





# Electrode engineering of graphite/silicon electrodes

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- develop carbon based anode material with specific charge > 450 mAh/g by addition of small amounts of silicon
- compatibility with common industrial processes
- understand influence of different electrode components  $\bullet$

- 4.75 wt% Si, 90.25 wt% intercalating carbon, 1 wt% SuperC65, 4 wt% binder
- for comparison: 95 wt% intercalating carbon, 1 wt% SuperC65, 4 wt% binder
- 20 mA/g first cycle, following 50 mA/g between 5 mV and 1.5 V vs. Li<sup>+</sup>/Li
- 1M LiPF<sub>6</sub> in EC/DMC 1:1 (w:w) unless stated otherwise

## **Role of carbon material**



electrodes with PVDF binder

- carbon/Si electrodes: only carbon contributes towards specific charge after ~10 cycles
- for SLP30/Si electrodes, cycling of graphite part is negatively affected by presence of Si

#### **Binder influence**

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carbon black



cross-sectional view

- larger particles & higher aspect ratio for SLP30 than KS6  $\rightarrow$ preferred orientation of SLP30 parallel to substrate
- electrodes with PVDF shown, electrodes with CMC/PAA exhibit the same morphology

## **Electrode density**





using CMC/PAA: improved cycling stability for both KS6 and SLP30 based electrodes containing Si

## FEC electrolyte additive



#### electrodes with CMC/PAA binder

Cycle number

enhanced cycling stability in presence of fluoroethylene



#### faster charge fading with densified electrodes containing Si

# Graphite/Si ratio

carbonate (FEC) additive with 90 % capacity retention from 2<sup>nd</sup> to 100<sup>th</sup> cycle



- electrodes with CMC/PAA binder
- first cycles: higher specific charge with 10 % Si in active material
- stronger fading with 10 % Si in active material

# Conclusions

- KS6: best of the different types of intercalating carbon for combination with Si - better cycling stability than SLP30 based electrodes
  - compared to PC6, less Si is needed to obtain same specific charge
- binder and electrolyte have strong influence on cycling stability

