

ELECTROCHEMISTRY LABORATORY

Correlation of electrode expansion and cyclability in graphite-based negative electrodes

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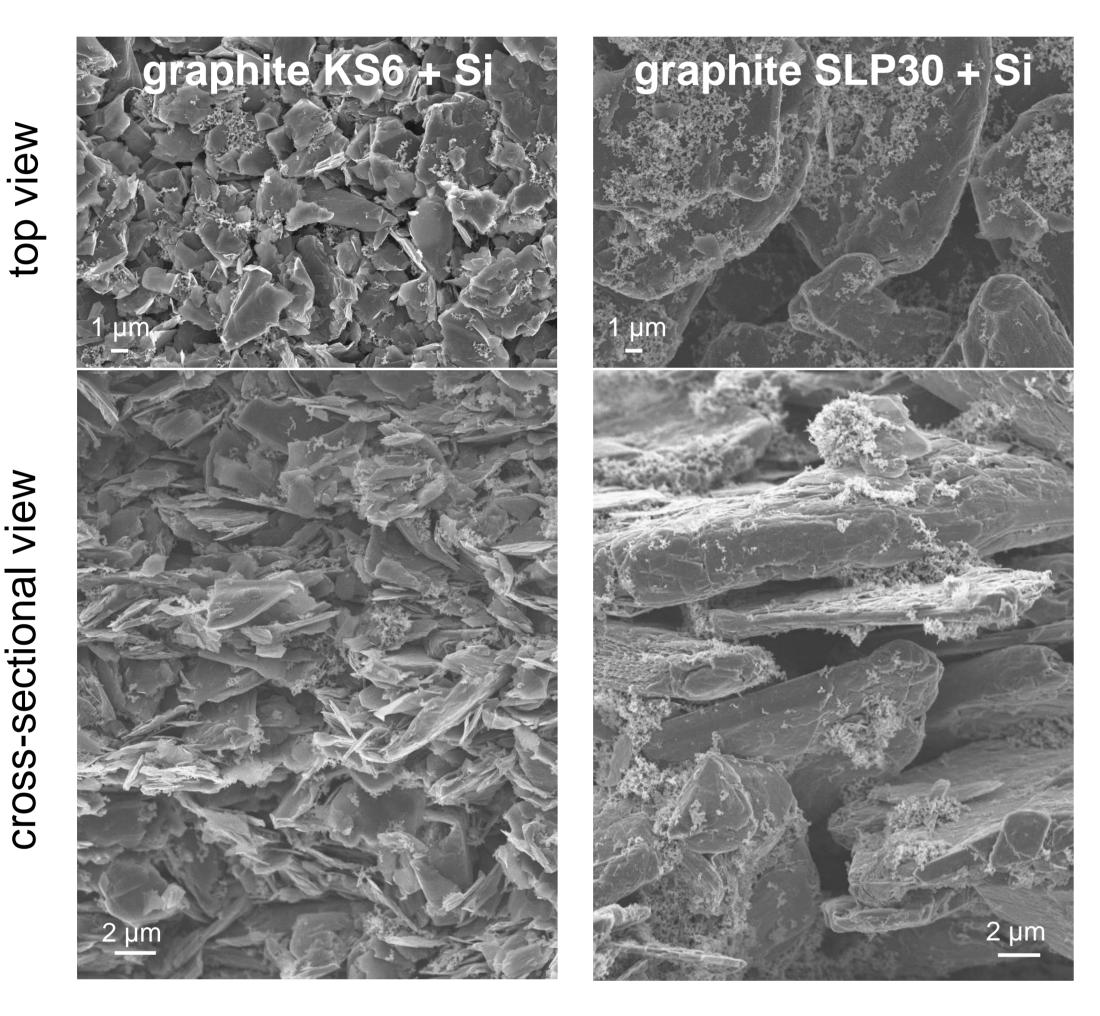
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Motivation / Targets

