

Au dispersion

'Influence of Methyl Halide Treatment on Gold Nanoparticles Supported on Activated Carbon' Angew. Chem. Int. Ed. 50 (**2011**) 8912-8916.

J. Sá, A. Goguet, S. F. Rebecca Taylor, R. Tiruvalam, C. J. Kiely, M. Nachtegaal, G. J. Hutchings, C. Hardacre

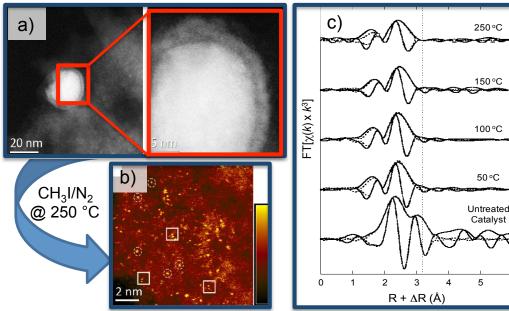


Figure 1: Au/C treated in 0.4%CH₃I/N₂.

- a) Aberration corrected HAADF TEM Au/C after 5 min on stream @ 250 °C;
- b) Aberration corrected HAADF TEM Au/C after 15 min on stream @ 250 °C;
- c) Fourier transformed EXAFS of Au/C after 60 min on stream at different temperatures

Gold particles supported on carbon when subjected to a flow of methyl iodide or bromide redisperse from large ensembles to single atoms and/or dimers of gold (Figure 1b). Methyl halide oxidizes gold leading to gradual particle dissolution (Figure 1a). The process could be carried out at temperatures as low as 50 °C(Figure 1c). The excess of halide could be removed by a post-treatment of the material with $1\%H_2O/H_2$, which does not influence the metal dispersion. This remarkable transformation opens the possibility of re-activating gold catalysts that lost their performance due to metal particles sintering.

The finding was proposed based on a combination of in situ (XAS), ex situ characterization (aberration corrected HAADF-TEM, XRD and XPS) and kinetic measurements. The work was a combined effort from Queen's University Belfast (UK), Paul Scherrer Institute (Switzerland), Lehigh University (USA) and Cardiff University (UK).