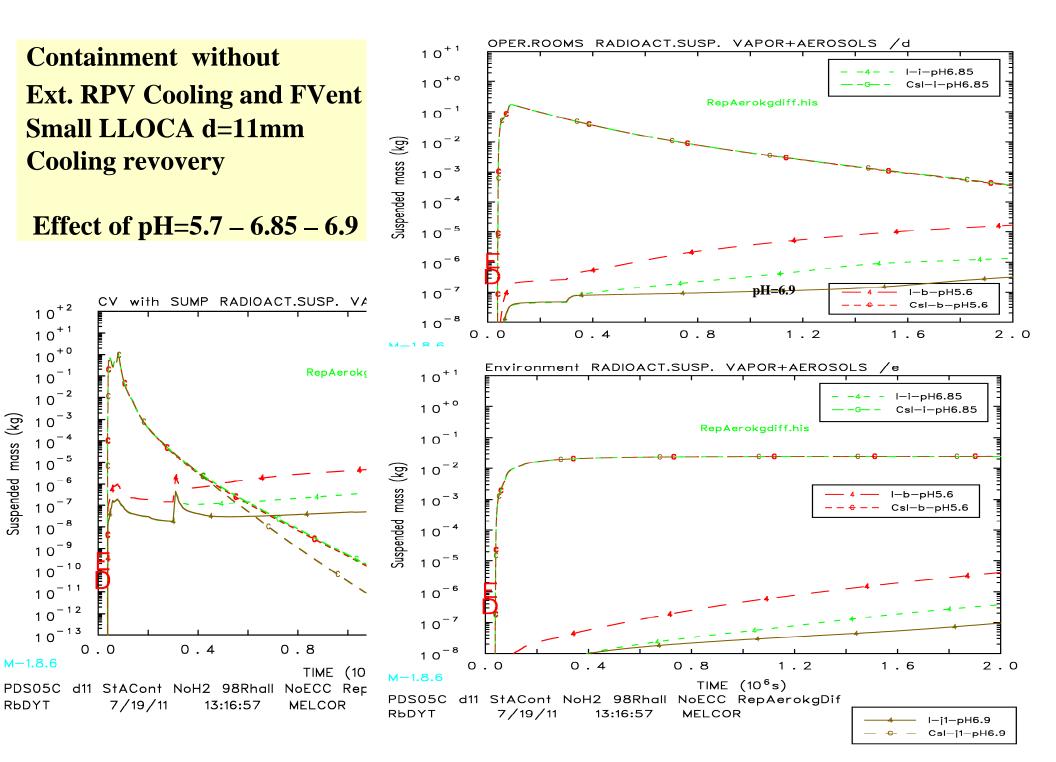


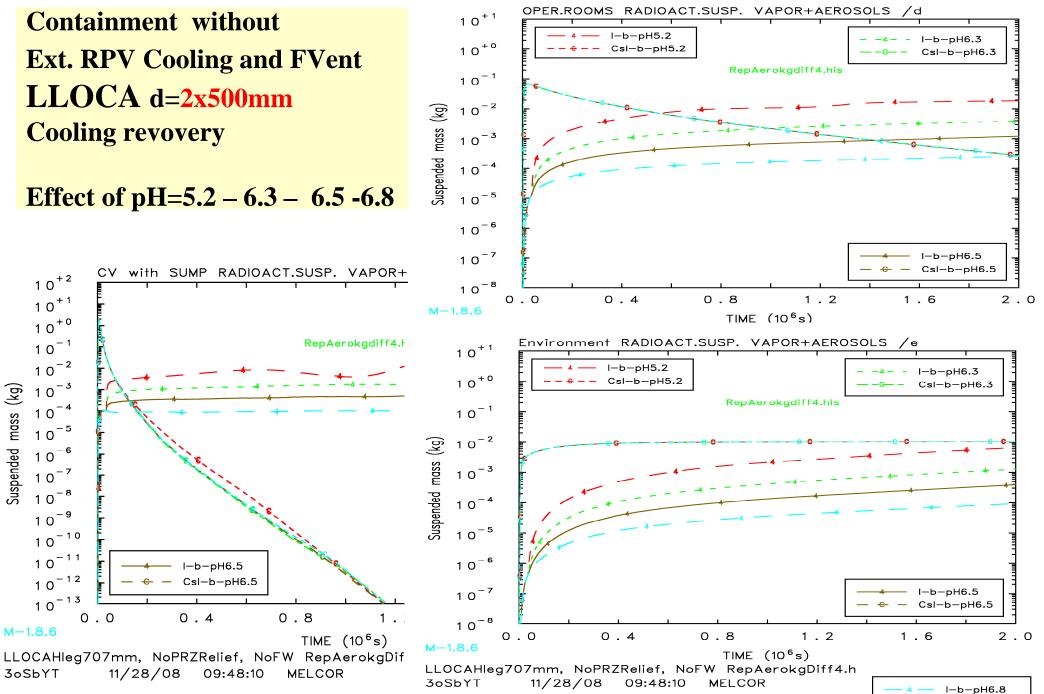
OPER.ROOMS RADIOACT.SUSP. VAPOR+AEROSOLS /d 1 0<sup>-1</sup> **Containment** with I-pH5.7 Csl-pH5.7 I-pH6.1 **Ext. RPV Cooling and FVent** RepAerokgdiff.his Csl-pH6.1 Small LLOCA d=11mm 1 0<sup>-2</sup> Suspended mass (kg) Effect of pH=5.7 – 6.1 – 7.5 1 0<sup>-3</sup> with SUMP RADIOACT.SUSP. VAPOR+AEROSOLS CV  $10^{-1}$ 4 RepAerokgdiff.his 10 0.4 0.0 0.2 0.6 0.8 1 0<sup>-2</sup> Environment RADIOACT.SUSP. VAPOR+AEROSOLS /e Suspended mass (kg)  $10^{-1}$ I-pH5.7 Csl-pH5.7 1 0<sup>-2</sup>  $10^{-3}$ Suspended mass (kg) RepAerokgdiff.his 1 0<sup>-3</sup> 1.0 0.0 0.2 0.4 0.6 0.1 M-1.8.6 TIME (10<sup>6</sup>s) I-pH6.1 PDS05Cd11 ContAlone NoH2 NoECC RepAerokgDiff4.his Csl-pH6.1 D1K0x8YT 7/09/11 18:45:53 MELCOR 10 0.2 0.0 0.4 0.6 0.8 TIME  $(10^6 s)$ I-pH7.5 PDS05Cd11 ContAlone NoH2 NoECC RepAerokgDiff4.his Nuclear Safe NUB Csl-pH7.5 D1K0×8YT 7/09/11 18:45:53 MELCOR I-pH8.0

1.0

1.0

Csl-pH8.0





— 4 — Т-в-рн6.8 — <del>6</del> — - Csl-b-рН6.8

# Effect of pH on Iodine in Environment

(kg)