Experimental

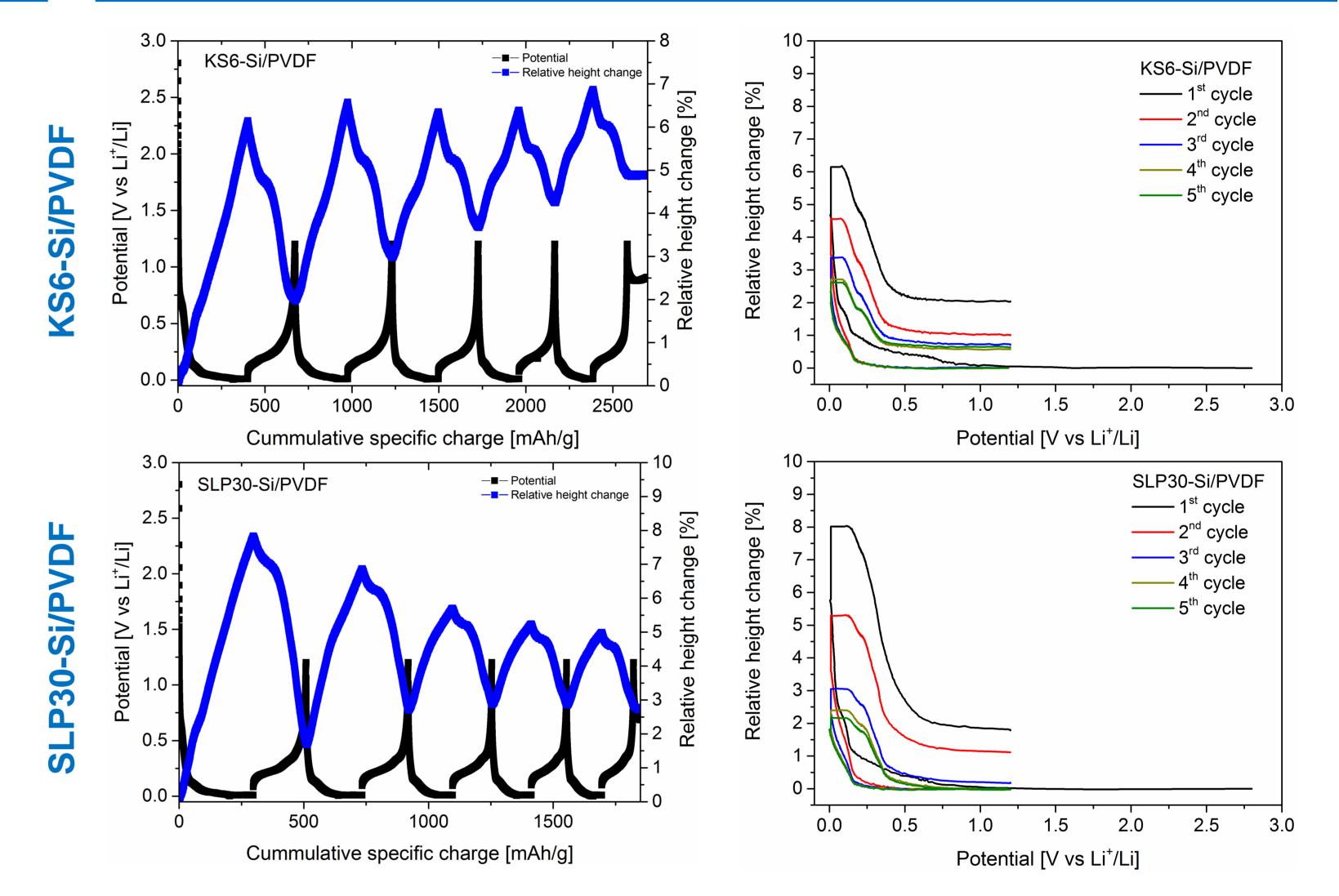
- Develop carbon based anode material with specific charge > 450 mAh/g by addition of small amounts of silicon
- Investigate the influence of graphite / binder combination on cycling
- Understand the relationship between expansion and electrode failure