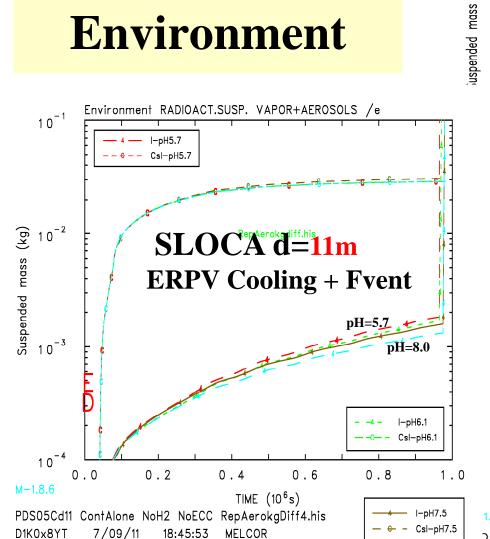
I-pH8.0

Csl-pH8.0

SbYT

11/28/08

09:48:10



Environment RADIOACT.SUSP. VAPOR+AEROSOLS /e 1 0<sup>+ 1</sup> I-i-pH6.85 10<sup>+0</sup> Csl-i-pH6.85 RepAerokgdiff.his  $10^{-1}$ 1 0<sup>-2</sup> I-b-pH5.6 10<sup>-3</sup> - Csl-b-pH5.6  $10^{-4}$ SLOCA d=11m 10<sup>-5</sup> **Cooling recovery** pH=5.6 1 0<sup>-6</sup> 10<sup>-7</sup> nH=6 1 0<sup>-8</sup> 0.4 0.8 1.2 0.0 1.6 2.0 Environment RADIOACT.SUSP. VAPOR+AEROSOLS /e 1 0<sup>+ 1</sup> I-b-pH5.2 - I-b-pH6.3 1 0<sup>+0</sup> Csl-b-pH5.2 Csl-b-pH6.3 RepAerokgdiff4.his  $10^{-1}$ 1 0<sup>-2</sup> pH=5.2 10<sup>-3</sup> 1 0<sup>−₄</sup> pH=6.8 1 0<sup>-5</sup> LLOCA d=2x500mm 10<sup>-6</sup> **Cooling recovery** 1 0<sup>-7</sup> I-b-pH6.5 Csl-b-pH6.5 10<sup>-8</sup> 0.0 0.4 0.8 1.2 1.6 2.0 1.8.6 TIME  $(10^6 s)$ CAHleg707mm, NoPRZRelief, NoFW RepAerokgDiff4.h

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I-b-pH6.8

Csl-b-pH6.8

## Volatile Iodine Release to Environment in Stand Alone Containment – Small LLOCA d=11mm and LLOCA 2x500mm: Summary

Environment after 23 days	(Aerosol release	ceases after 3 days)
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	CsI aerosol,kg		Volatile I,kg	Decrease factor
SLOCA d11mm + Ext.RPV cooling+ Fvent	pH=5.7	<i>3E-3</i>	2E-3	
	<i>pH</i> =8.0	same	0.3E-3	6
SLOCA d11mm + primary circ.cooling recovery	<i>pH</i> =5.6	2.5E-3	4E-6 (high humidity)	
	рН=6.9	same	1E-7 (high humidity)	40
SLOCA d=11mm + Ext.RPV cooling+ Fvent	<i>pH</i> =5.2	1E-2	8E-3	
<i>pH</i> =6.8	same	1E-4	80	



Possible explanation of small effect of pH in case d=11mm LOCA + Ext. RPV cooling + Filt Vent

d=11mm LOCA + Ext. RPV cooling + Filt Vent

Constant evaporation of volatile I from pool at certain pH Evaporated I interferes with Huge mass sink from atmosphere provided by FVent

Sink keeps the atmospheric volatile iodine concentration low No equilibrium between atmoshere and pool forms

As a result

- Sink effect dominates over pH effect
- Long term driving force of volatile I from pool to atmosphere



Possible explanation of Large effect of pH in case LOCAs WITHOUT Ext. RPV cooling and without Filt Vent

d=11mm LOCA and LLOCA d=2x500mm

Constant evaporation of volatile I from pool at certain pH Evaporated I accumulates in the atmosphere

Larger atmospheric volatile iodine concentration results in larger back current of I to the pool Equilibrium between atmosphere and pool forms according to Partition Coefficient

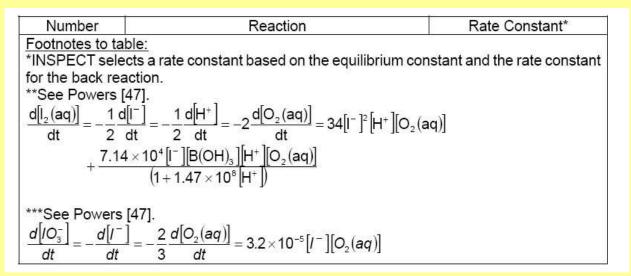
As a result

- *pH effect dominates the release* 



### MELCOR 1.8.6 In-Vessel Retention and IPM model: Problems

- Max 2 control volumes with IPM can be activated
- Mass balance error is between 8-100% (depends on sequence)
- Class assignment is confusing
- It is not clear if organic iodine is calculated or NOT
- (The framework for the organic reactions is in place, but the equations have not been entered, due to a lack of data to compare results. When data becomes available, the organic reactions can be activated by entering the equations into the EQINIT routine)
- I2 release from pool is very poorly described
- Reference to POOL SPECIATION MODEL POWERS is NOT available





### MELCOR 1.8.6 In Vessel Retention and IPM model: Problems

### **IPM**

- NRC questions the validity of the IPM model

#### However

- IPM reproduced the Phebus FPT-1 test well
- The plant calculated results seems to be reasonable
- Tendencies sometimes confusing but according to accepted Kanon
- Larger pH resulted in smaller Iodine release to environment

#### Major message

- External RPVessel Cooling + Filtered Vent NEEDS Volatile Iodine Filter



# Thank You For Your Attention