Graphite / Binder influence

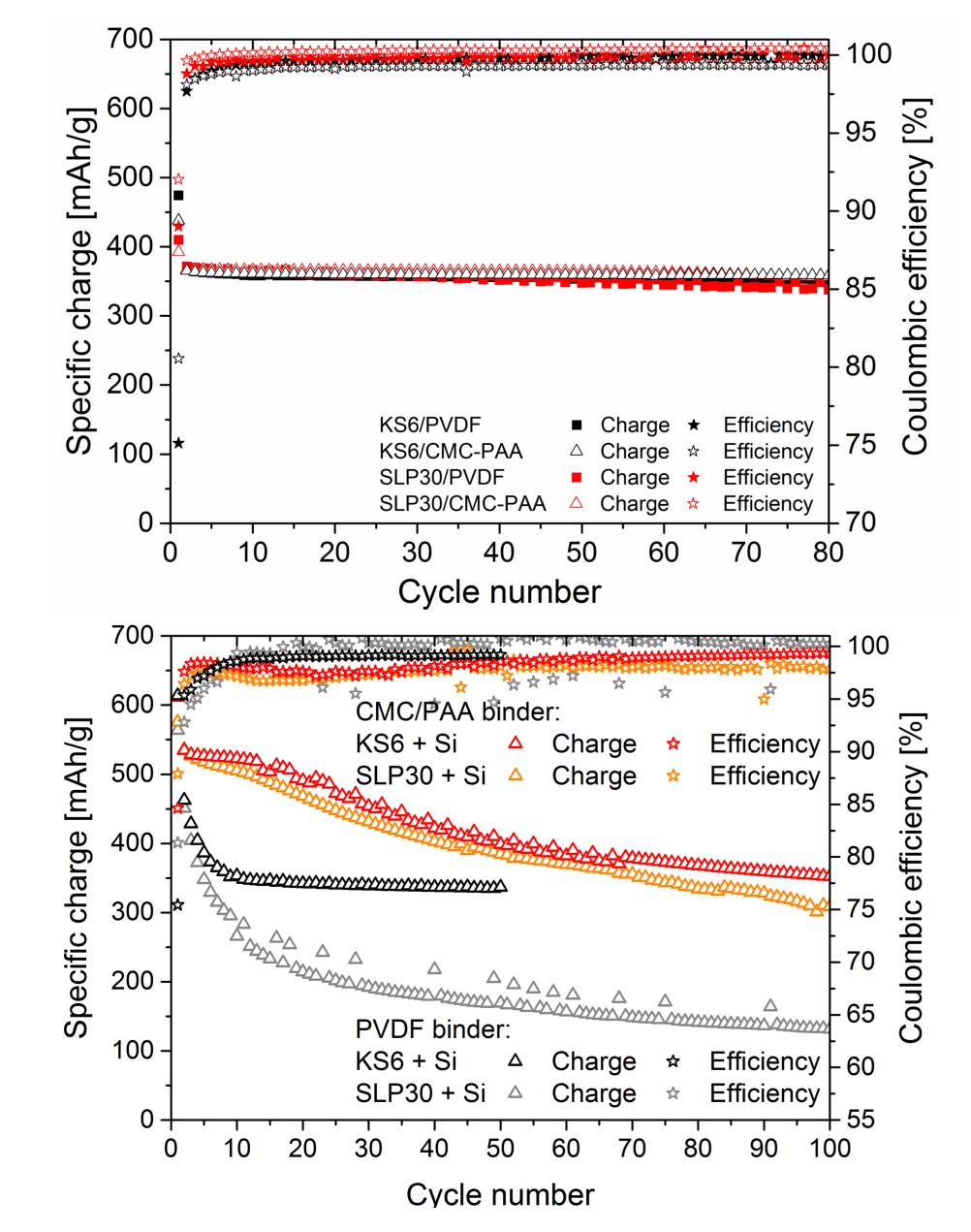


•4.75 wt% Si, 90.25 wt% intercalating carbon, 1 wt% SuperC65, 4 wt% binder
•Baseline: 95 wt% intercalating carbon, 1 wt% SuperC65, 4 wt% binder
•Electrochemistry: 20 mA/g 1st cycle, then 50 mA/g, 5 mV – 1.5 V vs Li⁺/Li
•Electrolyte: 1M LiPF₆ in EC/DMC 1:1 (w:w)

Electrochemical in situ dilatometry



➤ Larger particles & higher aspect ratio for SLP30 than KS6 → preferred orientation of SLP30 parallel to substrate

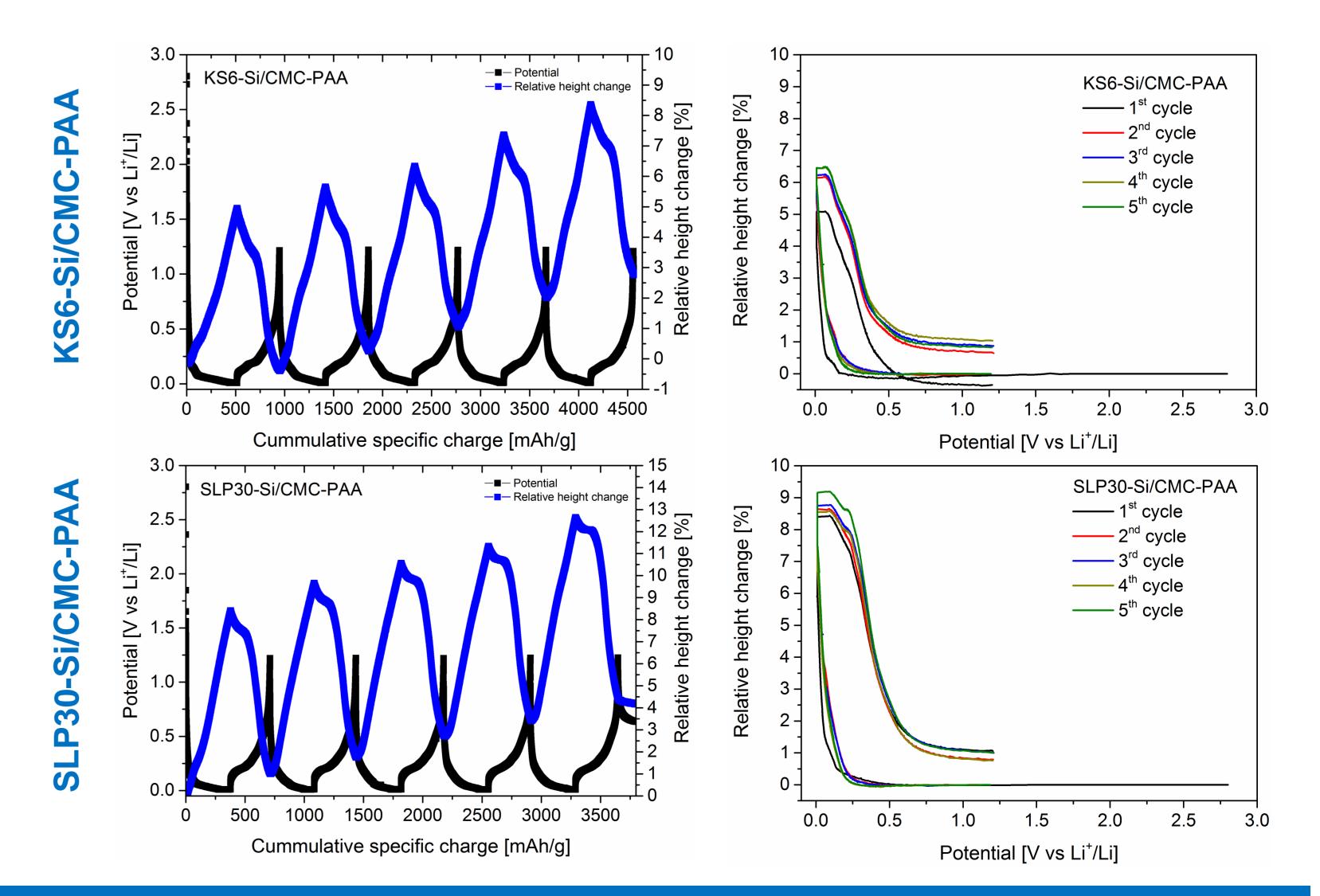


PVDF is a stiff binder

- KS6 and SLP30 electrodes expand irreversibly ca. 2% in 1st cycle
- Capacity fading linked to ''loss'' of active Si \rightarrow lower expansion in subseq. cycles

CMC-PAA is an elastic binder

- Irreversible expansion of ca. $1\% \rightarrow$ recovered every cycle
- Holds expansion of up to 9% with low irreversibility
- Capacity fade is **not** due to 'loss' of active Si \rightarrow different failure mechanism



- Graphite only: stable cycling regardless of the binder
- Graphite with silicon: improved cycling stability for both KS6 and SLP30 based electrodes with CMC-PAA crosslinked binder

Conclusions

- Both KS6 and SLP30 electrodes show stable cycling regardless of the binder
- Different failure mechanism with PVDF
 - KS6-Si keeps the graphite charge, SLP30-Si fails even below graphite's capacity
- Binder has strong influence on accommodation of volume changes
- Failure mechanism is influenced by the binder type:
 - CMC-PAA holds larger expansion avoiding loss of active Si
 - in situ dilatometry provides new insights into binder performance

